ELEVATED TEMPERATURE PROPERTIES AS INFLUENCED BY NITROGEN ADDITIONS TO TYPES 304 AND 316 AUSTENITIC STAINLESS STEELS





AMERICAN SOCIETY FOR TESTING AND MATERIALS

ELEVATED TEMPERATURE PROPERTIES AS INFLUENCED BY NITROGEN ADDITIONS TO TYPES 304 AND 316 AUSTENITIC STAINLESS STEELS

A symposium presented at the Seventy-second Annual Meeting AMERICAN SOCIETY FOR TESTING AND MATERIALS Atlantic City, N. J., 22–27 June 1969

ASTM SPECIAL TECHNICAL PUBLICATION 522 J. J. Heger and G. V. Smith, co-chairmen

List price \$10.50 04-522000-40



AMERICAN SOCIETY FOR TESTING AND MATERIALS 1916 Race Street, Philadelphia, Pa, 19103

© BY AMERICAN SOCIETY FOR TESTING AND MATERIALS 1973 Library of Congress Catalog Card Number: 72-88610

NOTE

The Society is not responsible, as a body, for the statements and opinions advanced in this publication.

> Printed in Baltimore, Md. February 1973

Foreword

The Symposium on Elevated Temperature Properties as Influenced by Nitrogen Additions to Types 304 and 316 Austenitic Stainless Steels was presented at an informal workshop session held at the 72nd Annual Meeting of the Society, in Atlantic City, N. J., 22–27 June 1969. The sponsors of this symposium included the Joint Committee on Effect of Temperature on the Properties of Metals, Metals Properties Council, American Society for Testing and Materials, and American Society of Mechanical Engineers. J. J. Heger, U. S. Steel Corporation, and G. V. Smith, consultant, served as co-chairmen.

Related ASTM Publications

Report on Elevated-Temperature Properties of Selected Superalloys, DS 7-S1 (1970), \$11.00

Evaluation of the Elevated Temperature Tensile and Creep-Rupture Properties of C-Mo, Mn-Mo, and Mn-Mo-Ni Steels, DS 47 (1971), \$6.50

Elevated Temperature Static Properties of Wrought Carbon Steel, STP 503 (1972), \$3.00

Contents

Introduction	1
Mechanical Properties of Hot-Extruded 304N and 316N Stainless Steel Pipe—PHILIP KADLECEK	3
High-Nitrogen Austenitic Stainless Steels—C. E. SPAEDER, W. F. DOMIS, AND K. G. BRICKNER	35
Elevated Temperature Properties of Nitrogen-Containing Type 304L Auste- nitic Stainless Steel—P. D. GOODELL AND J. W. FREEMAN	46
Influence of Nitrogen on the Creep-Rupture Properties of Type 316 Steel- T. M. CULLEN AND M. W. DAVIS	60
A Nitrogen Grade of Types 304 and 316 Austenitic Stainless Steels; Specifica- tion and Code Considerations—I. A. ROHRIG	79
Creep and Creep-Rupture Properties of Types 304N and 316N Stainless Steels-w. F. DOMIS	86
Service Experience with H Grades of Austenitic Steel—G. J. SCHNABEL	100
Effect of Elevated Temperatures on the Properties of Nitrogen-Bearing Type 216 Steel—J. A. CHIVINSKY	105

Introduction

A plan for an informal workshop discussion session was organized during 1968 by The Joint Committee on Effect of Temperature on the Properties of Metals for the purpose of reviewing and clarifying differences in creeprupture properties between the "regular" and the "H" grades of Types 304, 316, 321, and 347 austenitic stainless steels. The plan included consideration of the influence of carbon and nitrogen contents on the creep-rupture strengths plus preparation of a summary of short time elevated temperature properties.

As the plan developed, it became apparent that the paramount interest focussed on the nitrogen-bearing grades. The outcome was a jointly sponsored session held at the ASTM Annual Meeting at Atlantic City, N.J., June 1969, which presented a series of papers concerned with several aspects of the properties and uses of nitrogen-strengthened austenitic steels. Cosponsorship was contributed by The Metal Properties Council, The American Society for Testing and Materials and The American Society for Mechanical Engineers.

The session at the ASTM meeting was advertized as being restricted to informal verbal reporting and discussion of current data. At the completion of the session, however, it was apparent to all concerned that the presentations contained a sufficient wealth of excellent high temperature information to warrant publication. The Metal Properties Council, as a further means of fulfilling its function of service to the metals industry, undertook the task of inducing the speakers to prepare and submit for review written versions of their papers. This has been accomplished and the material is presented herewith.

The importance of the data contained in this Special Technical Publication lies in the needs of the design engineer which extend beyond the aids supplied by industry standards and codes. The basic function of the designer is to exercise an informed judgment in the selection of appropriate materials for safe design, which is achieved only through a thorough understanding of the behavior of metals under stress at elevated temperatures. The papers of this session offer a means of advancing this necessary understanding to an important degree now that they have been made available by publication through the efforts of The Metal Properties Council and the American Society for Testing and Materials.

Special acknowledgments and thanks are due to the authors of the papers; also to Mr. J. J. Heger, U. S. Steel Corporation, Monroeville, Pa., to Dr. G. V. Smith, Consultant, Ithaca, N.Y., to Dr. M. Semchyshen, Climax Molybdenum Co. of Mich., Ann Arbor, Mich., and to J. A. Fellows, Shaker Heights, Ohio, for their effective joint activities in initiating and preparing the workshop program. Appreciation is also due Dr. Smith for his excellent service as session moderator.

E. J. Rozic, Jr.

The Babcock and Wilcox Co. Beaver Falls, Pa.

Mechanical Property Data on Hot-Extruded 304N and 316N Stainless Steel Pipe

REFERENCE: Kadlecek, Philip, "Mechanical Property Data on Hot-Extruded 304N and 316N Stainless Steel Pipe," Elevated Temperature Properties as Influenced by Nitrogen Additions to Types 304 and 316 Austenitic Stainless Steels, ASTM STP 522, American Society for Testing and Materials, 1973, pp. 3-34.

ABSTRACT: The effects of nitrogen in both Types 304 and 316 stainless steel were investigated on a production scale and the results are presented in this paper. The test program revealed that nitrogen had an affirmative strengthening effect on wrought austenitic stainless steels. A program establishing hot tensile, stress-rupture, creep, and fatigue data, plus welding experiments, are reported.

KEY WORDS: nitrogen, austenitic stainless steels, welding, tensile strength, piping, creep rupture strength, mechanical properties, tubing

A customer's request for nitrogen-bearing Type 304 stainless steel stimulated interest at Cameron Iron Works as to the overall effects of nitrogen in both Type 304 and 316 stainless steels. At that time, little production data on tubing were available, even though a search through literature published during the past 30 years revealed numerous references to the effects of nitrogen. The data which were available from laboratory scale investigations did show that nitrogen had a pronounced strengthening effect on wrought austenitic materials.

Test Program

In 1968, a program was initiated at Cameron Iron Works to evaluate the effect of a controlled nitrogen addition on a production basis. The heats analyzed in this study were either electric arc or vacuum induction melted or arc remelted (see Tables 1A and 1B for compositions and tensile data). The minimum heat size was 25 tons. All test material was in the form of hot-extruded seamless pipe in the following sizes representing the range in-

¹ Chief development engineer, Cameron Iron Works, Inc., Houston, Tex. 77001.



FIG. 1-Effect of carbon and nitrogen on Type 316 stainless steel yield strength.



FIG. 2-Type 304N stainless steel.



FIG. 3-Type 316N stainless steel.



FIG. 4-Type 304N stainless steel hot tensile data.





FIG. 6—Type 304N stainless steel stressrupture test data.



FIG. 7-Type 316N stainless steel stress-rupture tests data.



FIG. 8-Minimum-creep rate data, Type 304N stainless steel.



FIG. 9-Minimum-creep rate data, Type 316N stainless steel.



FIG. 10-Fatigue data, Type 304N stainless steel.



FIG. 11-Fatigue data, Type 316N stainless steel.

cluded in the test program: 8.75 in. (222 mm) inside diameter by 1.0 in. (24.4 mm) wall, 21.89 in. (556 mm) inside diameter by 3.30 in. (83.8 mm) wall, and 29.0 in. (736 mm) inside diameter by 2.50 in. (63.5 mm) wall.

The standard tensile requirements on the production parts provided sufficient room temperature test data. For supplementary information, a program was established to provide hot tensile, stress-rupture, creep, and fatigue data. Also, welding tests were performed using standard and nitrogen-bearing electrodes.

All test material received a standard production heat treatment of one hour per in. (per 25.4 mm) of wall thickness at 1925 F (1052 C), water quench. Since the test material was in the form of hot extruded pipe, the average "as-extruded" grain size varied from an ASTM 2 to 5 and was not altered by this heat treatment.

Test Results

Examination of room temperature tension test results indicated a reasonable correlation between the nitrogen content and the strength of the material. Although the ultimate tensile strength increased with higher nitrogen contents, the most significant improvement was in the yield strength. Figure 1 shows the effect of nitrogen on the yield strength of one size of Type 316 stainless steel pipe. (Since carbon is also an effective strengthener,

				Η	at Treatr	nent: 1 h/	in. at 1925	5 F, water	r quench			
									Tens	ile Properties		Tests
Heat No.	C	Mn	Ъ	ß	Si	Ċ	Ni	Z	Ultimate Yield Strength, ksi ^a	Yield Strength, ksi ^a	Elongation,	Reduction of Area, %
C9739	0.08	1.56	0.012	0.010	0.24	18.61	10.51	0.13	84.9	40.0	52.5	59.8
C9739	0.08	1.56	0.012	0.010	0.24	18.61	10.51	0.13	82.8	39.0	64.7	53.0
52390	0.08	1.68	0.015	0.010	0.18	18.73	10.67	0.11	84.3	40.4	53.5	72.7
									84.2	42.3	55.0	68.1
F0186	0.07	1.47	0.015	0.006	0.28	18.37	10.50	0.13	87.3	45.0	54.0	75.5
									87.8	47.0	53.0	73.5
F0186	0.07	1.47	0.015	0.006	0.28	18.37	10.50	0.13	86.3	41.2	52.0	68.4
									89.5	44.9	49.0	66.1
F0186	0.07	1.47	0.015	0.006	0.28	18.37	10.50	0.13	87.0	43.3	52.0	72.7
									87.1	44.0	52.0	70.1
F0186	0.07	1.47	0.015	0.006	0.28	18.37	10.50	0.13	83.3	38.7	51.0	71.4
									86.6	43.2	52.0	70.1
F0262	0.06	1.46	0.006	0.008	0.25	18.77	10.40	0.11	82.2	40.0	56.3	78.6
									86.1	43.5	55.5	54.7
F0262	0.06	1.46	0.006	0.008	0.25	18.77	10.40	0.11	87.3	42.6	54.0	70.07
									85.0	39.5	57.0	71.8
F0262	0.06	1.46	0.006	0.008	0.25	18.77	10.40	0.11	81.9	38.5	54.4	70.4
									84.4	43.6	55.0	69.7
F0262	0.06	1.46	0.006	0.008	0.25	18.77	10.40	0.11	83.8	39.0	53.5	68.8
									88.4	44.5	51.6	70.4
F0313	0.07	1.50	0.005	0.005	0.30	18.80	10.66	0.10	85.0	45.3	58.6	74.2
F0313	0.07	1.50	0.005	0.005	0.30	18.80	10.66	0.10	81.4	42.1	60.5	75.8
F0313	0.07	1.50	0.005	0.005	0.30	18.80	10.66	0.10	82.9	44.9	59.5	76.6
									79.5	39.7	61.0	74.2

10

TABLE 1A—Chemistry and room temperature mechanical properties of Type 304N stainless steel, 0.10 to 0.16 percent nitrogen.

F0316	0.06	1.50	0.015	0.005	0.35	18.73	10.57	0.10	80.2	43.1	61.5	75.8
									84.6	45.1	58.0	74.5
0318	0.06	1.53	0.005	0.012	0.35	18.62	10.33	0.11	83.4	45.1	58.0	77.3
									87.0	44.9	55.7	75.1
0314	0.07	1.53	0.009	0.008	0.29	18.54	10.69	0.12	81.4	42.0	60.0	76.2
									83.4	45.7	56.5	75.4
0313	0.07	1.50	0.005	0.005	0.30	18.80	10.66	0.10	82.4	43.1	50.0	56.1
									85.9	43.9	54.0	75.1
$^{0}0315$	0.08	1.58	0.017	0.005	0.31	18.65	10.89	0.10	89.7	44.9	53.7	73.3
0315	0.08	1.58	0.017	0.005	0.31	18.65	10.89	0.10	89.4	45.2	53.5	74.7
0316	0.06	1.50	0.015	0.005	0.35	18.73	10.57	0.10	88.9	42.4	55.0	72.5
									88.7	44.9	53.2	71.0
0316	0.06	1.50	0.015	0.005	0.35	18.73	10.57	0.10	83.7	40.9	55.8	75.5
									88.6	45.9	56.6	73.3
0316	0.06	1.50	0.015	0.005	0.35	18.73	10.57	0.10	83.0	46.1	58.0	76.6
									87.9	49.8	62.5	77.3
0318	0.06	1.53	0.005	0.012	0.35	18.62	10.33	0.11	84.0	41.4	57.0	72.7
									85.9	41.7	56.1	73.0
0318	0.06	1.53	0.005	0.012	0.35	18.62	10.33	0.11	86.2	46.1	55.5	78.4
									84.6	46.9	55.0	71.3
0318	0.06	1.53	0.005	0.012	0.35	18.62	10.33	0.11	86.6	42.7	55.0	75.5
	!								85.9	42.7	56.2	74.7
0317	0.07	1.45	0.015	0.008	0.35	18.54	10.49	0.10	85.0	47.1	57.5	75.4
									84.9	45.4	59.1	76.2
0319	0.06	1.52	0.005	0.006	0.33	18.89	10.69	0.10	85.4	39.9	54.1	74.1
									85.0	41.4	53.5	73.5
0319	0.06	1.52	0.005	0.006	0.33	18.89	10.69	0.10	86.2	48.8	50.9	59.7
									82.2	46.9	51.5	70.4
0320	0.06	1.49	0.014	0.007	0.24	18.49	10.71	0.10	87.1	42.1	56.4	72.6
									86.2	45.5	55.0	73.4
0314	0.07	1.53	0.009	0.008	0.29	18.54	10.69	0.12	81.0	43.9	55.0	73.4
									83.4	42.5	58.0	70.4
0317	0.07	1.45	0.015	0.008	0.35	18.54	10.49	0.10	83.2	42.2	55.0	73.4
									85.6	47.3	55.0	74.2

TABL	E 1AC/	hemistry o	ınd room te	emperature	mechanic	al properti	ies of Type	e 304N sta	tinless steel, 0.	10 to 0.16 per	rcent nitrogen	Continued
				H	eat Treat	ment: 1 h,	/in. at 192	25 F, wate	er quench			
									Tens	ile Properties		Tests
T T									Ultimate . Yield	Yield	:	
No.	C	Mn	Ч	Ø	Si	Cr	Ni	N	Strengto, ksi ^a	ourengun, ksiª	Elongation, %	Reduction of Area, %
F0353	0.06	1.55	0.005	0.007	0.38	18.34	10.52	0.10	79.1	38.7	59.4	73.9
									84.0	41.1	55.1	70.1
F0353	0.06	1.55	0.005	0.007	0.38	18.34	10.52	0.10	81.2	37.1	56.8	69.9
F0353	0.06	1.55	0.005	0.007	0.38	18.34	10.52	0.10	80.1	39.0	56.8	70.8
F0314	0.07	1.53	0.009	0.008	0.29	18.54	10.69	0.12	85.5	42.9	57.5	70.4
F0314	0.07	1.53	0.009	0.008	0.29	18.54	10.69	0.12	83.7	39.0	55.0	70.4
F0368	0.08	1.49	0.005	0.005	0.25	18.75	10.50	0.12	83.1	43.3	58.6	77.3
F0368	0.08	1.49	0.005	0.005	0.25	18.75	10.50	0.12	87.4	45.0	55.0	50.8
F0379	0.08	1.55	600.0	0.006	0.21	18.87	10.86	0.10	84.9	41.4	55.2	70.6
F0379	0.08	1.55	0.009	0.006	0.21	18.87	10.86	0.10	84.1	40.0	57.0	73.0
F0379	0.08	1.55	600.0	0.006	0.21	18.87	10.86	0.10	84.5	40.0	53.8	72.5
F0372	0.06	1.46	0.005	0.010	0.16	18.43	10.41	0.10	82.9	41.1	57.0	72.8
									84.8	41.2	55.3	68.9
F0372	0.06	1.46	0.005	0.010	0.16	18.43	10.41	0.10	85.5	41.2	55.2	72.6
F0380	0.07	1.55	0.008	0.005	0.35	18.04	10.74	0.10	82.0	40.4	56.5	72.5
F0380	0.07	1.55	0.008	0.005	0.35	18.04	10.74	0.10	85.4	36.4	68.0	76.6
F0380	0.07	1.55	0.008	0.005	0.35	18.04	10.74	0.10	83.9	41.3	56.2	73.5
F0372	0.06	1.46	0.005	0.010	0.16	18.43	10.41	0.10	82.5	42.5	53.5	71.4
E1463	0.07	1.44	0.007	0.007	0.28	18.91	10.47	0.10	83.2	40.0	56.6	73.0
E1463	0.07	1.44	0.007	0.007	0.28	18.91	10.47	0.10	83.7	42.5	55.0	73.0
E1463	0.07	1.44	0.007	0.007	0.28	18.91	10.47	0.10	82.2	40.4	56.4	72.7
E1462	0.07	1.40	0.008	0.008	0.23	18.77	10.41	0.10	82.5	42.0	53.3	71.4
F0391	0.08	1.76	0.008	0.005	0.22	18.79	10.85	0.12	79.6	39.6	59.5	71.4
F0392	0.08	1.85	0.010	0.005	0.21	18.95	10.50	0.11	87.8	44.6	53.7	73.5

12

76.9	72.5	72.5	72.5	71.4	66.8	1.07	72.7	73.0	72.7	74.7	74.7	78.0	74.3	73.0	76.7	73.5	73.3	75.4	71.8	72.5	72.9	73.2	76.3	75.6	75.8	74.1	78.0	71.3	75.8	75.0	76.7	75.2	78.6	73.9
55.5	55.5	59.0	57.5	53.5	56.7	54.2	58.0	57.5	56.0	52.9	51.3	58.8	56.5	53.6	56.5	58.0	58.0	56.6	60.09	58.0	55.0	55.0	57.6	57.0	57.1	56.5	57.3	68.0	62.5	57.8	55.0	55.1	53.7	55.4
42.8	44.6	36.1	37.4	43.6	40.5	40.1	38.9	36.7	41.2	42.8	41.5	37.9	41.4	41.0	41.9	38.0	40.0	42.7	36.5	37.3	38.8	41.2	40.1	42.3	41.8	41.0	40.5	51.2	48.5	42.4	41.0	44.3	45.0	44.0
84.6	84.1	81.2	83.4	83.8	85.9	85.4	82.9	84.6	81.8	88.6	84.7	81.4	85.0	85.4	83.0	79.2	83.0	85.1	82.0	81.0	83.5	84.0	82.4	81.6	86.5	82.0	83.0	85.0	93.4	85.9	85.5	87.6	90.3	86.4
0.10	0.10	0.10	0.11	0.11	0.10	0.11	0.10	0.10	0.10	0.11	0.11	0.13	0.13	0.11	0.11	0.12	0.12	0.12		0.12	0.12	0.12		0.12		0.12	0.12	0.12		0.12		0.12		0.10
10.64	10.64	10.64	10.71	10.71	10.41	10.71	10.49	10.49	10.49	10.82	10.82	10.72	10.72	10.50	10.50	10.51	10.51	10.51		10.65	10.42	10.42		10.42		10.53	10.53	10.53		10.42		10.53		10.41
18.70	18.70	18.70	18.55	18.55	18.77	18.55	18.76	18.76	18.76	19.08	19.08	18.77	18.77	18.95	18.95	18.62	18.62	18.62		18.74	18.39	18.39		18.39		18.56	18.56	18.56		18.39		18.56		18.77
0.28	0.28	0.28	0.21	0.21	0.23	0.21	0.31	0.31	0.31	0.25	0.25	0.22	0.22	0.21	0.21	0.27	0.27	0.27		0.21	0.14	0.14		0.14		0.39	0.39	0.39		0.14		0.39		0.23
0.005	0.005	0.005	0.005	0.005	0.008	0.005	0.005	0.005	0.005	0.005	0.005	0.005	0.005	0.005	0.005	0.005	0.005	0.005		0.008	0.005	0.005		0.005		0.005	0.005	0.005		0.005		0.005		0.008
0.014	0.014	0.014	0.005	0.005	0.008	0.005	0.005	0.005	0.005	0.013	0.013	0.013	0.013	0.010	0.010	0.014	0.014	0.014		0.007	0.005	0.005		0.005		0.014	0.014	0.014		0.005		0.014		0.008
1.45	1.45	1.45	1.38	1.38	1.40	1.38	1.47	1.47	1.47	1.80	1.80	1.79	1.79	1.85	1.85	1.47	1.47	1.47		1.49	1.42	1.42		1.42		1.50	1.50	1.50		1.42		1.50	1	1.40
0.07	0.07	0.07	0.07	0.07	0.07	0.07	0.06	0.06	0.06	0.08	0.08	0.08	0.08	0.08	0.08	0.07	0.07	0.07		0.07	0.08	0.08		0.08		0.06	0.06	0.06		0.08		0.06	1	0.07
E1466	E1466	E1466	E1467	E1467	E1462	E1467	E1465	E1465	E1465	53270	53270	53271	53271	53273	53273	E1499	E1499	E1499		E1496	E1497	E1497		E1497		E1498	E1498	E1498		E1497		E1498		E1462

TABL	E IAC	hemistry (ana room v	emperature H	: mecnanu: eat Treati	al properu ment: 1 h/	tes of Type /in. at 1921	3041V sta 5 F, wate:	r quench	10 to U.16 pe	rcent nutrogen	Continued
									Tens	ile Propertie:	s-Transverse	\mathbf{Tests}
Heat No.	C	Mn	Ъ	Ø	Si	Ċ	Ņ	Z	Ultimate Yield Strength, ksi ^a	Yield Strength, ksi ^a	Elongation,	Reduction of Area, %
F0390	0.08	1.79	0.013	0.005	0.22	18.77	10.72	0.13	91.3	48.3	53.4	73.0
F0389	0.08	1.80	0.013	0.005	0.25	19.08	10.82	0.11	85.9	40.5	53.7	73.5
F0134	0.07	1.53	0.009	0.008	0.29	18.54	10.69	0.12	85.2	39.8	54.0	73.5
									85.7	42.0	55.0	75.2
E1494	0.06	1.52	0.007	0.008	0.23	18.58	10.56	0.12	88.6	45.7	57.6	69.1
									88.2	46.5	58.2	81.6
E1496	0.07	1.49	0.007	0.008	0.21	18.74	10.65	0.12	86.2	42.5	61.5	75.0
									88.2	42.1	74.4	77.3
E1494	0.06	1.52	0.007	0.008	0.23	18.58	10.56	0.12	84.9	42.0	56.0	70.8
E1494	0.06	1.52	0.007	0.008	0.23	18.58	10.56	0.12	87.2	40.7	56.0	69.5
E1496	0.07	1.49	0.007	0.008	0.21	18.74	10.65	0.12	84.2	39.2	55.0	67.8
E1496	0.07	1.49	0.007	0.008	0.21	18.74	10.65	0.12	85.9	42.5	55.0	72.5
E1492	0.08	1.45	0.006	0.013	0.39	18.87	10.27	0.12	93.0	34.8	56.9	73.0
D8686	0.08	1.47	0.015	0.013	0.27	18.71	11.00	0.11	81.1	40.0	55.5	68.4
D8685	0.08	1.45	0.020	0.015	0.30	18.84	10.85	0.13	84.9	42.9	56.0	71.8

^a To convert ksi to megapascals, multiply by 6.894757.

41.6

Avg.

			\$			•		10		(0	
				H	eat Treatı	ment: 1 h/	in. at 192	5 F, wate	r quench	Tensile	Properties	-Transver	se Tests
Heat No.	C	Mn	Ч	ß	Si	C	Ni	Мо	Z	Ultimate Yield Strength, ksi ^a	Yield Strength, ksi ^a	Elonga- tion, %	Reduction of Area, %
E1472	0.07	1.69	0.017	0.009	0.35	16.92	13.51	2.21	0.10	85.0	43.3	54.0	71.3
E1472	0.07	1.69	0.017	0.009	0.35	16.92	13.51	2.21	0.10	85.8 85.4	44.5 45.3	58.0 54.0	72.6
										87.0	47.7	54.0	72.6
E1472	0.07	1.69	0.017	0.009	0.35	16.92	13.51	2.21	0.10	86.2	47.3	54.0	71.3
				1				1		87.0	42.9	54.5	71.3
E1473	0.08	1.69	0.015	0.005	0.36	16.93	13.54	2.17	0.11	85.4	45.3	55.0	74.2
E1473	0.08	1 60	0 015	0.005	0.36	16 03	12 54	0 17	11 0	87.8 8.6	47.3	52.0	60.9 75 0
	00.0	20.1	010.0	000.0		00.01	10.01		11.0	86.6 8	45.3	0. FU	6 74
E1473	0.08	1.69	0.015	0.005	0.36	16.93	13.54	2.17	0.11	85.4	41.1	57.5	75.8
										89.0	44.9	55.0	69.1
E1473	0.08	1.69	0.015	0.005	0.36	16.93	13.54	2.17	0.11	85.0	43.7	55.2	75.4
										88.2	44.1	55.4	75.4
E1474	0.08	1.62	0.015	0.007	0.33	17.13	13.50	2.37	0.11	89.9	45.1	55.0	79.7
		:								88.2	44.9	55.0	71.3
E1474	0.08	1.62	0.015	0.007	0.33	17.13	13.50	2.37	0.11	87.8	43.1	55.5	75.4
						I				88.2	47.1	55.1	74.6
E1474	0.08	1.62	0.015	0.007	0.33	17.13	13.50	2.37	0.11	89.0	44.1	61.0	72.6
										89.2	39.3	56.5	73.4
F0244	0.07	1.63	0.016	0.009	0.27	16.90	13.65	2.32	0.12	87.8	42.5	57.3	72.6
										86.6	43.7	58.5	72.1
F0228	0.06	1.69	0.010	0.006	0.27	17.04	13.58	2.29	0.13	88.2	45.0	54.3	69.1
										87.4	42.9	53.0	70.4

TABLE 1B-Chemistry and room temperature mechanical properties of Type 316N stainless steel, 0.10 to 0.16 percent nitrogen.

~
- E
nı
5
5
Ŭ
د ا
e
60
t^{r}
u
en
õ
36
~
16
0.
0
1
10
0
jə:
ste
ŝ
ŝ
nl_{i}
tiı
sta
20
N
16
ŝ
96
уp
Ē
4
of
es of
ties of
verties of
operties of
properties of
! properties of
al properties of
ical properties of
unical properties of
hanical properties of
echanical properties of
mechanical properties of
e mechanical properties of
ure mechanical properties of
ature mechanical properties of
erature mechanical properties of
perature mechanical properties of
mperature mechanical properties of
temperature mechanical properties of
m temperature mechanical properties of
om temperature mechanical properties of
room temperature mechanical properties of
d room temperature mechanical properties of
and room temperature mechanical properties of
t and room temperature mechanical properties of
try and room lemperature mechanical properties of
istry and room temperature mechanical properties of
mistry and room temperature mechanical properties of
temistry and room temperature mechanical properties of
Chemistry and room temperature mechanical properties of
-Chemistry and room temperature mechanical properties of
3-Chemistry and room temperature mechanical properties of
1BChemistry and room temperature mechanical properties of
E 1B-Chemistry and room lemperature mechanical properties of
LE 1BChemistry and room temperature mechanical properties of
BLE 1BChemistry and room temperature mechanical properties of
ABLE 1B-Chemistry and room temperature mechanical properties of

se Tests	Reduction of Area, %	71.8	68.4 57.3	67.5	71.4	70.8	71.4	67.0	65.9	71.0	69.1	6.99	66.4	66.0	70.9	71.3	74.2	73.4	68.2	71.3	71.4	75.5	64.7	70.2	62.0	68.2
-Transvers	Elonga- tion, <i>%</i>	57.3	57.3 47.5	48.4	50.5	52.0	51.0	49.1	49.7	53.0	53.7	56.5	57.6	57.5	58.5	59.6	60.09	0.09	53.7	0.09	52.5	50.0	46.5	47.5	53.0	48.0
Properties-	Yield Strength, ksi ^a	42.6	44.0 41.9	45.9	43.2	41.2	42.5	41.9	45.0	41.1	45.1	45.1	45.1	45.7	41.7	41.5	43.7	42.5	44.0	45.1	43.7	46.2	46.0	52.5	43.7	47.4
Tensile	Ultimate Yield Strength, ksi ^a	85.5 21.5	87.9 85.9	89.9	87.7	85.2	87.9	83.4	88.3	83.9	90.5	90.3	87.3	87.5	84.5	82.3	83.3	83.3	81.3	86.0	83.1	84.9	83.5	87.9	83.9	90.4
quench	Z	0.12	0.12		0.12		0.12		0.12		0.11		0.11		0.11		0.11		0.11		0.11		0.11		0.11	
F, water	Mo	2.35	2.32		2.32		2.35		2.35		2.37		2.37		2.17		2.17		2.17		20.0		2.20		2.20	
n. at 1925	Ni	13.58	13.65		13.65		13.58		13.58		13.50		13.50		13.54		13.54		13.54		13.35		13.35		13.35	
aent: 1 h/i	C	16.96	16.90		16.90		16.96		16.96		17.13		17.13		16.93		16.93		16.93		16.92		16.92		16.92	
at Treatr	ž	0.38	0.27		0.27		0.38		0.38		0.33		0.33		0.36		0.36		0.36		0.34		0.34		0.34	
He	ß	0.012	600.0		0.009		0.012		0.012		0.007		0.007		0.005		0.005		0.005		0.010		0.010		0.010	
	Ь	0.014	0.016		0.016		0.014		0.014		0.015		0.015		0.015		0.015		0.015		0.016		0.016		0.016	
	Mn	1.65	1.63		1.63		1.65		1.65		1.62		1.62		1.69		1.69		1.69		1.84		1.84		1.84	
	C	0.07	0.07		0.07		0.07		0.07		0.08		0.08		0.08		0.08		0.08		0.07		0.07		0.07	
	Heat No.	F0220	F0244		F0244		F0220		F0220		E1474		E1474		E1473		E1473		E1473		F0162		F0162		F0162	

68.2	69.3	65.9	61.1	72.5	69.3	71.4	69.3	72.5	69.3	72.5	74.5	74.5	73.9	71.6	68.8	71.8	70.4	73.5	66.69	72.5	71.4	73.9	71.4	69.3	71.4	69.3	73.5	73.5	72.7	71.3	73.6	71.4	74 0
48.5	50.7	50.2	47.2	51.2	52.4	55.0	51.2	54.2	52.2	50.7	57.5	58.2	54.2	53.0	52.5	52.0	52.0	52.8	52.2	53.5	52.4	55.0	53.7	49.2	52.5	52.0	54.0	52.5	51.0	52.3	53.7	50.2	50 2
40.6	40.0	40.9	42.5	42.8	45.0	43.0	42.1	42.5	43.2	45.0	42.5	42.2	45.0	41.0	44.4	42.0	44.5	39.7	42.2	41.0	45.0	42.0	43.0	47.0	41.0	45.0	43.5	45.0	43.9	45.1	42.9	47.5	46.0
83.4	80.5	83.0	83.5	86.3	87.5	85.0	86.0	83.3	87.1	89.9	87.4	86.1	6 .68	85.7	86.4	84.4	86.8	83.4	85.3	84.9	86.2	84.3	88.1	6.06	82.2	87.9	87.7	89.9	87.4	88.6	85.4	88.1	88.5
0.11		0.11		0.12		0.12		0.13		0.13		0.13		0.11		0.11		0.11		0.11			0.14		0.11		0.13		0.13		0.12		
2.18		2.18		2.32		2.32		2.38		2.29		2.29		2.38		2.38		2.38		2.38			2.32		2.38		2.29		2.29		2.32		
13.76		13.76		13.61		13.61		13.58		13.58		13.58		13.98		13.98		13.98		13.98			13.51		13.98		13.58		13.58		13.65		
16.95		16.95		16.81		16.81		17.06		17.04		17.04		17.15		17.15		17.15		17.15			16.82		17.15		17.04		17.04		16.90		
0.28		0.28		0.31		0.31		0.33		0.27		0.27		0.42		0.42		0.42		0.42			0.27		0.42		0.27		0.27		0.27		
0.016		0.016		0.012		0.012		0.011		0.006		0.006		0.013		0.013		0.013		0.013			0.008		0.013		0.006		0.006		0.009		
0.022		0.022		0.014		0.014		0.015		0.010		0.010		0.006		0.006		0.006		0.006			0.013		0.006		0.010		0.010		0.016		
1.71		1.71		1.71		1.71		1.72		1.69		1.69		1.79		1.79		1.79		1.79			1.56		1.79		1.69		1.69		1.63		
0.07		0.07		0.06		0.06		0.07		0.06		0.06		0.08		0.08		0.08		0.08			0.06		0.08		0.06		0.06		0.07		
F0216	I	F0216	1	F0226	I	F0226	I	F0227		F0228		F0228		F0229		F0229		F0229		F0229			F0230		F0229		F0228		F0228	:	F0244		

ed
nu
ti.
<u>6</u>
Ŷ
nen
60.
it
r L
en i
r <u>c</u>
pe be
9
2.1
6
5
7(
0
el,
ste
\$0
68
'n
tai
00
N S
H
() ()
ě
ŝ
N
5
s of 1
ies of 1
erties of ¹
perties of '
properties of 1
l properties of 1
cal properties of ¹
nical properties of ¹
hanical properties of ¹
echanical properties of ¹
mechanical properties of 1
re mechanical properties of 1
ture mechanical properties of ⁵
rature mechanical properties of ⁵
perature mechanical properties of ⁵
smperature mechanical properties of ⁵
temperature mechanical properties of ¹
om temperature mechanical properties of ⁵
room temperature mechanical properties of ⁵
d room temperature mechanical properties of ⁵
and room temperature mechanical properties of ⁵
y and room temperature mechanical properties of ⁵
stry and room temperature mechanical properties of ⁵
nistry and room temperature mechanical properties of ⁵
semistry and room temperature mechanical properties of ¹
Chemistry and room temperature mechanical properties of ¹
BChemistry and room temperature mechanical properties of 1
] 1B—Chemistry and room temperature mechanical properties of 1
LE 1BChemistry and room temperature mechanical properties of ¹
BLE 1B—Chemistry and room temperature mechanical properties of ¹
ABLE 1B-Chemistry and room temperature mechanical properties of 1

Tests	eudction of Area, %	73.9	71.4 68.4	73.9	79.5	72.5	75.5	72.9	71.5	73.3	72.3	71.2	71.9	71.0	71.4	68.2	71.0	69.3	73.9	71.4	76.7	74.5	74.1	71.6	76.7	73.9
ransverse	R Clonga- c ion, %	51.5	51.0 40.3	54.5	50.0	47.1	52.9	51.0	51.4	52.5	56.3	52.5	51.0	51.0	52.9	50.5	51.3	50.3	50.7	50.3	50.0	48.8	55.0	50.3	53.5	51.3
roperties—7	Yield trength, F ksi ^a t	44.8	48.3 46.8	36.9	44.3	48.0	47.2	44.0	45.7	45.0	43.0	43.7	36.3	45.2	41.5	43.3	40.1	45.5	44.2	47.0	46.0	46.7	41.4	44.8	42.3	43.7
Tensile P	Ultimate Yield Strength, S ksi ^a	88.9	93.2 78.7	84.7	90.9	92.7	88.7	88.88	86.9	90.5	85.3	86.8	86.2	90.7	86.1	88.9	86.9	6.16	87.1	91.16	90.8	0.16	85.8	89.7	89.5	89.8
quench	Z	0.12	0 12		0.12		0.12		0.10		0.10		0.10		0.11		0.11		0.11		0.12		0.11		0.12	
F, water	Mo	2.32	2, 32	 	2.32		2.32		2.30		2.30		2.30		2.28		2.28		2.28		2.26		2.28		2.26	
ı. at 1925	Ni	13.76	13 76	, , ,	13.76		13.76		13.50		13.50		13.50		13.55		13.55		13.55		13.42		13.55		13.42	
ent: 1 h/iı	Cr	17.33	17 33		17.33		17.33		16.92		16.92		16.92		17.19		17.19		17.19		17.16		17.19		17.16	
at Treatm	Si	0.25	0 25	, ,	0.25		0.25		0.33		0.33		0.33		0.27		0.27		0.27		0.26		0.27		0.26	
He	22	0.009	0 009)))	0.009		0.009		0.005		0.005		0.005		0.006		0.006		0.006		0.006		0.006		0.006	
	Ь	0.012	0.012)))	0.012		0.012		0.010		0.010		0.010		0.005		0.005		0.005		0.010		0.005		0.010	
	Mn	1.68	1 68)) 	1.68		1.68		1.67		1.67		1.67		1.74		1.74		1.74		1.67		1.74		1.67	
	a	0.07	0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.07	
	Heat No.	F0371	F0371	-	F0371		F0371		F0382		F0382		F0382		F0373		F0373		F0373		F0369		F0373		F0369	

.5 75.9 7 75.9		5 74.8 3 76.6	.0 71.8	1.17 0.1	5 73.9	.0 75.5	5 72.9	.0 73.5	.0 77.3	0.17 0.0	.5 71.9	0.5 70.1	5 66.5	.0 71.4	9.07 0.0	.0 72.9	0.0 71.9	.0 73.5	0.70 67.0	.0 71.0	0.069.3	.0 71.4	1.5 65.9	2 72.9	0.0 64.7	0.0 69.3	0.5 70.2	1.1 73.3	0.0 69.3	0.73.3
43.0 51	46.5 50	48.4 50 51 1 52	42.5 54	46.2 50	44.0 62	38.9 52	39.2 52	43.0 50	40.9 58	46.0 50	45.7 49	40.0 49	42.9 42	43.0 50	42.0 50	32.7 52	43.3 50	52.0 53	43.0 45	40.4 51	41.0 52	42.5 55	44.5 46	41.0 53	44.0 50	44.0 49	42.7 49	43.5 52	43.9 50	37.7 56
89.1 20.7	88.4	91.5 09.3	0.08	90.3	88.4	84.0	83.4	84.8	85.8	87.4	87.4	84.0	86.3	86.0	88.8	82.4	85.0	83.3	82.8	83.0	84.5	82.3	77.3	83.0	86.0	83.9	86.0	84.2	82.5	80.6
0.12	0.12	0 10	01.0	0.12		0.13		0.13		0.13		0.12		0.12		0.12		0.10		0.10		0.10		0.12		0.12		0.12		0.12
2.26	2.26	2,38		2.24		2.22		2.22		2.22		2.07		2.07		2.07		2.39		2.39		2.39		2.24		2.29		2.02		2.29
13.42	13.42	13 50	00.01	13.45		13.70		13.70		13.70		13.66		13.66		13.66		13.78		13.78		13.78		13.45		13.60		13.43		13.70
17.16	17.16	16 92		17.35		16.83		16.83		16.83		16.90		16.90		16.90		16.97		16.97		16.97		17.35		17.00		17.21		16.84
0.26	0.26	0.33		0.26		0.43		0.43		0.43		0.31		0.31		0.31		0.35		0.35		0.35		0.26		0.33		0.28		0.38
0.006	0.006	0.005		0.010		0.009		0.009		0.009		0.011		0.011		0.011		0.012		0.012		0.012		0.010		0.009		0.010		0.010
0.010	0.010	0.010		0.011		0.011		0.011		0.011		0.011		0.011		0.011		0.009		0.009		0.009		0.011		0.012		0.012		0.012
1.67	1.67	1.67		1.73		1.72		1.72		1.72		1.66		1.66		1.66		1.74		1.74		1.74		1.73		1.65		1.72		1.67
0.07	0.07	0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.07		0.06		0.07
F0369	F0369	F0382		F0213		F0188		F0188		F0188		F0190		F0190		F0190		F0214		F0214		F0214		F0213		F0221		F0212		F0189

õ
nı
5
ũ
8
Y
1.
n
ð
20
Ľ.
8
n
ы
E.
à
9
7
0
20
0
7
Ó.
~
ee
<i>st</i>
ŝ
es
lu
11.
stc
~
2
16
90
90
л
E
5
0
ŝ
<i>•</i>
tie
ertie
pertie
ropertie
propertie
al propertie
ical propertie
inical propertie
hanical propertie
echanical propertie
mechanical propertie
e mechanical propertie
tre mechanical propertie
tture mechanical propertie
rature mechanical propertie
perature mechanical propertie
mperature mechanical propertie
temperature mechanical propertie
n temperature mechanical propertie
om temperature mechanical propertie
room temperature mechanical propertie
1 room temperature mechanical propertie
nd room temperature mechanical propertie
and room temperature mechanical propertie
ry and room temperature mechanical propertie
stry and room temperature mechanical propertie
nistry and room temperature mechanical propertie
emistry and room temperature mechanical propertie
Themistry and room temperature mechanical propertie
-Chemistry and room temperature mechanical propertie
B-Chemistry and room temperature mechanical propertie
1 1BChemistry and room temperature mechanical propertie
.E 1BChemistry and room temperature mechanical propertie
3LE 1B-Chemistry and room temperature mechanical propertie
ABLE 1B—Chemistry and room temperature mechanical propertie

of Area, Reduction Tensile Properties—Transverse Tests Elonga-tion, % Strength, ksi⁴ Yield

44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
44
<td Ultimate Yield Strength, ksia Heat Treatment: 1 h/in. at 1925 F, water quench 0.120.120.120.120.120.120.13 0.120.11 0.11 0.11 0.11 0.11 \mathbf{z} 2.182.0229 29 2.022.302.292.292.242.302.302.38 33 Mo 2 2 сi 13.45 13.45 13.70 13.70 13.43 13.63 13.26 13.60 13.60 13.63 13.63 13.58 14.00 ž 16.95 16.88 17.00 17.35 16.88 16.88 17.06 16.8416.84 17.00 6617.21 17.21 Ċ 2 0.280.380.380.280.340.280.330.330.260.340.34 33 8 $\overline{\mathbf{S}}$ 0 o. 0.0120.0100.010 0.010 0.016 0.009 0.009 0.010 0.010 0.011 0.011 0.011 0.011 S 0.0120.012 0.013 0.012 0.012 0.014 0.012 0.012 0.014 0.014 0.0150.0220.011 Ъ 1.72 1.67 1.641.651.73 l.72 1.67 1.72 1.65 1.64 1.64 1.65 Мn 1.71 0.080.07 0.06 0.060.07 0.07 0.07 0.07 0.060.06 0.06 0.07 0.07 υ F0212 F0215 F0216 F0213 F0212 F0215 F0215 F0189 F0189 F0222 F0227 F0221 F0221 Heat No.

			43.5	Avg.										
	72.3	50.7	42.6	84.5										
	73.0	53.1	42.0	83.1	0.12	2.21	13.63	16.72	0.29	0.010	0.005	1.49	0.08	E1493
	73.0	54.0	42.1	87.9				1	, ,					
$ \begin{array}{llllllllllllllllllllllllllllllllllll$	78.8	55.0	39.7	88.3	0.12	2.23	13.55	16.71	0.36	0.007	0.006	1.71	0.07	E1468
	73.4	55.0	41.7	89.9						1		i		
	74.6	53.0	41.7	89.5	0.12	2.23	13.55	16.71	0.36	0.007	0.006	17.1	10.07	E1408
	75.4	55.0	42.7	90.7						1		i		
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	75.4	53.0	42.1	90.3	0.12	2.23	13.55	16.71	0.36	0.007	0.006	1.71	0.07	E1468
$ \begin{array}{llllllllllllllllllllllllllllllllllll$	73.4	53.0	42.7	90.3								i	L c	
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	75.4	53.1	41.1	87.9	0.12	2.23	13.55	16.71	0.36	0.007	0.006	1.71	0.07	E1468
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	70.9	50.5	44.9	90.3								j	1	
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	76.7	54.7	42.7	89.9	0.12	2.23	13.55	16.71	0.36	0.007	0.006	1.71	0.07	E1468
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	74.6	57.7	44.9	89.4									1	
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	72.6	53.9	41.6	85.6	0.12	2.35	13.58	16.96	0.38	0.012	0.014	1.65	0.07	F0220
$ \begin{array}{{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	73.8	55.0	44.1	88.2									;	
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	73.0	57.6	42.9	87.0	0.12	2.35	13.58	16.96	0.38	0.012	0.014	1.65	0.07	F0220
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	74.2	55.0	43.8	88.2										
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	73.4	55.2	44.9	89.4	0.12	2.32	13.65	16.90	0.27	0.009	0.016	1.63	0.07	F0244
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	68.2	60.1	44.1	89.4									1	
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	70.9	56.0	48.9	87.8	0.12	2.32	13.65	16.90	0.27	0.009	0.016	1.63	0.07	F0244
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	72.1	55.1	40.9	87.8										
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	67.8	53.1	45.3	88.2										
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	73.0	55.6	42.5	88.6	0.14	2.32	13.51	16.82	0.27	0.008	0.013	1.56	0.06	F0230
F0222 0.08 1.65 0.013 0.012 0.26 17.66 14.00 2.33 0.11 82.7 41.0 50.1 70.4 F0222 0.08 1.65 0.013 0.012 0.26 17.66 14.00 2.33 0.11 82.7 41.0 50.1 70.4 F0222 0.08 1.65 0.013 0.26 17.66 14.00 2.33 0.11 85.1 43.5 53.0 68.2 F0227 0.07 1.72 0.011 0.33 17.06 13.58 2.38 0.13 86.0 44.5 48.2 68.2 F0230 0.06 1.56 0.013 0.27 16.82 13.51 2.32 0.14 89.0 43.5 54.3 70.4	70.4	50.0	47.2	89.5							1			
F0222 0.08 1.65 0.013 0.012 0.26 17.66 14.00 2.33 0.11 82.7 41.0 50.1 70.4 F0222 0.08 1.65 0.013 0.012 0.26 17.66 14.00 2.33 0.11 87.2 44.5 48.7 68.2 F0222 0.08 1.65 0.013 0.26 17.66 14.00 2.33 0.11 85.1 43.5 53.0 68.2 F0227 0.07 1.72 0.011 0.33 17.06 13.58 2.38 0.13 86.0 44.5 48.2 68.2 65.0 1.72 0.011 0.33 17.06 13.58 2.38 0.13 85.0 44.5 48.2 68.2 66.0 44.5 48.2 68.2 68.2 68.2 68.2 68.2	73.8	54.3	43.3	89.0	0.14	2.32	13.51	16.82	0.27	0.008	0.013	1.56	0.06	+0230
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	70.4	51.7	42.5	85.1									,	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	68.2	48.2	44.5	86.0	0.13	2.38	13.58	17.06	0.33	0.011	0.015	1.72	0.07	F0227
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	67.0	47.7	44.3	85.6								i	!	
F0222 0.08 1.65 0.013 0.012 0.26 17.66 14.00 2.33 0.11 82.7 41.0 50.1 70.4 87.2 44.5 48.7 68.2 -1000	68.2	53.0	43.5	85.1	0.11	2.33	14.00	17.66	0.26	0.012	0.013	1.65	0.08	+0222
F0222 0.08 1.65 0.013 0.012 0.26 17.66 14.00 2.33 0.11 82.7 41.0 50.1 70.4	68.2	48.7	44.5	87.2										I
	70.4	50.1	41.0	82.7	0.11	2.33	14.00	17.66	0.26	0.012	0.013	1.65	0.08	F0222

KADLECEK ON HOT-EXTRUDED 304N AND 316N PIPE 21

		Heat Trea	atment: 1	l h/in. at 1	1925 F, Wat	er Quench	L	
Chemist	ry:							
Heat								
No.	C	Mn	Р	S	Si	Cr	Ni	Ν
J-1295 ^a	0.07	1.36	0.012	0.006	0.43	18.33	10.41	0.10
E-1499 ⁶	0.08	1.49	0.014	0.010	0.23	18.93	10.53	0.12
E-1497¢	0.08	1.44	0.013	0.005	0.15	18.60	10.33	0.12
F-0262d	0.08	1.46	0.005	0.008	0.29	18.70	10.42	0.11
		Heat J.	1295			Heat E	-1499	
Tem-	Ultimate				Ultimate			
pera-	Tensile	Yield		Reduc-	Tensile	Yield		Reduc-
ture,	Strength,	Strength,	Elonga-	tion of	Strength,	Strength,	Elonga-	tion of
deg F	ksi*	ksi ^e	tion, %	Area, %	ksi ^e	ksi e	tion, %	Area, %
100	81.0	39.5	63	75	84.0	35.6	60	75
200	74.8	35.0	50	71	76.8	29.2	55	72
300	70.5	31.9	49	70	72.8	26.9	52	69
400	69.0	29.4	45	67	70.2	22.0	50	68
500	68.0	26.4	45	69	69.7	20.1	50	67
600	68.4	25.4	45	65	69.9	19.2	50	66
700	67.9	25.0	46	65	69.8	19.5	50	68
800	66.0	24.4	47	65	67.4	18,5	51	67
900	61.7	21.9	43	64	64.0	16.8	48	67
1000	60.1	21.4	43	64	60.6	15.3	46	67
1100	53.2	20.5	43	64	58.0	15.7	48	64
1200	47.3	18.5	43	57	51.9	14.9	43	52
1300	40.5	20.0	43	47	43.2	14.5	41	48
		Heat E-	1497			Heat F-	0262	
Tem-	Ultimate		•		Ultimate			
pera-	Tensile	Yield		Reduc-	Tensile	Yield		Reduc-
ture.	Strength.	Strength.	Elonga-	tion of	Strength.	Strength.	Elonga-	tion of
deg F	ksi*	ksi e	tion, %	Area, %	ksie	ksi*	tion, %	Area, $\%$
100	82.8	34 8	60	72	83.7	37.2	78	74
200	77.0	29 6	54	74	76.7	36.5	63	73
300	73.5	25.9	52	72	69.1	32.9	55	75
400	70 3	23.0	51	70	67.5	27 0	54	72
500	69.3	20.9	50	63	65 2	26 2	50	68
600	70 0	20.4	50	65	64.6	24 6	52	67
700	69.8	19.1	50	67	65 9	24 7	50	64
800	68.8	10.1	40	63	63 6	24 6	52	67
900	65 8	18.1	48	68	63 1	24 4	50	66
1000	61.8	15.5	47	68	59.2	22 7	47	66
1100	55 9	14 7	45	67	56.3	20.8	46	64
1200	51.5	15.4	43	60	48.6	20.4	45	48
1300	42.9	14 0	40	43	40.8	20.2	41	41
	10.0	11.0	10		10.0			

TABLE 2A-Hot tensile results of Type 304N stainless steel, 0.10 to 0.16 percent nitrogen.

 a Heat J-1295 tested at strain rate of 0.01 in./in./min to yield, 0.05 in./in./min to fracture.

^b Heat E-1499 tested at strain rate of 0.005 in./in./min to yield 0.05 in./in./min to fracture.

 $^{\rm c}$ Heat E-1457 tested at strain rate of 0.005 in./in./min to yield, 0.05 in./in./min to fracture.

 $^{\rm d}$ Heat F-0262 tested at strain rate of 0.05 in./in./min to yield, 0.1 in./in./min to fracture.

• To convert ksi to megapascals, multiply by 6.894757.

			He	eat Treatme	nt: I h/in. at Hot Tensile	t 1925 F, V Results	Water Qu	ench				
Unemistry Heat No.	C	Mn	H	0.	S	Si	ΰ	•	Ni	Мо		7
F0369 E1478 V0630	0.07 0.08 0.08	1.67 1.65 1.70	0.0	010 005 018	0.006 0.009 0.019	0.26 0.27 0.28	17. 16.9 17.	96 11 11	13.42 13.34 13.42	2.26 2.00 2.00	000	$12 \\ 10 \\ 08^a$
		Heat 1	70369			Heat 1	E1478			Heat V	70630	
	Ultimate Tensile	Yield		Reduction	Ultimate Tensile	Yield	-	Reduc-	Ultimate Tensile	Vield	2 	Reduc-
Temperature, deg F	Strength, ksi ^b	Strength, ksi ^b	Elonga- tion, <i>%</i>	of Area, %	Strength, ksi ^b	Strength ksi ^b	Elonga- tion, %	tion of Area, %	Strength, ksi ^b	Strength, ksi ^b	Elonga- tion, %	tion of Area %
100	87.6	43.3	47.1	75.4	83.1	41.2	50.8	67.5	82.5	41.1	52.3	74.1
200	84.7	40.7	49.5	75.1	79.3	39.5	50.9	76.0	78.5	39.2	52.1	75.2
300	76.5	32.5	50.1	75.3	74.7	34.5	45.7	66.2	74.6	33.7	48.7	72.5
400	75.0	31.5	48.9	73.0	73.8	31.3	46.5	70.2	72.4	31.3	47.7	71.5
500	74.5	28.1	47.7	70.5	72.8	29.5	45.8	66.5	71.7	29.4	45.9	67.1
600	74.3	29.1	46.3	70.1	73.1	28.1	45.1	60.2	72.0	28.7	47.3	69.8
650	75.1	29.5	45.4	67.6	72.5	28.3	45.5	64.6	72.4	28.2	42.7	68.1
200	74.8	28.9	48.3	70.4	74.1	29.8	47.9	66.1	72.0	28.0	47.4	69.2
750	74.5	28.5	48.7	71.1	72.5	28.1	47.2	60.1	74.5	28.7	49.7	68.4
800	74.1	27.2	46.2	62.6	72.1	28.0	46.6	63.2	71.1	26.5	48.1	70.1
850	73.7	27.3	47.4	71.5	72.3	25.6	40.5	66.3	71.0	25.4	45.5	0.66
006	73.4	26.1	45.8	68.3	70.07	23.5	44.8	68.6	68.9	24.0	45.3	67.0
950	73.1	25.0	47.0	69.7	68.0	22.9	37.8	45.0	68.5	23.8	44.5	61.0
1000	71.5	25.2	45.4	68.8	67.0	22.5	46.2	6.99	67.0	22.9	44.8	65.9
1050	70.4	23.7	44.9	68.6	66.8	22.1	46.2	66.1	65.5	23.6	45.3	66.3
1100	66.0	24.2	47.1	67.0	62.8	21.7	42.2	67.7	62.0	22.2	44.3	67.0
1150	64.9	25.2	42.9	67.0	58.8	20.8	41.9	67.7	57.0	21.9	43.4	65.8
1200	59.9	24.3	42.4	59.8	54.8	20.0	43.3	68.8	53.4	22.5	43.9	67.0
1250	55.0	23.8	38.8	51.4	48.7	20.1	47.8	68.1	52.8	21.0	45.3	65.9
1300	48.8	23.1	31.8	40.0	45.3	21.0	47.0	62.5	45.3	20.7	47.7	68.6
^a Included t ^b To convert	o show the p t ksi to meg	gradual imp apascals, mı	rovement v ultiply by 6	with increas 3.894757.	ing nitrogen.							

TABLE 2B-Type 316N stainless steel, 0.10 to 0.16 percent nitrogen.

	Heat Treatme	ent: 1 h/in. a	t 1925 F, Wa	ater Quench	
		Stress-Ru	pture Data		T - M(1)
Heat No.	Tempera- ture, deg F	Stress, ksi ^a	Life, h	Elongation, %	Parameter $C = 20.0$
E1499	1050	44.0	129.2	20.2	33.38
E1499	1050	41.0	262.3	17.5	33.85
E1498	1050	40.0	275.4	20.0	33.88
E1498	1050	38.0	755.8	15.0	34.54
E1498	1050	33.0	3768.5	14.0	35.59
D8685	1200	29.0	57.2	13.7	36.11
D8685	1200	29.0	58.1	11.8	36.12
D8685	1200	29.0	68.7	12.7	36.24
D8685	1200	29.0	73.3	13.1	36.29
D8685	1200	29.0	78.3	13.2	36.34
D8685	1200	29.0	57.3	14.4	36.11
D8685	1200	29.0	64.2	17.5	36.20
D8685	1200	29.0	85.1	13.6	36.40
E1499	1200	29.0	106.8	13.7	36.56
D8686	1200	26.0	180.8	15.2	36.94
D8686	1200	26.0	255.0	12.3	37.19
D8686	1200	26.0	266.0	18.8	37.22
D8686	1200	26.0	189.2	12.5	36.97
D8686	1200	26.0	163.1	12.0	36.87
D8686	1200	26.0	175.9	15.2	36.92
F0262	1200	26.0	161.6	14.6	36.86
E1499	1200	25.0	277.9	10.7	37.25
E1498	1200	25.0	538.1	12.0	37.73
E1498	1200	22.5	952.8	10.0	38.14
E1498	1200	18.5	3836.9	6.0	39.15
E1497	1350	16.0	116.0	19.2	39.93
E1498	1350	13.5	310.9	11.0	40.71
E1498	1350	12.0	603.2	15.0	41.21
E1498	1350	9.0	2797.7	10.0	42.43
E1498	1500	6.0	526.1	10.0	44.53
E1498	1500	5.2	838.5	8.0	44.93
E1498	1500	4.2	2397.1	2.0	45.82

TABLE 3—CIW stress-rupture data for smooth bars, transverse locations of Type 304N.

^a To convert ksi to megapascals, multiply by 6.894757.

the combined effect of carbon plus nitrogen is shown. The carbon contents of these heats varied from 0.06 to 0.08 percent). By grouping heats according to various nitrogen ranges, it was possible to look at a series of probability analyses on both 304N and 316N material. The probability analyses shown in Figs. 2 and 3 indicate that with a 0.10 to 0.16 percent nitrogen addition either of these alloys in the form of heavy wall pipe would meet an 80 000 psi (552 MPa) ultimate tensile strength and a 35 000 psi (241 MPa) yield strength specification. The probability data were determined by first tabulating a frequency distribution of all data points, and then

	Heat Treatme	ent: I n/m. a	at 1925 r, wa	ater Quench	
		Stress-Ru	pture Data		Largon Miller
Heat No.	Tempera- ture, deg F	${\mathop{\rm Stress}\limits_{{\mathop{\rm ksi}}^a}}$	Life, h	Elongation, %	Parameter $C = 20.0$
F0244	1100	38.0	104.2	18.0	34.34
F0244	1100	38.0	88.8	18.0	34.23
F0230	1100	38.0	110.2	10.2	34.38
E1493	1100	38.0	67.9	16.1	34.05
F0244	1100	40.0	42.9	20.0	33.74
F0230	1100	40.0	105.2	15.5	34.35
F0244	1100	40.0	62.8	19.7	34.00
F0244	1100	37.0	128.5	17.6	34.48
E1493	1100	33.0	235.8	11.5	34.90
F0244	1100	28.0	7055.8	9.0	37.20
F0230	1200	29.0	289.8	27.8	37.28
E1493	1200	31.0	94.1	13.3	36.47
F0230	1200	26.0	501.0	27.2	37.68
F0244	1200	26.0	260.5	14.1	37.21
E1493	1200	27.0	388.7	14.2	37.49
F0244	1200	27.0	416.0	18.0	37.54
F0244	1200	24.0	1105.0	21.0	38.25
F0244	1200	21.0	3552.6	21.0	39.09
F0225	1200	17.0	15791.0	31.6	40.16
F0244	1350	17.0	295.3	31,2	40.65
E1493	1350	20.0	110.8	41.2	39.90
E1493	1350	16.0	385.7	54.4	40.88
F0244	1350	15.0	618.3	60.0	41.25
F0244	1350	13.5	1172.7	58.0	41.75
F0244	1350	11.0	2863.1	65.0	42.45
F0244	1500	8.0	486.4	59.0	44.46
F0244	1500	6.5	1482.8	56.0	45.41
F0244	1500	5.2	4648.5	33.0	46.38

TABLE 4-CIW stress-rupture data for smooth bars, transverse locations of Type 316N.

. . /.

^a To convert ksi to megapascals, multiply by 6.894757.

calculating the cumulative percentage through each frequency interval. The cumulative percentages are then plotted on the probability paper [1].²

A series of hot tension tests was completed on four heats of 304N and three heats of 316N stainless steel. The test results along with "least squares" regression curves of the ultimate tensile and yield strengths are plotted in Figs. 4 and 5. The actual data are tabulated in Tables 2A and 2B. Strengthwise, these results are consistently higher than for the standard material without nitrogen in the same product form. (Average data for 304 and 316 without nitrogen are shown as dashed lines on the figures.)

Stress-rupture results for both materials are tabulated in Tables 3 and 4.

² The italic numbers in brackets refer to the list of references appended to this paper.

E-1498.
Heat
steel,
stainless
304N
Type
for
data.
Creep
5
TABLE

Heat Treatment: 1 h/in. at 1925 F, Water Quench

Ľ	Pomonon true				, ,	lime n to				Minimum-
Specimen No.	deg F	, Stress, ksi ^a	0.01%	0.05%	0.1%	0.2%	0.5%	1.0%	Rupture	Oreep Kate, 10 ⁻⁴ %/h
5	1050	40.0		-	5	10	31	93	275.4	72
18	1050	38.0		3.2	7.5	18	45	273	755.8	18
17	1050	33.0	•	•	15	42	225			4.0
23	1050	20.0	35	730		disconti	nued at 978	8.8 h		0.18
11	1050	14.0	60	:		disconti	nued at 150	65.0 h		0.045
22	1100	15.0	130		-	liscontinued	l at 815 h			0.13
×	1200	25.0			0.8	2.6	19	77	538.1	80
19	1200	22.5			2.5	12	72	182	952.8	800
21	1200	18.5	:		40	150	465	1025		9.5
7	1200	10.0	34	815		disconti	nued at 98	5 h		0.46
20	1200	7.0	25	•		disconti	nued at 96	5.6 h		0.12
4	1300	7.5	9	115	395		discontinue	3d at 648 h		1.5
°	1350	13.5					:	:	310.9	:
16	1350	12.0	:	3.0	8.8	23	76	144	603.2	58
15	1350	0.6	1.0	18	63	255	785	1177	2797.7	4.6
10	1350	4.0	38		þ	scontinued	at 965.3 h			0.27
2	1350	3.0	100	450		disconti	nued at 98	3.3 h		0.10
6	1500	6.0			:		:	:	526.1	33
24	1500	5.2		4.5	16	56	195	358	838.5	22
14	1500	4.0	2.5	60	245	615				2.6
12	1500	2.0	20	75		disconti	nued at 124	16 g h		0.28

^a To convert ksi to megapascals, multiply by 6.894757.

26 NITROGEN ADDITIONS TO AUSTENITIC STAINLESS STEEL

Heat F-0244.
stainless steel,
$Type \ 316N$
data for
6-Creep
TABLE

L					L	'ime, h to				Minimum-
Specimen No.	ı emperature deg F	s, Stress, ksi ^a	0.01%	0.05%	0.1%	0.2%	0.5%	1.0%	Rupture	Creep Kate, 10 ⁻⁴ %/h
28	1100	38.0		0.5	1.7	5.8	25	45	140.2	180
13	1100	38.0			2.8	7.3	22	44	88.88	218
×	1100	34.0			4.2	13	42	<u>93</u>	247.7	69
24	1100	28.0		0.7	3.0	19	115	720		4.4
4	1200	27.0	•		0.55	1.5	5.7	26	416.0	115
22	1200	24.0	•	1.2	2.7	5.7	17	56	1105.0	49
15	1200	21.0	:		5.0	10	85	328		20
18	1200	12.5	5	300		discont	inued at 69	$4.1\mathrm{h}$		0.96
23	1200	7.5	10	280		discont	inued at 93	2.4 h		0.37
17	1200	7.5	6.3	350		discont	inued at 98	9.0 h		0.28
10	1350	15.0			5.4	11	23	44	618.3	175
26	1350	13.5	0.7	4.2	8.7	18	42	83	1172.7	120
14	1350	11.0	2.5	17	28	54	132	250		38
12	1350	5.0	14	310	880	q	iscontinued	at 864.6 l	-	06.0
11	1350	3.0	30			q	iscontinued	at 1092.5	Ч	0.11
19	1500	8.0			1.9	4.1	11	22	486.4	450
16	1500	6.5	:	3.5	7.5	17	39	68	1482.8	126
6	1500	5.2	0.5	8.0	25	65	190	330		24
25	1500	3.0	50	140	265	460	discon	tinued at	478.6 h	4.0
21	1500	2.0	5	48	180	q	iscontinued	at 1295.6	h	0.46

N filler metal.	
pipe and 308.	
nalysis of 304N	
s. Chemical a	
ss steel weld tests	
d 316N stainle	
] 7—304N an	
FABLE	

	E						Ch	emistry				
Material	Lype of Weld	Test Lab	C	Mn	Si	S	Р	ڻ ا	Ni	Z	Cu	లి
304N base metal		CIW	0.07	1.45	0.26	0.006	0.015	18.36	9.85	0.13	0.08	0.06
T-7746 weld	manual electric arc	Arcos CIW	$\begin{array}{c} 0.036\\ 0.04 \end{array}$	$2.32 \\ 2.25$	$0.65 \\ 0.64$	0.011	0.008 0.008	$19.81 \\ 20.35$	$\begin{array}{c} 10.44 \\ 9.94 \end{array}$	$\begin{array}{c} 0.15\\ 0.153\end{array}$: :
T-7726 weld	manual electric arc	Arcos CIW	0.038 0.030	$\begin{array}{c} 2.29\\ 2.23\end{array}$	$\begin{array}{c} 0.62 \\ 0.66 \end{array}$	$\begin{array}{c} 0.018 \\ 0.018 \end{array}$	0.009 0.008	20.41 21.00	10.33 9.70	$\begin{array}{c} 0.164 \\ 0.171 \end{array}$: :	::
T-7730	automatic sub-arc	Arcos CIW	$0.058 \\ 0.06$	$\begin{array}{c}1.77\\1.70\end{array}$	$0.85 \\ 0.69$	NA⁴ 0.012	NA⁴ 0.018	20.20 20.09	$\begin{array}{c} 10.25\\ 9.58\end{array}$	0.109 0.113	::	::
304N composition range		:	0.08 max	2.0 max	0.75 max	0.03 max	0.03 max	18.0 to 20.0	8.0 to 11.0	0.10 to 0.16	::	::
308 Fiiler metal composition rang	Ð		0.08 max	2.50 max	0.90 max	0.03 max	0.04 max	18.0 to 21.0	9.0 to 11.0	NA 		: :

^a NA = not available.

filler metal.
oipe and 316N
of 316N p
analysis c
Chemical
d tests.
teel wel
stainless s
and 316N
304N
TABLE

	J. T.	to T						Chem	iistry				ĺ
Material	Weld	Lab	C	Mn	Si	ß	Ч	ڻ ت	Ni	N	Mo	Cu	S C
316N Base metal	•	CIW	0.07	1.64	0.41	0.009	0.005	16.92	13.58	0.12	2.34	0.08 (.06
T-7745 Weld	manual electric arc	Arcos CIW	$\begin{array}{c} 0.036\\ 0.05 \end{array}$	$\begin{array}{c} 1.98\\ 2.15\end{array}$	$0.32 \\ 0.43$	0.013 0.016	0.011 0.010	$\frac{18.15}{17.70}$	12.71 12.50	$\begin{array}{c} 0.150\\ 0.133 \end{array}$	$2.40 \\ 2.65$	· · · ·	::
T-7727 Weld	manual electric arc	Arcos CIW	0.031 0.06	2.35 1.89	0.63 0.94	$0.015 \\ 0.010$	$0.013 \\ 0.013$	19.75 19.39	$\frac{11.64}{13.13}$	$0.170 \\ 0.210$	$\begin{array}{c} 2.16\\ 2.22 \end{array}$::	::
T-7710 Weld	automatic sub-arc	Arcos CIW	$\begin{array}{c} 0.048 \\ 0.05 \end{array}$	$\begin{array}{c} 1.99\\ 1.85 \end{array}$	$0.82 \\ 0.95$	$0.011 \\ 0.013$	$\begin{array}{c} 0.019\\ 0.018\end{array}$	$19.46\\19.32$	12.85 13.27	0.111 0.112	$\begin{array}{c} 2.22\\ 2.24\end{array}$::	::
316N Composi- tion range		• •	0.08 max	2.00 max	0.75 max	0.03 max	0.03 max	16.0 to 18.0	11.0 to 14.0	0.10 to 0.16	2.0 to 3.0	÷	÷
316 Filler metal composition r	ange		0.08 max	2.5 max	0.90 max	0.03 max	0.04 max	17.0 to 20.0	11.0 to 14.0	: : : : : :	2.0 2.5		::

nidgage	
approximately 1	
is at	
interface	
retal	
weld base n	
ipe (
4N p	1 S).
ł 30.	imer
n weldec	test spec
results o	ngth on
Tensile	Le Le
weld tests.	
steel	
stainless	
316N	
and	
304N	
ြ	
TABLE	
· • ·	

		- Carbon Content	0.04	0.04	0.03	0.03	0.06	0.06	0.04	0.03	0.06	0.04	0.03	0.06	0.04	0.03	0.06	0.04	0.03	0.06
ta	Content	Arcos	0.15	0.15	0.164	0.164	0.109	0.109	0.15	0.164	0.109	0.15	0.164	0.109	0.15	0.164	0.109	0.15	0.164	0.109
eld Metal Da	Nitrogen	CIW	0.153	0.153	0.171	0.171	0.113	0.113	0.153	0.171	0.113	0.153	0.171	0.113	0.153	0.171	0.113	0.153	0.171	0.113
M	Content	Arcos	0.0	0.0	4.2	4.2	3.9	3.9	0.0	4.2	3.9	0.0	4.2	3.9	0.0	4.2	3.9	0.0	4.2	3.9
	Ferrite (CIW	1.8	1.8	3.4	3.4	3.4	3.4	1.8	3.4	3.4	1.8	3.4	3.4	1.8	3.4	3.4	1.8	3.4	3.4
	To iliuno	Location	base metal	weld metal	base metal	base metal	weld metal													
	Reduction	01 Alea,	75.4	79.2	76.6	76.6	77.3	76.2	54.5	71.3	72.9	47.8	43.2	37.0	59.7	73.4	33.8	49.0	40.1	39.0
	Шоман	tion, %	40.0	42.0	39.5	40.0	46.5	44.3	25.5	31.9	33.8	27.0	26.5	30.5	19.0	31.5	21.6	21.9	30.5	24.5
	set Yield	ourengun, ksiª	53.7	53.4	60.2	61.6	60.2	58.2	36.1	34.5	38.7	33.5	30.9	32.1	28.7	31.1	26.9	31.7	28.6	27.6
1114: monto	Tensile	ourengun, ksi ^a	94.6	95.2	97.3	95.9	67.9	98.3	70.8	72.8	74.4	69.4	68.2	72.4	63.0	64.4	65.2	56.5	60.2	60.2
Ē	Tempera-	ture, deg F	75	75	75	75	75	75	600	009	600	800	800	800	1000	1000	1000	1100	1100	1100
		Weld No.	T-7746	T-7746	T-7726	T-7726	T-7730	T-7730	T-7746	T-7726	T-7730									

^a To convert ksi to megapascals, multiply by 6.894757.
		- Carbon Content	0.06	0.06	0.05	0.05	0.05	0.05	0.06	0.05	0.05	0.06	0.05	0.05	0 06	0.05	0.05	0.06	0.05	0.05
:	Content	Arcos	0.170	0.170	0.150	0.150	0.111	0.111	0.170	0.150	0.111	0.170	0.150	0.111	0.170	0.150	0.111	0.170	0.150	0.111
fetal Data	Nitrogen	CIW	0.210	0.210	0.133	0.133	0.112	0.112	0.210	0.133	0.112	0.210	0.133	0.112	0.210	0.133	0.112	0.210	0.133	0.112
Weld N	Content	Arcos	3.4	3.4	0.0	0.0	3.4	3.4	3.4	0.0	3.4	3.4	0.0	3.4	3.4	0.0	3.4	3.4	0.0	3.4
	Ferrite (CIW	5.4	5.4	1.2	1.2	1.8	1.8	5.4	1.2	1.8	5.4	1.2	1.8	5.4	1.2	1.8	5.4	1.2	1.8
specimens).	Railuro	Location	base metal	base metal	base metal	base metal	weld metal	weld metal	base metal	weld metal	weld metal ^b	base metal	weld metal	weld metal	base metal	base metal	weld metal ^b	base metal	weld metal	$\mathbf{base} \ \mathbf{metal}$
ength on test	Reduction	20 m ca,	79.9	79.5	79.2	79.2	30.6	30.6	71.7	49.0	20.3	69.1	43.2	40.1	66.4	71.3	21.6	71.3	43.2	69.1
1	Flonge-	tion, %	34.5	29.0	35.5	33.8	24.0	26.9	31.1	24.0	17.0	29.8	24.0	21.5	30.0	33.1	15.0	30.0	23.5	33.5
907, Off	set Yield Strength	ksi ^a	54.5	57.3	54.1	57.8	58.6	58.2	40.1	35.7	37.6	32.9	34.1	36.7	30.6	33.6	34.4	30.1	28.1	33.6
I Thimato	Tensile Strength	ksia	92.2	90.2	90.2	91.4	9.06	89.8	71.9	72.2	63.2	70.8	71.2	69.2	68.3	65.6	59.3	63.8	61.2	62.2
E F	Tempera-	deg F	75	75	75	75	75	75	600	600	600	800	800	800	1000	1000	1000	1100	1100	1100
		Weld No.	T-7727	T-7727	T-7745	T-7745	T-7710	T-7710	T-7727	T-7745	T-7710	T-7727	T-7745	T-7710	T-7727	T-7745	T-7710	T-7727	T-7745	T-7710

 a To convert ksi to megapascals, multiply by 6.894757. b Fractured through weld defect.

TABLE 11-304N and 316N stainless steel weld tests. Stress-rupture results on welded 304N pipe (weld base metal interface is at approximately midgage length on test specimens).

	, Coukor	Content	0.04	0.03	0.06	0.04	0.03	0.06	0.04	0.03	0.06
	Content	Arcos	0.15	0.164	0.109	0.15	0.164	0.109	0.15	0.164	0.109
etal Data	Nitrogen	CIW	0.153	0.171	0.113	0.153	0.171	0.113	0.153	0.171	0.113
Weld M	Content	Arcos	0.0	4.2	3.9	0.0	4.2	3.9	0.0	4.2	3.9
	Ferrite	CIW	1.8	3.4	3.4	1.8	3.4	3.4	1.8	3.4	3.4
	Toilineo	Location	weld metal	base metal	weld metal	weld metal	weld metal	weld interface	weld metal	weld metal ^b	weld metal
	<u> </u>	Liongation,	8.6	3.4	6.1	6.1	5.0	9.4	5.6	5.6	U
		Life, h	232.9	540.3	405.1	408.3	364.3	478.3	177.8	76.3	270.2
	Church	ksi ^a	40.0	40.0	40.0	25.0	25.0	25.0	13.5	13.5	13.5
	Test Tem-	perature, deg F	1050	1050	1050	1200	1200	1200	1350	1350	1350
		Weld No.	T-7746	T-7726	T-7730	T-7746	T-7726	T-7730	T-7746	T-7726	T-7730

^a To convert ksi to megapascals, multiply by 6.894757.

^b Possible weld defect.

^e Cannot measure elongation due to irregular type fracture.

TABLE 12-304N and 316N stainless steel weld tests. Stress-rupture results on welded 316N pipe (weld base metal interface is at approximately midgage length on test specimens).

	Conhor	Content	0.06	0.05	0.05	0.06	0.05	0.05	0.06	0.05	0.05
	Content	Arcos	0.170	0.150	0.111	0.170	0.150	0.111	0.170	0.150	0.111
etal Data	Nitrogen	CIW	0.210	0.133	0.112	0.210	0.133	0.112	0.210	0.133	0.112
Weld M	Content	Arcos	3.4	0.0	3.4	3.4	0.0	3.4	3.4	0.0	3.4
	Ferrite	CIW	5.4	1.2	1.8	5.4	1.2	1.8	5.4	1.2	1.8
	- Doiluno	Location -	base metal	base metal	weld interface	base metal	base metal	weld interface	weld metal	base metal	weld metal ^b
	Flowertion	Elougation,	2.5	2.2	13.0	19.8	14.5	7.1	7.7	42.5	13.0
		Life, h	613.1	739.2	365.4	492.0	687.9	95.2	107.4	264.0	36.9
	Ctroco	ksia	35.0	35.0	35.0	27.0	27.0	27.0	16.0	16.0	16.0
	Test Tem-	deg F	1100	1100	1100	1200	1200	1200	1350	1350	1350
		Weld No.	T-7727	T-7745	T-7710	T-7727	T-7745	T-7710	T-7727	T-7745	T-7710

 $^{^{\}alpha}$ To convert ksi to megapascals, multiply by 6.894757. b Possible weld defect.

The Larson-Miller plots of each set of data are shown in Figs. 6 and 7. Minimum-creep rate data were obtained on single heats of 304N and 316N stainless steel. These data are tabulated and plotted in Tables 5 and 6 and Figs. 8 and 9, respectively.

Welding programs were conducted at Cameron Iron Works as well as under a joint program with the Arcos Corporation. Both manual electric arc and semiautomatic submerged arc welding processes were used in weldability experiments. Some welds were completed using the standard electrodes for 304 and 316 stainless steels. Others were made using electrodes and wire specially prepared to produce a weld deposit containing the same amount of nitrogen as the base metal. In all cases, both 304N and 316N material displayed excellent weldability. If one considers only the tensile strength, the standard E308 and E316 electrodes are adequate. However, for service applications of 1000 F (538 C) or higher, stress rupture and creep properties become very important. The addition of nitrogen to the weld metal significantly increases the stress-rupture strength within the temperature range of 1000 to 1200 F (538 to 649 C). Above 1200 F (649 C), as is the case with wrought material, the benefits of nitrogen diminish. Tables 7 through 12 list the welding test results. (The weld defect mentioned in notes to the tables was the result of an equipment problem rather than material weldability.)

Rotating cantilever beam fatigue tests at room temperature were performed on both 304N and 316N grades, and the data are plotted in the S-N graphs of Figs. 10 and 11, respectively. The fatigue strengths are indicated to be approximately 44.5 and 46.5 ksi, respectively. These values are about 3 ksi above their average yield strengths, suggesting that the early cycles of fatigue testing must have accomplished measurable work hardening.

Conclusion

The data resulting from this testing program have demonstrated, on a production basis, that a controlled addition of nitrogen to both 304 and 316 stainless steels produces a consistent improvement in mechanical properties.

The various specification groups have recognized the importance of the nitrogen-bearing materials and have recently established design stresses for the new Grades 304N and 316N stainless steels.

Acknowledgments

The author is grateful to the management of Cameron Iron Works, Inc., for permission to publish these results. Special thanks is extended to R. David Thomas, Jr., president, Arcos Corporation, for his company's contribution to the welding program.

Reference

[1] Lewis, C. F., "Graphical Statistics," Slide Rule, Houston Engineers Club, June 1951.

High-Nitrogen Austenitic Stainless Steels

REFERENCE: Spaeder, C. E., Domis, W. F., and Brickner, K. G., "High-Nitrogen Austenitic Stainless Steels," Elevated Temperature Properties as Influenced by Nitrogen Additions to Types 304 and 316 Austenitic Stainless Steels, ASTM STP 522, American Society for Testing and Materials, 1973, pp. 35-45.

ABSTRACT: The present paper summarizes the results of a number of investigations in which the effects of nitrogen on the mechanical properties of an austenitic stainless steel were evaluated. The results of studies on commercially produced plate indicated that increasing the nitrogen content of Type 304L steel to the range 0.10 to 0.13 percent increased the room and elevated temperature strength of this steel such that its strength was equivalent to that of Type 304 steel in both the welded and unwelded condition.

The results of room and elevated temperature tension tests on product from four commercial heats of high-nitrogen (0.12 to 0.16 percent) Type 304 steel indicated that the yield and tensile strength of this steel was significantly increased. Creep-rupture tests were too limited to assess the effect of nitrogen on the creep-rupture strength of the steel. Some failures on punch marks of the creep-rupture test specimens were observed suggesting the possibility of some notch sensitivity in this steel.

The results of a limited study on laboratory heats of Type 316L containing nitrogen in the range 0.02 to 0.19 percent indicated a significant increase in the elevated temperature yield and tensile strength of this steel. The creep-rupture strength at 1300 F also increased with increasing nitrogen content.

KEY WORDS: nitrogen, creep rupture strength, austenitic stainless steel, tensile strength, welding, stresses

The results of many investigations reported in the literature indicate that nitrogen in solid solution markedly increases the strength of austenitic stainless steels. Accordingly, as part of an extensive program to develop high strength steels, the U. S. Steel Research Laboratory has studied the effect of nitrogen on the mechanical properties of austenitic stainless steels, such as AISI Types 304 and 304L stainless steels. The initial studies were concerned with the development of a Type 304L steel with a strength equiv-

¹Senior research engineer, research engineer, and section supervisor, respectively, United States Steel Corp., Research Laboratory, 125 Jamison Lane, Monroeville, Pa. 15146.

Steel	Heat No.	С	Mn	Р	s	Si	Ni	Cr	Ν
		High	-nitrog	en Type	304L				
A B	X18338 X18342	$\begin{array}{c} 0.030\\ 0.025\end{array}$	1.47 1.63	0.022 0.026	$\begin{array}{c} 0.024 \\ 0.025 \end{array}$	$\begin{array}{c} 0.64\\ 0.55 \end{array}$	9.72 9.87	19.13 18.87	$\begin{array}{c} 0.10\\ 0.13\end{array}$
		I	Regular	Type 30	04				
С	X20792	0.070	1. 62	0.026	0.026	0.79	9.51	18.66	0.04

TABLE 1-Compositions of stainless steels investigated, percent (Check Analyses).



FIG. 1—Effect of temperature on the yield and tensile strengths of annealed high-nitrogen Type 304L and regular Type 304 steels. (To convert ksi to megapascals, multiply by 6.894757).

	10	1000-h Rupture Strength, at Test Temperature											
Steel	900 F (482 C)	1100 F (593 C)	1300 F (704 C)	1500 F (816 C)									
			pe 304LN										
	ksi (MPa)	ksi (MPa)	ksi (MPa)	ksi (MPa)									
A B	$\begin{array}{ccc} 52.0 & (359) \\ 52.0 & (359) \end{array}$	27.0 (186) 26.0 (179)	11.5 (79 .3) 11.0 (75 .8)	$\begin{array}{c} 4.6 & (31.7) \\ 4.7 & (32.4) \end{array}$									
		$Type \ 3$	04										
С	54.0 (372)	27.5 (190)	11.5 (79.3)	5.6 (38.6)									

TABLE 2-Creep-rupture strength of Types 304LN and 304 stainless steels [1].



FIG. 2—Comparison of tensile strength of welded high-nitrogen Type 304L stainless steel plates with that required to meet ASME code stresses for AISI 304 stainless steel. (To convert ksi to megapascals, multiply by 6.894757).

Steel	Heat No.	С	Mn	Р	S	Si	Cu	Ni	Cr	Ν
D	X24189	0.05	1.53	0.025	0.015	0.52	0.09	9.40	18.77	0.16
Е F	X24191 X24216	0.05 0.04	$1.40 \\ 1.45$	$0.025 \\ 0.025$	$0.011 \\ 0.017$	$0.48 \\ 0.62$	$0.10 \\ 0.10$	$9.30 \\ 9.40$	$18.48 \\ 18.52$	$0.13 \\ 0.14$
G	X24225	0.04	1.63	0.025	0.012	0.50	0.09	9.60	18.18	0.12

TABLE 3-Chemical compositions of the commercial stainless steels investigated, percent.



FIG. 3—Effect of annealing temperature on the room temperature yield and tensile strengths of Type 304N stainless steel. (To convert from ksi to megapascals, multiply by 6.894757).



FIG. 4—Comparison of the room and elevated temperature yield strengths of Types 304N and Type 304 stainless steels. (To convert from ksi to megapascals, multiply by 6.894757).

alent to that of the regular grade Type 304 steel. In these studies $[1]^2$, three 3-in. thick plates from commercial heats (compositions shown in Table 1) were investigated. Note that the steels differ significantly only with respect to carbon and nitrogen. Type 304LN steel contained 0.10 to 0.13 percent nitrogen compared with 0.04 percent for the Type 304 steel, and Type 304LN steel contained 0.03 and 0.025 percent carbon compared with 0.07 percent for the Type 304 steel.

A comparison of the yield and tensile strengths of the three steel plates is shown in Fig. 1. Over the temperature range investigated from room

² The italic numbers in brackets refer to the list of references appended to this paper.



FIG. 5—Comparison of the room and elevated temperature tensile strengths of Types 304N and 304 stainless steels. (To convert from ksi to megapascals, multiply by 6.894757).

temperature through 1200 F (649 C), the strength of the Type 304LN steel is about the same as that of the Type 304 Steel.

Table 2 presents a comparison of the 1000-h rupture strength of the three steels at 900, 1100, 1300, and 1500 F (482, 593, 704 and 816 C) [1]. These data indicate that in the range 900 to 1300 F (482 to 704 C) the 1000-h rupture strength of the steels is about the same. However, at 1500 F (816 C), Type 304 steel is stronger than Type 304LN steel. This is consistent with results in the literature, which indicate that the strengthening effect of nitrogen decreases with increasing temperature.

Because of the importance of weldability of these high-nitrogen steels in pressure vessel applications, plate specimens of the Type 304LN steel were welded by the manual metal-arc process with AWS E308L-16 and E347-16 electrodes. The plates were welded and evaluated in accordance with

		Ste	el F		
Test Temperature, deg F (deg C)	Stress, ksi (MPa)	Time to Rupture, h	Elongation in 1 in. (25.4 mm), %	Reduction of Area, %	Minimum- Creep Rate, %/h
1200 (649)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	99.0 443.1 902.5 1411.9 3197.5 5197.0	13ª 13ª 10ª 9ª 10ª 10ª	28.8 26.0 20.0 18.4 16.0 21.1	$\begin{array}{c} {\rm ND}^{b} \\ 0.00688 \\ 0.00350 \\ 0.00152 \\ 0.00113 \\ 0.00069 \end{array}$
1300 (704)	$\begin{array}{cccc} 25.0 & (172) \\ 20.0 & (138) \\ 17.5 & (121) \\ 15.0 & (103) \end{array}$	76.5 360.7 799.6 1849.4	14ª 21ª 31 27ª	30.8 30.8 34.1 33.4	ND 0.0218 0.0128 0.00502

TABLE 4-Results of creep and creep-rupture tests on Type 304N stainless steel.

^a Failed at gage marks. ^b ND = not determined.

TABLE 5-Chemical compositions of stainless steels investigated, percent.

Steel	С	Mn	Р	s	Si	Ni	\mathbf{Cr}	Mo	Ν
				Typ	oe 316L				
н	0.03	1.45	0.01	0.014	0.63	12.5	17.0	2.44	0.02
				Type	e 316LN				
I	0.02	1.46	0.008	0.013	0.57	12.3	16.6	2 . 50	0.10
J	0.02	1.51	0.008	0.007	0.58	12.2	16.8	2.48	0.14
К	0.02	1.49	0.008	0.011	0.56	12.0	17.0	2.47	0.19

TABLE 6-Effect of nitrogen on the 1300 F (704 C) creep-rupture properties of Type 316L stainless steel.

	Stress, for Rupture in					
Nitrogen Content, %	1000 h	10 000ª h				
	ksi (MPa)	ksi (MPa)				
0.02	12.0 (82.7)	9.4 (64.8)				
0.10	15.0 (103)	10.0 (68.9)				
0.19	17.0 (117)	11.0 (75.8)				

^a Extrapolated values.



FIG. 6—Effect of test temperature on the yield strength of Type 316L stainless steel containing nitrogen. (To convert from ksi to megapascals, multiply by 6.894757).

American Society of Mechanical Engineers (ASME) Boiler and Pressure Vessel Code, Section IX, Welding Qualification. All the welds were sound and exhibited room temperature tensile strengths in the range 88 to 91 ksi (607 to 627 MPa). Shown in Fig. 2 are the results of room and elevated temperature tension tests on specimens machined from the welded plates. Also shown in this figure are the tensile properties required to meet the present ASME code stresses for Type 304 steel. The results of these tests indicate that the strength of the welded Type 304LN plates exceeds the ASME strength requirements for Type 304 steel. Thus, it is concluded that the addition of a small amount of nitrogen to Type 304L steel generally increases its yield and tensile strengths to about the same level as regular Type 304 steel.

Additional studies were conducted to compare the properties of Type 304N steel with Type 304 steel. The chemical compositions of four commercial heats of Type 304N steel that were used in these studies are presented in Table 3. The compositions are typical of regular Type 304 steel except for the nitrogen content, which was 0.12 to 0.16 percent.

The effect of annealing temperature on the room temperature yield and tensile strengths of these Type 304N steels is shown in Fig. 3. Note that the yield and tensile strengths decrease slightly with increasing annealing tem-



FIG. 7—Effect of test temperature on the tensile strength of Type 316L stainless steel containing nitrogen. (To convert ksi to megapascals, multiply by 6.894757).



FIG. 8—Creep-rupture elongation at 1300 F (704 C) for Type 316L stainless steel containing 0.02 and 0.19 percent nitrogen.

perature, but that the strengths at all annealing temperatures are considerably above the strength requirements of the current American Society for Testing and Materials (ASTM) specifications for Type 304 steel.

A comparison of the yield strength of Type 304N steel with the yield strengths reported in the literature [2] for seven heats of Type 304 steel is shown in Fig. 4. The Type 304 data were obtained on annealed bar, plate, tubing, and pipe that, insofar as could be determined, had not been subjected to significant amounts of cold work after the final anneal. Cold work introduced by straightening operations can markedly affect the yield strength of Type 304 steel.

As shown in Fig. 4, the range of yield strengths observed for the Type 304N steel is above the top of the range observed for Type 304 steel at room temperature and 200 F (93 C). For example, at room temperature the range for Type 304N steel is 40.1 to 43.6 ksi (276 to 301 MPa) compared with the range of 31 to 36.9 ksi (214 to 254 MPa) for Type 304 steel. Between 400 and 1200 F (204 to 649 C), the range of yield strengths observed for the Type 304N steel overlaps the high side of the range of yield strengths observed for the Type 304 steel. A similar comparison for tensile strength is shown in Fig. 5. The range of tensile strengths for Type 304N steel is considerably above the top of the range observed for Type 304 steel at all temperatures in the range between room temperature and 1200 F (649 C). Therefore, these data indicate that the addition of nitrogen significantly increases the yield and tensile strengths of Type 304 steel.

Creep-rupture test data at 1200 and 1300 F (649 and 704 C) for one of the Type 304N steels are presented in Table 4 [1]. Note that most of the failures occurred on the punch marks. Because failure on punch marks is also indicative of low notched-bar creep-rupture strength, the effect of nitrogen on the elevated temperature ductility requires additional study. With regard to the effect of nitrogen on the creep-rupture strength, an assessment of the effect has been deferred until the results of additional testing are completed.

Studies also have been made on the effect of nitrogen on the elevated temperature properties of laboratory heats of Type 316L steel. The compositions of the steels investigated are shown in Table 5. Note that the nitrogen content of the four steels varies from 0.02 to 0.19 percent. A plot of the yield strength versus test temperature for the four steels is shown in Fig. 6. Between room temperature and 1200 F (649 C) the yield strength increased with increasing nitrogen content. Likewise, as shown in Fig. 7, the tensile strength increased with increased with increasing nitrogen content.

Given in Table 6 are 1000 and 10 000-h rupture strengths at 1300 F (704 C) of the 0.02, 0.10, and 0.19 percent nitrogen Type 316L steels. These data indicate a beneficial effect of nitrogen on the creep-rupture strength. As shown in Fig. 8, the elongation of 0.19 percent nitrogen steel is initially lower than the elongation of the 0.02 percent nitrogen steel. However, the

elongation of the 0.19 percent nitrogen steel increased with time, whereas the elongation of the 0.02 percent nitrogen steel decreased slightly with time, such that for times greater than about 2200 h at 1300 F (704 C), the steel with the 0.19 percent nitrogen content exhibited higher elongation compared with the 0.02 percent nitrogen steel. The reason for this divergent behavior is not understood.

In summation, the results of these studies indicate a general beneficial effect of nitrogen on the elevated temperature strength of the austenitic steels. However, failure on punch marks in some of the creep-rupture tests suggest the possibility of some notch sensitivity in Type 304N steel.

References

- [1] Brady, R. R., Proceedings, American Society for Testing and Materials, Vol. 59, 1959, pp. 774–785.
- [2] Simmons, W. F. and Van Echo, J. A., Report on the Elevated Temperature Properties of Stainless Steel, DS5-S1, American Society for Testing and Materials, 1965.

Elevated Temperature Properties of Nitrogen-Containing Type 304L Austenitic Stainless Steel

REFERENCE: Goodell, P. D. and Feeman, J. W., "Elevated Temperature **Properties of Nitrogen-Containing Type 304L Austenitic Stainless Steel**," *Elevated Temperature Properties as Influenced by Nitrogen Additions to Types 304 and 316 Austenitic Stainless Steels, ASTM STP 522*, American Society for Testing and Materials, 1973, pp. 46–59.

ABSTRACT: Experimental heats of Type 304L steel, compositionally balanced to be wholly austenitic, were prepared to determine the influence of nitrogen on elevated temperature properties of this steel. Nitrogen is shown to increase the 1200 F (649 C) and 1350 F (732 C) yield strength slightly and to increase markedly the creep and rupture strength at 1200 F (649 C). The effect of nitrogen on creep and rupture strengths at 1350 F (732 C) and 1500 F (816 C), however, is less pronounced. A beneficial effect of nitrogen on creep and rupture properties at 1200 F (649 C) is attainable from a considerable range of heat treatments and grain sizes.

KEY WORDS: nitrogen, austenitic stainless steel, creep properties, yield, high temperature, grain size, heat treatment, rupture strength

The elevated temperature properties of the nonstabilized 300 Series stainless steels can be affected by several major and minor element compositional factors. Among these, nitrogen has been considered by numerous investigators and, at temperatures of about 1200 F (649 C), has been reported to improve creep-rupture strength [1-12].³ The magnitude of this strengthening, however, varies for the different circumstances considered. When expressed, for example, using the 1000-h rupture strength at 1200 F

www.astm.org

¹Formerly University of Michigan where the work was performed, now with Paul D. Merica Research Laboratory, International Nickel Company, Inc., Sterling Forest, Suffern, N. Y. 10901.

² Deceased, former Professor of Chemical and Metallurgical Engineering, University of Michigan, Ann Arbor, Mich. 48104.

³ The italic numbers in brackets refer to the list of references appended to this paper.

(649 C) as a common basis for comparison, the estimated increment of strengthening per 0.01 percent nitrogen ranges from 250 to 1000 psi (1.72 to 6.89 MPa). Additionally, various reports indicate comparisons of less, equal, and greater strengthening due to nitrogen than carbon.

In a previous paper by the present authors and T. M. Cullen, nitrogen was considered along with several other elements influencing the creeprupture properties of Type 304 steel [10]. The magnitude of increase of the 1000-h rupture strength at 1200 F (649 C) was found to be 950 psi (6.55 MPa) per 0.01 percent nitrogen up to about 0.14 percent; this was determined independently of the influence of carbon which was approximately 25 percent less effective. The alloy examined was balanced to be wholly austenitic and, except for the low carbon content, was representative in base alloy chemistry of compositions used in the manufacture of seamless tubing for elevated temperature service.

The preceding results represent behavior particular to nitrogen in the absence of such other factors as austenite/ferrite balance, manganese level, and carbon effects. In light of this rather substantial strengthening influence, further work was undertaken to characterize more fully the behavior and to determine the manner in which nitrogen was operative [13]. The present paper considers the results indicating the influence of nitrogen on yield strength and minimum creep rate, as well as creep-rupture characteristics. Properties were evaluated at 1200 (649 C), 1350 (732 C), and 1500 F (816 C), and effects of grain size and temperature of heat treatment are included.

Experimental Procedure

The nominal composition (in weight percent) of the alloy base used in this investigation is as follows:

\mathbf{Cr}	Ni	Mn	Si	С
18.5	10.5	1.5	0.5	0.03

This analysis was selected to provide a wholly austenitic structure and is typical (except for carbon) of Type 304 steel as usually balanced for elevated temperature service for seamless tubing. Heats of this base composition were prepared in which the nitrogen content was systematically varied from 0.01 to 0.22 percent. These limits were dictated, respectively, by the residual nitrogen of the melt stock and by the solubility limit of nitrogen in the liquid alloy during melting. The actual compositions are given in Table 1.

The alloys were prepared as 10-lb, (4.54-kg) carbon deoxidized, vacuum induction melts which were poured into 5-lb (2.27-kg) ingot molds approximately 2 in. (51 mm) in diameter. The ingots were 5 in. in length after discard of the hot top. Virgin melt stock was used consisting of electrolytic

	Chemical Composition ^a (weight percent)										
Heat No	С	N	Mn	Ni	Cr	Si					
1310A	0.01	0.01 ^b	1.42	10.38	18.46	0.46					
1470A	0.02°	0.023	1.5^{c}	10.5°	18.5°	0.5^{c}					
1468A	0.02	0.037	1.5	10.70	18.40	0.44					
1470B	0.02	0.038	1.48	10.60	18.42	0.42					
1361A	0.03	0.051	1.56	10.95	18.06	0.5°					
1470C	0.02°	0.087	1.5°	10.5^{c}	18.5^{c}	0.5°					
1423A	0.03	0.115	1.72	10.60	18.60	0.46					
1410A	0.03	0.14	1.60	11.00	18.57	0.52					
1411A	0.03	0.14	1.5°	10.5°	18.5°	0.53					
1343A	0.03^{c}	0.145	1.5°	10.3°	18.3°	0.5^{c}					
1412A	0.03	0.17	1.66	10.85	18.37	0.51					
1343B	0.02	0.18	1.62	10.00	17.92	0.48					
1412B	0.03°	0.22	1.5°	10.5^{c}	18.5^{c}	0.5^{c}					

TABLE 1—Composition of the experimental heats.

 a In addition to the elements shown, the alloys contain residual amounts of phosphorus sulphur, and molybdenum which are typically less than 0.005, 0.007, and 0.03 percent, respectively.

^b No intentional addition.

^c Estimated values based on the aim analysis, comparison to other half of split heat, comparison to similar heats, and melting experience.

iron, nickel and manganese, aluminum-reduced chromium, and ferrosilicon. Nitrogen additions were made during the latter stages of melting by adding chromium-nitride powder or by way of pickup from a nitrogen atmosphere introduced into the furnace chamber or both.

The ingots were hot rolled at 2150 F (1177 C) to 1-in. (25-mm) square bars. After hot rolling and sampling for chemical analyses, the bars were cold rolled to 0.7-in. (18-mm) square bars, heat treated for $\frac{1}{2}$ h at 2050 F (1121 C) and water quenched. These bars were given a final cold reduction of 50 percent in cross-sectional area to produce $\frac{1}{2}$ -in. (12.7-mm) square bar. Specimen blanks were cut from this bar, given a final heat treatment, and water quenched; unless specified otherwise, this heat treatment was for $\frac{1}{2}$ h at 2050 F (1121 C) with water quench. Grain sizes following this treatment were between ASTM 5 and 4. These final procedures approximate the production of seamless tubing.

Creep-rupture tests were conducted using specimens having a 1-in. (25-mm) reduced section gage length of 0.250 in. (6.35 mm) diameter. Tests were of the constant load type and were conducted in simple beam or direct loaded individual test units. Elongation measurements were made using a modified Martens-type rotating mirror extensioneter affixed to the specimen shoulder. Strain was corrected to that of the reduced section gage length. Creep-rupture tests at 1200 F (649 C) were conducted at stress levels corresponding to rupture times ranging from about 50 to 2000 h. Additional, though less extensive, tests were carried out at 1350 F (732 C) and 1500 F (816 C). Stress-strain measurements on loading the creep tests were used to indicate the 0.2 percent offset yield strengths and work hardening trends. The minimum creep rate and time to rupture were the principal quantities determined from the creep-rupture tests; from these, 1000-h rupture strengths and 0.01 percent per hour creep strengths were derived for each alloy.

Results and Discussion

Yield Behavior

At 1200 F (649 C) the yield strength was found to be controlled by nitrogen content and grain size but was not independently influenced by the temperature of heat treatment. This is shown by the results in Fig. 1 which combines the loading results from numerous creep-rupture tests of several alloys. These data also indicate no discernible influence of nitrogen on the work hardening characteristics of this alloy; this fact was also confirmed by the stress dependence of the initial strain of the creep tests.

The magnitude of the influence of nitrogen on yield strength was similar at both 1200 F (649 C) and 1350 F (732 C), corresponding to an increase of 250 psi (1.72 MPa) per 0.01 percent nitrogen; the effect of nitrogen was considerably less at 1500 F (816 C). These effects are indicated by the results in Fig. 2. Recognizing slight variations in the grain sizes of various alloys, the yield strengths have been corrected to a common grain size. This correction was of the conventional Petch-Hall form, and the constants were determined from among the experimental data.

Creep and Rupture Strengths

Creep resistance at 1200 F (649 C) was markedly improved by increasing nitrogen content in the base alloy considered. Figure 3 illustrates the strengthening in terms of the stress to produce a minimum creep rate of 0.01 percent per hour.⁴ This creep strength was increased by 950 psi (6.55 MPa) for each 0.01 percent nitrogen up to about 0.14 percent nitrogen; at higher nitrogen levels the effect was less pronounced. In contrast to the behavior at 1200 F (649 C), the effect of nitrogen was considerably less at 1350 F (732 C) and 1500 F (816 C). The stress sensitivity of the minimum creep rate was also found to depend on nitrogen content. Values for a power

⁴ This value is within the measured range of data and does not require extrapolation. Examples of several typical creep curves, as well as minimum creep rate and rupture time data, will be presented in Figs. 6 and 7, in conjunction with heat treatment and grain size effects.



FIG. 1—Influence of nitrogen content and grain size on the 1200 F (649 C) stress-strain characteristics of several heats. These data are taken from loading of the creep-rupture tests. (To convert ksi to megapascals, multiply by 6.894757).

law stress exponent are included in Fig. 3. These data indicate a maximum stress sensitivity at between 0.09 and 0.14 percent nitrogen at 1200 F (649 C). A significant consequence of this maximum is that at lower creep rates a more pronounced inflection would be apparent at about 0.14 percent



FIG. 2—Influence of nitrogen on the 0.2 percent offset yield strength of the nitrogen series heats at 1200, 1350, and 1500 F (649, 732, and 816 C). (To convert ksi to megapascals, multiply by 6.894757).

nitrogen than is indicated in Fig. 3.⁵ Thus, optimum long time creep resistance for this particular alloy base would probably fall within this intermediate range of nitrogen content.

⁵ Stress sensitivity of the minimum creep rate was based on engineering stress and strain relative to the initial dimension of the test specimen and applied load which was constant. Expressed in terms of true instantaneous stress and strain rates, the absolute values for stress sensitivity were lower than when expressed in engineering terms, but also decreased as nitrogen was increased above 0.14 percent.



FIG. 3—Influence of nitrogen on the 0.01 percent per hour creep strength (based on minimum creep rate) of the nitrogen series heats at 1200, 1350, and 1500 F (649, 732, and 816 C). The numbers adjacent to the symbols are the stress exponent n for the relation $e = A\sigma^n$ at a minimum creep rate of 0.01 percent per hour. (To convert ksi to megapascals, multiply by 6.894757).

The influence of nitrogen on the 1000-h rupture strength appears, in Fig. 4, to be similar to that on the 0.01 percent per hour creep strength (Fig. 3). At 1200 F (649 C) the 1000-h rupture strength increased by 950 psi (6.55 MPa) for each 0.01 percent nitrogen up to about 0.14 percent; thereafter the effect of increasing nitrogen was much less. At 1350 F (732 C) and 1500 F

(816 C), the effect of nitrogen over the entire composition range was considerably less pronounced than at 1200 F (649 C). Smith et al [4] and Kawabe et al [11] have also reported a decrease in strengthening influence of nitrogen with increasing temperature. In each of these cases, however, the behavior was at least partially associated with interrelated effects of high carbon levels.



FIG. 4—Influence of nitrogen on the 1000-h rupture strength of the nitrogen series heats at 1200, 1350, and 1500 F (649, 732, and 816 C). The numbers adjacent to the symbols are the estimated elongation for rupture in 1000 h. (To convert ksi to megapascals, multiply by 6.894757).



FIG. 5—Examples illustrating the type of relationship between minimum creep rate and time to rupture of tests of the nitrogen series heats at 1200 F (649 C).

Figure 4 also indicates the percent elongation on rupture in 1000 h. These values exhibit a generally decreasing trend with increasing nitrogen content. Despite this trend, the elongation remains quite high within the regime of 1200 F (649 C) for which the strengthening influence of nitrogen is greatest; only one anomalously low value was noted at less than 0.14 percent nitrogen. At nitrogen levels above about 0.14 percent, elongation was generally lower.

Stress exponents at 1000 h for a power function relation between rupture life and stress were slightly less than those given for the minimum creep rate in Fig. 3, but they exhibited a similar maximum in the intermediate range of nitrogen contents. A direct comparison between minimum creep rate and rupture life is presented in Fig. 5, which is based on data over a wide range of stress and nitrogen levels. A power function correlation is indicated with creep rate proportional to rupture life to the -1.15 power for all material examined. The proportionality factor appears primarily dependent on nitrogen content and slightly dependent on grain size (see following section). Thus, Fig. 5 corroborates that creep and rupture processes under these conditions are closely related and similarly affected by nitrogen content.

Taken together, the characteristics illustrated in Figs. 3, 4, and 5 indicate the influence of nitrogen at 1200 F (649 C) to be characterized by two regimes of behavior. These are distinguished by strengthening magnitude, ductility and stress sensitivity of creep rate, and rupture life. The most advantageous combination of these characteristics pursuant to long time strength and ductility can be derived from the present results to be attained at about 0.14 percent nitrogen. Note, however, that alloys containing less than 0.10 percent manganese exhibited a lower strength and nitrogen level for this optimum combination of properties [10].⁶ Hence the influence of nitrogen can be affected to some extent by base alloy chemistry and warrants further consideration.

Heat Treatment and Grain Size

Tests were performed to determine trends resulting from variations in alloy heat treatment. Figure 6 illustrates aspects of both temperature and grain size as evident in an alloy of intermediate nitrogen content. These results indicate that at nearly constant grain size, decreasing the temperature of heat treatment reduced the rupture life and increased the minimum creep rate (curves B, C, D, and E). At a fixed temperature of heat treatment (1750 F (954 C) in Fig. 6, curves D and F), the minimum creep rate was unaffected as the grain size was reduced by about one half, whereas rupture elongation and life were extended. Figure 7 provides further illustration of the effect of heat treatment and indicates essentially similar behavior in alloys of both low and intermediate nitrogen content. Consequently, the effects of temperature of heat treatment and grain size are not viewed as being characteristically related to any influence of nitrogen but, rather, are associated with behavior of the base alloy.

In contrast, alloys subject to precipitation, such as of chromium carbide in Type 304 steel [10] or titanium carbide in Type 321 steel [14], exhibit much greater sensitivity to heat treatment conditions than do the nitrogen strengthened alloys illustrated in Figs. 5 and 7. The relative independence of the influence of nitrogen from effects of heat treatment appears consistent

⁶ Decreasing manganese would be expected to increase nitrogen activity, thus lowering solubility. This suggests, therefore, that the transition between the two regimes of behavior is in some way influenced by nitrogen activity. This and further relations of properties to nitrogen solubility will be considered in a subsequent publication.









with the reported high solubility of nitrogen [15, 16] at the heat treating temperatures employed. As a result, a strengthening influence of nitrogen may be attainable with a wide range of processing conditions.

Summary

The influence of nitrogen on elevated temperature mechanical properties has been examined in a wholly austenitic stainless steel alloy base of 18.5Cr-10.5Ni-1.5Mn-0.5Si-0.02C. The following conclusions are drawn:

1. The effect of nitrogen on creep and rupture properties at 1200 F (649 C) resulted in two regimes of behavior with the strengthening effect of nitrogen being quite pronounced at levels below 0.14 percent nitrogen. For this alloy base, optimum long time properties are expected at nitrogen levels of about this value, or slightly less.

2. Nitrogen has only a small strengthening effect on high temperature yield strength.

3. Nitrogen offers only limited benefit to creep and rupture properties at 1350 F (732 C) and 1500 F (816 C).

4. Low carbon, nitrogen-strengthened heats exhibited relative freedom from deleterious effects of heat treatment and grain size variation.

A cknowledgment

The authors gratefully acknowledge the support for this work provided by the Metallurgy and Piping Task Force of the Prime Movers Committee of the Edison Electric Institute and the Department of Chemical and Metallurgical Engineering of the University of Michigan.

References

- [1] Hum, J. K. Y. and Grant, N. J., Transactions, American Society for Metals, Vol. 45, 1953, pp. 105-133.
- [2] Dulis, E. J., Smith, G. V., and Houston, E. G., Transactions, American Society for Metals, Vol. 45, 1953, pp. 42-76.
- [3] Monkman, F. C., Price, P. E., and Grant, N. J., Transactions, American Society for Metals, Vol. 48, 1956, pp. 418-445.
- [4] Smith, G. V., Garofalo, F., and Zwell, L., "Influence of Carbon and Nitrogen on Creep and Creep-Rupture Properties and Structure of 18Cr-8Ni and 18Cr-8Ni-Mo Steels," lecture presented at the American Institute of Mining, Metallurgical, and Petroleum Engineers Symposium, The Relation of Structure to High Temperature Properties, Chicago, 4 Nov. 1957.
- [5] Brady, R. R., Transactions, American Society for Testing and Materials, Vol. 59, 1959, p. 774.
- [6] Smith, G. V., Garofalo, F., Whitmore, R. W., and Burt, R. R., Transactions, American Society of Mechanical Engineers, Vol. 81-D, 1959, p. 610.
- [7] Nakaoawa, R. and Otoouro, Y., Transactions, National Research Institute for Metals, Vol. 5, 1963, p. 129.
- [8] Gerlach, H., "The Influence of Carbon, Nitrogen and Boron on the Precipitation Behavior of an Austenitic Steel Containing 16% Chromium, 16% Nickel and 2% Molybdenum With and Without Additions of Niobium," thesis to Rheinisch-Westfalische Technische Hochschule Aachen, 1966.
- [9] Mercier, A., Leveque, and Remy, G., Revue de Metallurgie, Vol. 64, 1967, p. 1085.

- [10] Goodell, P. D., Cullen, T. M., and Freeman, J. W., Transactions, American Society of Mechanical Engineers, Vol. 89, Series D, 1967, p. 517.
- [11] Kawabe, Y., Nakagawa, R., and Mukoyama, T., Transactions, Iron and Steel Institute of Japan, Vol. 8, 1968, p. 353.
- [12] Yoshikuni, K., Nyuichi, N., and Tamotsu, M., "Strengthening Effects of Boron, Nitrogen and Molybdenum Additions on the High-Temperature Properties of 18Cr-12Ni-0.2C Austenitic Heat-Resistant Steel," Tetsu to Hagane, Vol. 54, 1968, p. 473.
- [13] Goodell, P. D., "The Influence of Nitrogen on the Elevated Temperature Properties of an Austenitic Stainless Steel," thesis to the Department of Chemical and Metallurgical Engineering, University of Michigan, Ann Arbor, 1971, University Microfilms Inc., Ann Arbor, Mich., No. 72-14878.
- [14] White, J. E. and Freeman, J. W., Transactions, American Society of Mechanical Engineers, Vol. 85, Series A, 1963, p. 108.
- [15] Eckel, J. F. and Cox, J. B., Journal of Materials, Vol. 3, 1968, p. 604.
- [16] Masumoto, T. and Imai, Y., Journal of the Institute of Metals, Japan, Vol. 33, 1969, p. 1364.

Influence of Nitrogen on the Creep-Rupture Properties of Type 316 Steel

REFERENCE: Cullen, T. M. and Davis, M. W., "Influence of Nitrogen on the Creep-Rupture Properties of Type 316 Steel," Elevated Temperature Properties as Influenced by Nitrogen Additions to Types 304 and 316 Austenitic Stainless Steels, ASTM STP 522, American Society for Testing and Materials, 1973, pp. 60–78.

ABSTRACT: An experimental program has been conducted to determine the influence of nitrogen on the creep and rupture properties of Type 316 stainless steel. A series of commercial and laboratory heats with similar base composition but varying nitrogen contents were utilized in the study. The nitrogen content of the heats ranged from 0.039 to 0.15 percent.

The creep and rupture strengths of the steel at 1200 F increased with increasing nitrogen level. The stress for rupture in 1000 and 10 000 h was approximately 40 percent higher for the 0.15 percent nitrogen heat than for the 0.039 percent nitrogen alloy. The stress to produce a minimum creep rate of 0.01 percent per 1000 h increased by 50 percent over the same nitrogen range.

A ferrite containing heat was also tested in the program. Ferrite was shown to decrease creep resistance and increase rupture ductility. No significant influence of this phase was noted with respect to rupture strength.

Various parameter extrapolation methods were compared in their ability to predict the location of a break in the rupture curve of one of the nitrogen containing heats. Long time rupture tests were conducted to pinpoint the exact position of this change in slope.

KEY WORDS: nitrogen, creep strength, austenitic stainless steels, creep rupture strength, ferrite, microstructure, extrapolations, creep rate

As part of an effort to determine the influence of residual elements on the strength characteristics of the austenitic steels, a detailed evaluation has been made of the variation of creep-rupture properties of wrought Type 316 stainless steel with nitrogen content. Both laboratory and commercially produced materials were used in this study. Heats were selected on the

¹Section manager and supervisor, respectively, Materials Evaluation, Metallurgical Research, and Development Dept., C-E Combustion Div., Combustion Engineering, Inc., Chattanooga, Tenn. 37401.

basis of similarity of chemical composition, mechanical processing, and thermal history. The nitrogen content of the five heats used in this investigation ranged from 0.039 to 0.15 percent.

This investigation received added impetus from an American Society of Mechanical Engineers (ASME) Boiler and Pressure Vessel Code Case to permit higher allowable stresses for nitrogen-containing grades of Types 304 and 316 steel at temperatures up to 1000 F (538 C). This case was based on improved tensile properties in steels which contain 0.10 to 0.16 percent nitrogen.

One heat of a duplex austenitic ferritic Type 316 steel was also tested in the research program. This material contained approximately 16 percent ferrite by volume. The objective of this phase of the study was to determine the influence of ferrite on long time elevated temperature strength and ductility.

Experimental Material

The alloys used in this research program, with one exception, consisted of sections of seamless superheater tubing made from centrifugally cast and tube reduced hollows. The exception was the ferrite-containing material, Heat F, which was air induction melted in the laboratory and statically cast. The resultant ingot was hot forged from 2100 F (1149 C) to a $\frac{1}{2}$ -in. (0.0127-m) thick slab.

The chemical compositions (in percentages by weight) of the test materials are given in Table 1. With the exception of the ferrite-containing material, the base compositions of all the materials were similar. In fact, nitrogen was the only significant variable among the first five heats listed.

All alloys evaluated in this study were heat treated in the same manner; 15 min at 1975 F (1079 C), air cooled. Following heat treatment, specimen blanks were cut from the test samples. Standard 0.252-in. (6.40-mm) uniaxial tension specimens were machined from these coupons and used for both the tensile and the creep-rupture tests.

Heat	С	Mn	Si	Cr	Ni	Mo	Ν
A	0.042	1.55	0.50	16.25	12.55	2.62	0.039
В	0.041	1.18	0.39	16.00	11.80	2.28	0.044
С	0.042	1.47	0.50	16.20	12.05	2.43	0.070
D	0.058	1.50	0.47	16.75	12.95	${f 2}$. 55	0.099
\mathbf{E}	0.041	1.55	0.36	16.71	14.25	2.58	0.15
\mathbf{F}	0.040	1.56	0.42	19.86	8.24	2.90	0.11

TABLE 1-Chemical composition of experimental materials.

Test Temperature, deg F (deg C)	Ultimate Strength, ksi (MPa)	0.2% Offset Yield Strength, ksi (MPa)	Elongation in 1 in., %, (25 mm)	Reduction of Area, %
	Heat A, 0.0	39 percent nitrogen		
\mathbf{RT}^{a}	85.0 (586)	32.4(223)	62	78
200 (93)	76.5 (527)	26.9 (185)	52	79
400 (204)	69.0 (476)	31.2 (214)	45	73
600 (316)	67.4(465)	18.0 (124)	42	73
800 (427)	66.4 (458)	16.8 (116)	42	69
1000 (538)	64.3 (443)	13.8 (95.1)	45	69
1200 (649)	52.4 (361)	14.7 (101)	48	70
	Heat B, 0.0	44 percent nitrogen		
1200 (649)	50.0 (348)	15.6 (108)	44	72
	Heat C 0.02	70 percent nitrogen		
рŢ	86 0 (502)	27 9 (956)	63	78
200 (02)	76 9 (595)	37.2 (200) 38 8 (100)	50	10
200 (93) 400 (904)	70.3 (320)	20.0 (199)	47	75
600 (204)	68 8 (474)	18.0 (100)	44	73
800 (310)	67 8 (467)	16.0 (124) 16.8 (116)	44	70
1000 (538)	65 2 (450)	15.0 (103)	43	69
1200 (649)	54.6 (376)	13.8 (95.1)	43	70
. ,	Heat D, 0.0	99 percent nitrogen		
	,			
\mathbf{RT}	91.8 (633)	41.6 (287)	62	78
\mathbf{RT}	94.0(648)	37.2(256)	57	78
200 (93)	83.6 (576)	31.8 (219)	54	78
400 (204)	77.6 (535)	26.4(182)	49	75
600 (316)	76.4 (527)	21.6(149)	47	72
800 (427)	75.2 (518)	20.8 (143)	43	68
1000 (538)	71.0 (490)	18.0(124)	45	69
1200 (649)	58.8 (405)	16.8 (116)	47	67
	Heat E , 0.1	15 percent nitrogen		
\mathbf{RT}	92.2 (636)	40.8 (281)	52	70
200 (93)	83.0 (512)	33.0(228)	48	75
400(204)	76.4(527)	27.6 (190)	48	75
600 (316)	74.0 (510)	22.8(157)	43	69
800 (427)	72.6 (501)	20.4 (141)	47	69
1000 (538)	68.0 (469)	19.0 (131)	44	66
1200 (649)	58.4 (403)	18.8 (130)	48	65
	Heat F, I	16 percent ferrite		
рт	104 4 (720)	50 0 (345)	50	73
1200(649)	55.6 (383)	27.6 (190)	50	70

 TABLE 2—Tensile properties of experimental materials.

^a RT = room temperature.

Experimental Results

Varying Nitrogen Series

The results of the tension tests conducted on the various experimental alloys are shown in Table 2. Four of the five alloys which made up the nitrogen series were tested, at 200 F (111 C) increments, over a range from room temperature up to 1200 F (649 C). The results of these tests are plotted in Fig. 1. In general, both yield strength and ultimate strength at each temperature increment increased as nitrogen content increased. The differences, however, were not as great as might have been expected.

The creep-rupture test data obtained from these materials are shown in Table 3. An extensive test program was conducted on the five alloys. In each case the alloy was tested at 1200 F (649 C), and the rupture curve established out to at least 2500 h. In addition, Heat C was tested at 1300,



FIG. 1—Tensile and yield strengths of varying nitrogen alloys as a function of test temperature.

Stress, ksi (MPa)	Temperature, deg F (deg C)	Rupture, Time, h	Elonga- tion, %	Reduction of Area, %	Minimum- Creep Rate, %/1000 h
	Heat.	A, 0.039 perces	nt nitrogen		
$\begin{array}{ccc} 30 & (207) \\ 28 & (193) \\ 25 & (172) \\ 23 & (159) \\ 21 & (145) \end{array}$	$\begin{array}{c} 1200 & (649) \\ 1200 & (649) \\ 1200 & (649) \\ 1200 & (649) \\ 1200 & (649) \\ 1200 & (649) \end{array}$	$\begin{array}{r} 68.4 \\ 163.9 \\ 450.8 \\ 994.5 \\ 3309.2 \end{array}$	41 43 55 72 81	39 43 54 60 69	128 50 30 9 8
20 (138)	1200 (649)	4394.2	70	72	6.6
	Heat	B, 0.044 percen	nt nitrogen		
$\begin{array}{ccc} 30 & (207) \\ 28 & (193) \\ 24 & (165) \\ 22 & (152) \\ 20 & (138) \end{array}$	1200 (649) 1200 (649) 1200 (649) 1200 (649) 1200 (649) 1200 (649)	32.5 67.7 250.3 753.2 2575.0	32 68 37 54 47	36 39 39 48 51	620 308 108 32 9.6
$\begin{array}{cccc} 30 & (207) \\ 28 & (193) \\ 25 & (172) \\ 22 & (152) \\ 20 & (138) \\ 18.5 & (128) \\ 17 & (117) \end{array}$	1200 (649) 1200 (649) 1200 (649) 1200 (649) 1200 (649) 1200 (649) 1200 (649)	74.6 213.0 656.2 3476.1 6825.3 10076.5 15790.8	26 24 32 51 74 64 59	32 28 41 54 64 71 70	209 59 22 7.2 2.3 1.06 0.189
$\begin{array}{cccc} 25 & (172) \\ 22 & (152) \\ 20 & (138) \\ 19 & (131) \\ 18 & (124) \\ 17 & (117) \\ 16.5 & (114) \\ 16 & (110) \\ 15 & (103) \\ 13.8 & (95.1) \\ 13 & (89.6) \\ 12 & (82.7) \\ 11 & (75.8) \end{array}$	1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704) 1300 (704)	$\begin{array}{c} 36.5\\ 104.1\\ 228.1\\ 258.1\\ 319.0\\ 377.5\\ 785.3\\ 753.7\\ 1232.5\\ 1854.6\\ 2421.0\\ 4078.3\\ 6258.1 \end{array}$	50 85 60 74 70 97 91 100 112 81 110 86 90	53 66 60 63 68 72 76 75 76 74 71 80 75	$1280 \\ 363 \\ 136 \\ 152 \\ 100 \\ 95 \\ 51 \\ 60 \\ 41 \\ 22 \\ 19 \\ 10.8 \\ 8.54$
$\begin{array}{cccc} 25 & (172) \\ 15 & (103) \\ 12.5 & (86.2) \\ 10 & (68.9) \\ 15 & (103) \end{array}$	1400 (760) 1400 (760) 1400 (760) 1400 (760) 1450 (788)	$\begin{array}{r} 2.7 \\ 83.3 \\ 251.2 \\ 921.0 \\ 27.9 \end{array}$	75 90 105 115 108	60 76 75 89 71	563 236 66
$\begin{array}{cccc} 12.5 & (86.2) \\ 16.4 & (113) \\ 12.5 & (86.2) \\ 10 & (68.9) \\ 9 & (62.1) \\ 7 & (48.3) \\ 5 & (34.5) \end{array}$	$\begin{array}{ccc} 1450 & (788) \\ 1500 & (816) \\ 1500 & (816) \\ 1500 & (816) \\ 1500 & (816) \\ 1500 & (816) \\ 1500 & (816) \end{array}$	$75.2 \\ 5.0 \\ 40.6 \\ 87.9 \\ 170.9 \\ 614.9 \\ 2892.3$	97 111 126 97 101 89 70	72 78 75 73 73 66 42	390 562 398 87 16

TABLE 3—Cree	n-runture i	nronerties i	of test	materials
	p-, apprais p		<i>oj 1</i> 001	manerano

S ksi	Stress, (MPa)	Temperature, deg F (deg C)	Rupture Time, h	Elonga- tion, %	Reduction of Area, %	Minimum- Creep Rate %/1000 h
		Heat C, 0.07	'0 percent nitre	ogen—Contir	nued.	
25	(172)	1175 (635)	3142.9	34	47	5.1
25	(172)	1225 (663)	290.9	39	39	85
45	(172)	1250 (677)	186.5	47	46	205
25	(172)	1275 (691)	81.5	40	42	
25	(172)	1325 (718)	21.5	80	55	
25	(172)	1350 (732)	9.9	54	50	
		Heat.	D, 0.099 perce	ent nitrogen		
36	(248)	1200 (649)	71.1	13	15	
35	(241)	1200 (649)	103.0	14	14	67
32	(221)	1200 (649)	248.8	15	15	25
31	(214)	1200 (649)	400.4	16	20	14.5
30	(207)	1200 (019) 1200 (649)	935.3	20	30	7 6
29	(200)	1200 (649)	1162 0	30	34	83
20 90	(103)	1200 (049)	1565 0	40	43	7.0
40 96	(170)	1200 (049)	2067 5	49	51	1.0
$20 \\ 22.5$	(115)	1200 (049) 1200 (649)	7427.0	75	64	1.2
28	(193)	1300 (704)	43.1	33	37	380
25	(172)	1300 (704)	227.0	58	52	80
22	(152)	1300 (704)	517.9	67	59	40
20	(138)	1300 (704)	865.2	75	63	25
18	(124)	1300 (704)	1390.8	67	58	19
16	(110)	1300 (704)	2343 0	62	56	9 55
15	(103)	1300 (704)	3532.6	64	5 7	8.55
		Heat	E, 0.15 percer	ıt nitrogen		
40	(276)	1200 (649)	49.5	20	22	
35	(241)	1200(649)	148.8	16	17	42
32.F	5 (224)	1200 (649)	532.8	18	17	10.6
30	(207)	1200 (649)	1091.3	17	18	6 14
28	(193)	1200 (619)	3210.9	39	40	3.40
		He	at F, 16 percer	nt ferrite		
35	(241)	1200 (649)	106.1	65	54	
30	(207)	1200 (649)	370.7	65	68	84
27	(186)	1200 (649)	1004.8	80	67	33
25	(172)	1200 (649)	1485 0	68	70	21
23	(159)	1200 (649)	3081.2	92	70	10.2
	(145)	1200 (640)	5738 0	04	73	5 43

TABLE 3—Continued.



FIG. 2-Stress-rupture time behavior of 0.039 percent nitrogen material.



FIG. 3—Stress-rupture time properties of 0.044 percent nitrogen alloy.



FIG. 4-Stress-rupture time properties of 0.07 percent nitrogen alloy.


FIG. 5-Stress-rupture time properties of 0.099 percent nitrogen material.



FIG. 6-Stress-rupture time properties of 0.15 percent nitrogen alloy.



FIG. 7—Comparison of 1200 F (649 C) rupture properties of varying nitrogen alloys.



FIG. 8—Variation of 1000 and 10 000 h rupture strength at 1200 F (649 C) with nitrogen content.

1400, and 1500 F (732, 760, and 816 C) and Heat D at 1300 F (704 C). Test durations in excess of 15 000 h were recorded in this phase of the experimental program.

The rupture curves obtained from individual alloys are shown in Figs. 2 through 6. Straight lines seem to fit the data quite well with the exception of Heats C and D which show distinct breaks in their rupture curves at 1200 and 1300 F (649 and 704 C). Breaks of this type are generally manifestations of structural instability and, in this case, it is thought that the downward breaks were associated with a precipitation reaction occurring in the material.

The 1200 F (649 C) rupture results for the five alloys with varying nitrogen contents are given in Fig. 7. This comparison clearly shows the beneficial effect of nitrogen on the rupture properties of Type 316 stainless steel. The 1000 and 10 000 h rupture strengths of these materials are plotted as a function of nitrogen content in Fig. 8. Again, the strong dependence of rupture strength on nitrogen content is clearly evident.

Due to the presence of sharp downward breaks in the rupture curves of the 0.070 and 0.099 percent nitrogen alloys, no attempt has been made to relate 100 000 h rupture strength to nitrogen level. This is because similar breaks may also be characteristic of the rupture curves of the other alloys and may occur, at times, beyond the duration of the longest test. If such breaks did exist, then straight line extrapolations of the short time data could lead to large errors in the assessment of long time strength.

The minimum-creep rates determined for each of the tests at 1200 F (649 C) are plotted in Fig. 9. These data again show the beneficial influence



FIG. 9—Comparison of minimum-creep behavior at 1200 F (649 C) of varying nitrogen alloys.



FIG. 10—Variation of stress to produce a minimum-creep rate of 0.01 percent per 1000 h with nitrogen content.

of nitrogen on creep strength. At any given stress, the higher the nitrogen level the lower the minimum-creep rate. These results are in good agreement with the rupture test results shown in Fig. 7.

The variation of stress required to produce a minimum-creep rate of 0.01 percent per 1000 h as a function of nitrogen content is shown in Fig. 10. The values shown are extrapolations of actual data and are intended only to show the trend of the data. The stress required to produce a minimum-creep rate of 0.01 percent per 1000 h is a material property considered in the setting of ASME Boiler Code allowable stresses and, as such, is a property commonly used to compare high temperature behavior.

Microstructural Studies

Figure 11 depicts the microstructures of several rupture specimens from Heat C. These photomicrographs show an increasing amount of matrix precipitate as testing time increases. This precipitate is thought to be a carbide phase, probably $M_{23}C_6$. Particles of sigma phase were also noted





FIG. 11—Microstructures of ruptured specimens from Heat C (0.07 percent nitrogen). Etched in 10 percent oxalic acid ($\times 500$).



FIG. 12—Microstructures of specimens with different nitrogen contents having approximately equivalent rupture times at 1200 F (649 C). Etched in 10 percent oxalic acid ($\times 1000$).

in the microstructure of this material. These particles, however, were neither very large nor very numerous even in the specimen which ruptured in 15 791 h.

The microstructures of four specimens from different heats which exhibited approximately equivalent rupture times are shown in Fig. 12. The amount of matrix precipitate was about the same for each heat. The small variance in quantity of precipitate which did exist is believed to be more related to the slightly higher carbon content of Heat D than to any other variable.

Duplex Material—Heat F

The influence of ferrite on high temperature behavior is not well understood. Consequently, Heat F (containing 16 percent ferrite) was subjected to the same type of evaluation as the other alloys in order to measure its effect on high temperature strength and ductility.



FIG. 13—Stress-rupture time properties of ferrite containing material compared to those of 0.099 percent nitrogen alloy.

The rupture results obtained from this material at 1200 F (649 C) are plotted in Fig. 13. For comparison purposes, a line representing the rupture data obtained from Heat D is also plotted in this graph. Heats D and F have similar carbon and nitrogen levels but vary greatly in chromium and nickel. These latter two elements were intentionally modified in order to stabilize ferrite in the microstructure. As can be seen, the rupture properties of Heat F are only slightly lower than those recorded from Heat D.

The creep properties of these two heats are plotted in Fig. 14. In this case, the creep resistance of Heat F is considerably less than that of Heat D. Evidently, the presence of the ferrite drastically reduced the creep resistance of the material.

Examination of the rupture ductilities recorded in Table 3 shows the pronounced influence of ferrite on ductility. The elongations recorded from the fractured specimens of Heat F were approximately four times those of



FIG. 14—Minimum-creep rate behavior of ferrite containing material compared to that of 0.99 percent nitrogen alloy.



FIG. 15—Ferrite content of Heat F as a function of aging time.

Heat D for the corresponding shorter test times at 1200 F (649 C). This is probably due to the transformation of ferrite to sigma phase during the course of the test.

Figure 15 illustrates how percent of ferrite varies with aging time at 1200 and 1300 F (649 and 704 C), while the microstructure of these specimens are shown in Fig. 16. The ferrite level of the alloy was measured with a Magne-Gage. Since sigma phase is noted for its beneficial effects on elevated temperature rupture ductility, it is thought that most of the ferrite transformed to sigma phase during the high temperature exposures.

The transformation of the ferrite during exposure was associated with a reduction in creep resistance (see Fig. 14). The improved rupture ductility of the material, however, almost negated the accompanying lowering of rupture strength. In other words, equivalent rupture times were obtained in alloys of significantly different creep resistance due to higher ductility of the lower creep strength alloy. The increased rupture ductility prolonged the tertiary creep and thereby extended life. For this reason, the ferrite did not have any significant effect on rupture strength of this material at 1200 F (649 C).

Parameter Extrapolation

One of the most interesting phases of this study was the effort to predict the position (time) of the break in the 1200 F (649 C) rupture curve of Heat C. The variation in slope between the rupture curves drawn through the short time test results at 1200 and 1300 F (649 and 704 C) made it







FIG. 17—Correlation of rupture data from 0.07 percent nitrogen alloy using Dorn parameter.

apparent that the 1200 F (649 C) rupture curve had to break sharply downward at some time beyond 3400 h. In order to predict the position of this break, the rupture data from this alloy were analyzed using the Larson-Miller [1],² the Dorn-Sherby [2], and the Manson-Haferd [3] timetemperature parameters. In addition, the portion of the linear rupture curve through the short time data was extrapolated as a straight line to define an outer limit to the estimate of rupture life. Next, three stresses were selected at which to conduct long time 1200 F (649 C) rupture tests. It was expected that one or more of these tests would fall beyond the break, and thus enable its position to be pinpointed. The stresses selected for these tests were 20, 18.5, and 17 ksi (138, 128, and 117 MPa).

Dorn Parameter—This parameter is based on the Arrhenius rate equation and has the following form:

$$P = t_r(e^{-\Delta H/RT})$$

where

P =parameter value,

 t_r = rupture time, h,

 ΔH = activation energy for creep,

² The italic numbers in brackets refer to the list of references appended to this paper.

R = gas constant, and

T = temperature, deg K. The value of the activation energy is determined from the slope of the

curve of rupture time versus reciprocal absolute temperature for tests conducted at constant stress. The slope of this curve equals $\Delta H/R$. The results of the tests conducted at 25 ksi (172 MPa) on Heat C fell on a curve with a slope equal to 5×10^4 . From this value, an activation energy (ΔH) of 99 000 cal/g mole was determined. Figure 17 is a graph of the rupture data from Heat C utilizing the Dorn parameter for correlation.

Larson-Miller Parameter—This parameter is perhaps more widely used than any other because of its simplicity. The Larson-Miller parameter has the following form:

$$P = T(\log t_r + C)$$

where

P =parameter value, T =temperature, deg R,



FIG. 18—Correlation of rupture data from 0.07 percent nitrogen alloy using Larson-Miller parameter.



FIG. 19—Correlation of rupture data from 0.07 percent nitrogen alloy using Manson-Haferd parameter.

 $t_r =$ rupture time, h, and

C = constant, usually assumed to be equal to 20.

The value of the constant can be determined by plotting log rupture time at constant stress versus 1/T and extrapolating the resultant curve to 1/T = 0. The data obtained were programmed into a computer for determination of the value of the parameter constant. A value of C = 20.4 was obtained, which agreed remarkably well with the normally assumed value of C = 20.

Figure 18 depicts a Larson-Miller correlation of the rupture data from Heat C. In this correlation a value of C = 20 was used.

Manson-Haferd Parameter-This parameter assumes the following form:

$$P = \frac{\log t_r - \log t_a}{T - Ta}$$

where

P =parameter value, $t_r =$ rupture time, h,

		Time to Rupture, h	
Extrapolation Method	20 ksi (138 MPa)	18.5 ksi (128 MPa)	17 ksi (117 MPa)
Larson-Miller	4500	6200	11000
Dorn-Sherby	7200	10000	16000
Manson-Haferd	2800	5800	7800
Straight line	8800	21500	58000
Actual test result	6825.3	10076.5	15790.8

TABLE 4—Summary of parameter testing, 1200 F (649 C).

 t_a = material constant, equals 12.4,

T = absolute temperature, deg R, and

Ta =material constant, equals 900.

The values of the two constants can be determined by plotting two or more sets of constant stress data on a graph of log rupture time versus absolute temperature. The coordinates of the point of intersection of the lines are log t_a and Ta. (See Fig. 19 for the Manson-Haferd correlation of the data.)

Evaluation of Parameter Fit—The three parameters, along with a linear extrapolation of the 1200 F (649 C) log stress-log rupture time curve, were used to estimate time to rupture at 1200 F (649 C) under stresses of 20, 18.5, and 17 ksi (138, 128, and 117 MPa) (see Table 4). Evaluation of these results makes it quite obvious that the Dorn-Sherby parameter came closest by far in predicting actual rupture test results.

Conclusions

1. Nitrogen significantly improves rupture strength in Type 316 steel at 1200 F (649 C) for time periods up to 10 000 h.

2. Whether or not nitrogen is beneficial to the $100\ 000\ h$ rupture strength of the alloy at $1200\ F\ (649\ C)$ is not known due to the presence of breaks in the rupture curves of some of the alloys. The other alloys may or may not exhibit similar behavior in longer time tests.

3. Ferrite lowers the creep resistance of Type 316 steel at 1200 F (649 C), possibly through a mechanism involving its transformation to the sigma phase.

4. Any decrease in rupture strength due to the presence of ferrite is masked by its beneficial influence on rupture ductility.

References

- [1] Miller, J. and Larson, F. R., *Transactions*, The American Society of Mechanical Engineers, Vol. 74, 1952.
- [2] Orr, R. L., Sherby, O. D., and Dorn, J. E., Transactions, American Society for Metals, Vol. 46, 1954, pp. 113–128.
- [3] Manson, S. S. and Haferd, A. M., "A Linear Time-Temperature Relation for Extrapolation of Creep and Stress Rupture Data," NASA-TN-2890, National Aeronautics and Space Administration, March 1953.

A Nitrogen Grade of Types 304 and 316 Austenitic Stainless Steels; Specification and Code Considerations

REFERENCE: Rohrig, I. A., "A Nitrogen Grade of Types 304 and 316 Austenitic Stainless Steels; Specification and Code Considerations," Elevated Temperature Properties as Influenced by Nitrogen Additions to Types 304 and 316 Austenitic Stainless Steels, ASTM STP 522, American Society for Testing and Materials, 1973, pp. 79–85.

ABSTRACT: In the past twenty years it has been observed by test that there has been an increase in the high temperature strength of the austenitic stainless steel Types 304, 316, 321, and 347 with the result that the ASME Boiler and Pressure Vessel Committee has permitted two increases in allowable stress values since 1952 based on additional data. Creep and stress rupture studies of these materials led to the conclusion that nitrogen in controlled additions had a strengthening effect. Nitrogen grades of Types 304 and 316 referred to as 304N and 316N are permitted for use through ASME Boiler and Pressure Vessel Committee Code Case 1423 originally issued in September 1969 (now Case 1423-2), and efforts are underway to include 304N and 316N in ASTM specifications. Three public utilities have installed 304N for high temperature steam piping, and additional installations are expected.

KEY WORDS: nitrogen, mechanical properties, austenitic stainless steels, piping, pressure vessels, boilers, tubing

Approximately 15 years ago, there was considerable activity in the studies being made of Types 321 and 347 austenitic stainless steels, due to the premature failures that had occurred in 321 superheater tubing and in heavy wall 347 high temperature steam piping adjacent to welded joints. The studies of these two alloys were carried out under the auspices of the Joint Committee on the Effect of Temperature on the Properties of Metals (Joint Committee of American Society for Testing and Materials (ASTM), American Society of Mechanical Engineers (ASME), and Metal Properties

¹ Materials engineer, Detroit Edison Co., Detroit, Mich. 48226.

Council (MPC)), and were designated as SP-5 (concerned with Type 347) and SP-6 (dealt with Type 321).

The effect of nitrogen in improving the high temperature strength of Type 304 was observed and mentioned by White and Freeman $[1]^2$ during the course of the SP-6 study. In 1962 the Edison Electric Institute contracted with the University of Michigan for further study of mechanisms for high temperature strengthening of Type 304. The underlying reason for this was to prevent the occurrence of low-strength heats of material getting into service in superheaters.

The correlation of the effects of carbon, manganese, nitrogen, and high temperature heat treatment in improving the high temperature strength of Type 304 was verified by Goodell, Cullen, and Freeman [2]. The investigators concluded that the 304 made today is consistently stronger at high temperature than 304 manufactured two decades ago, and that such improvement appears to be due to the controlled addition of nitrogen.

A further objective of this study was to obtain sufficient information for the writing of a specification, or a specified composition, which would assure a high level of strength in Type 304 austenitic stainless steel. The final result of the study appeared in April 1967 [2].

Prior to the presentation of the final report, the Metallurgy and Piping Task Force of the Prime Movers Committee of the Edison Electric Institute proposed the inclusion of a nitrogen grade of Type 304 into ASTM specifications for tubular products; namely, A213, A249, A312, A376, and A430. This was done at the 1967 Annual Meeting of ASTM, in Boston, and was presented before Subcommittee X of A-1 at its meeting of 27 June 1967. It was proposed that the new grade, to be called 304N, would have the following requirements:

Carbon	0.04 to 0.10
Nitrogen	0.07 to 0.20
Manganese	1.0 min
Heat treatment at	1800 F (982 C) min, solution treatment

No change was proposed for the room temperature physical properties which would remain at 75 000 psi (517 MPa) tensile strength, 30 000 psi (207 MPa) yield strength, and no change in the allowable stresses was requested. The allowable stress for Type 304H had been increased from 3600 psi (24.8 MPa) to 4500 psi (31.0 MPa) at 1200 F (649 C) in 1952 and to 5500 psi (37.9 MPa) in 1965, a total increase of 35 percent with no change in specified composition. It was believed that the nitrogen grade would ensure the user that he could rely on the material having the 5500 psi (37.9 MPa) later increased to 6050 psi (42.8 MPa) allowable strength or creep resistance at 1200 F (649 C). The increase in S-values made in 1965

² The italic numbers in brackets refer to the list of references appended to this paper.

	_		Meta	l Temperat	ures	
Material	Year	1000 F (538 C)	1050 F (566 C)	1100 F (593 C)	1150 F (621 C)	1200 F (649 C)
304 304 304H 304 304H 304N 316 316 316H 316H	1949 1952 ^b 1965 ^c 1969 ^d 1969 ^d 1969 ^e 1949 1952 ^b 1965 ^c 1969 ^d	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	8 000 10 000 11 100 12 150 12 150 12 400 9 000 12 200 13 300 14 500	6 000 7 500 8 700 9 750 9 750 9 750 9 750 7 000 10 400 12 000 12 400	4 600 5 750 6 900 7 700 7 700 7 700 5 300 8 500 9 500 9 800 9 800	$\begin{array}{cccccccccccccccccccccccccccccccccccc$
316N	1969¢	17 400	15 800	12 400 12 400	9 800 9 800	7 400
321 321 321H 321 321 321H	1949 1952^{b} 1965^{c} 1969^{d} 1969^{d}	10 000 13 500 14 500 13 800 14 000	8 000 13 100 13 700 9 600 11 700	6 000 10 300 10 500 6 900 9 050	4 600 7 600 7 700 5 000 6 900	$\begin{array}{cccc} 3 & 600 \\ 5 & 000 \\ 5 & 500 \\ 3 & 600 \\ 5 & 350 \end{array}$
347 347 347H 347 347 347H	$1949 \\ 1952^{b} \\ 1965^{c} \\ 1969^{d} \\ 1969^{d} \\ 1969^{d}$	$\begin{array}{cccc} 10 & 000 \\ 13 & 500 \\ 14 & 500 \\ 14 & 000 \\ 14 & 400 \end{array}$	8 000 13 100 13 900 12 100 14 100	6 000 10 300 12 000 9 100 13 000	$\begin{array}{ccc} 4 & 600 \\ 7 & 600 \\ 8 & 500 \\ 6 & 100 \\ 10 & 500 \end{array}$	36005000600044007900

TABLE 1-Maximum allowable stress, psi.ª

^a To convert psi to megapascals, multiply by .006894757.

^b Increase due to change in the ASME Boiler Code Stress Criteria.

^c Revised stresses approved by Main Committee following review of data, published June 1965. (ASME Boiler and Pressure Vessel Committee.) The 1965 values are the upper stresses only, of the H grades. Table PG-23.1, Section 1 Power Boilers, ASME Code 1965.

^d ASME Boiler and Pressure Vessel Code, Winter 1969 Addenda. Table PG-23.1; the values are the upper stresses only for both the standard and the H grades.

• ASME Boiler and Pressure Vessel Code Case 1423-1, October 1969. The values shown are the upper stresses only for the nitrogen-bearing grades, 304N and 316N.

and a further increase in 1969 are shown in Table 1 which also shows, for comparison, the allowable stress values for Types 304, 316, 321, and 347. The upper stress values only are shown for the dual stresses permitted for these materials. For detailed tabulations regarding published allowable stresses, the reader is referred to the appropriate tables in Section I, Section III, and Section VIII of the ASME Boiler and Pressure Vessel Code $\lceil 3 \rceil$.

The proposal for Type 304N grade was accepted at the June 1967 Meeting of Subcommittee X of ASTM Committee A-1, but when sent out for letter ballot received negative votes. The negative voters questioned the need for 304N, considering that 304H existed. They also questioned the need for nitrogen in the range of 0.07 to 0.20, because the opposition stated that their product was equally strong at 1200 F (649 C) with nitrogen consistently below 0.07 percent.

In October 1968 the Westinghouse Corporation formally proposed a Case to the ASME Boiler and Pressure Vessel Committee for use of Types 304 and 316 modified with controlled nitrogen additions of 0.10 to 0.16 percent. The nitrogen grade of both 304 and 316 was proposed to have a tensile strength of 80 000 psi (552 MPa) in place of 75 000 psi (517 MPa) for the regular grade, and yield strength of 35 000 psi (241 MPa) in place of 30 000 psi (207 MPa). Allowable stresses were requested to 1000 F (538 C) for Section I and Section VIII, Division 1, and also for Section III and Section VIII, Division 2. Considerable data on commercial heats of 304N and 316N had been generated by Cameron Iron Works in support of the Case proposed by Westinghouse.

In November 1968 at the ASTM Committee A-1 Meeting at Williamsburg, Va., discussions were held regarding the negative votes that had been received in Subcommittee X, with no resolution of the differences. A meeting in Pittsburgh in December 1968 ended with the suggestion that the Grade 304H be replaced with 304N, and that the nitrogen content be reduced from a range of 0.10 to 0.20 to 0.10 to 0.16. This also failed to resolve the negative votes and brought out the further objection that use of nitrogen grades should be restricted to temperatures below 1350 F (732 C).

The request to the ASME Boiler Code Committee for a Case to permit use of the nitrogen grades of 304 and 316 had received negative votes based on the suggestion that the case should be broadened to include plate, sheet, and forgings rather than be limited to pipe products, as Case 1423 was written originally. Case 1423 was approved in May 1969. Although data on tubular products had been received from a number of sources, there had been no data received for the other wrought product forms until June 1969, when data for plates and forgings were submitted.

The Case was revised; Boiler and Pressure Vessel Committee Code Case 1423-1 was adopted in October 1969, and expanded the use of Grades 304N and 316N to include wrought product forms for welded construction for temperatures up to 1200 F (649 C) with an allowable stress at that temperature of 6050 psi (41.7 MPa), see Tables 1 and 2. (Note that in Table 2, dual values are shown under the heading "Maximum Allowable Stress"). As shown in Table 1, the improved strength of the nitrogen Grades 304N and 316N was carried up to 1050 F (566 C). At temperatures of 1100, 1150, and 1200 F (593, 621, and 649 C), the allowable stress is the same for both the nitrogen grades, the "H" Grades, and the regular grades of 304 and 316. The allowable stresses, or design stress intensity values, permitted by Case 1423-1 and 1423-2 are shown in Table 2. For detailed tabulations regarding published allowable stresses, the reader is referred to the appropriate tables

	N	Iaximum All	owable Str	ess	a. T			
		Class 2 au	section III		Stress I Section III	ntensity Class 1, and		
		Section VIII	, Division	1	Section VIII, Division			
Temperature					Grade	Grade		
deg F (deg C)	Grade	304N	Grade	e 316N	304 N	316N		
75 (24)	20.0	20.0	20.0	20.0	23.33	23.33		
100 (38)	20.0	20.0	20.0	20.0	23.33	23.33		
200 (93)	17.9	20.0^{b}	19.4	20.0^{b}	23.33	23.33		
300 (149)	15.7	19.0^{b}	17.8	19.2^{b}	22.6	23.33		
400 (204)	14.1	18.3^{b}	16.5	18.8^{b}	20.4	22.33		
500 (260)	13.0	17.8^{b}	15.4	18.6^{b}	18.7	22.2		
600 (316)	12.4	17.40	14.6	18.6^{b}	17.8	21.1		
650 (343)	12.2	17.30	14.2	18.6^{b}	17.55	20.5		
700 (371)	11.9	17.150	13.9	18.6^{b}	17.2	20.1		
750 (399)	11.75	16.9^{b}	13.6	18.5^{b}	16.9	19.6		
800 (427)	11.55	16.6^{b}	13.3	18.40	16.65	19.2		
850 (454)	11.3	16.3^{b}	13.1	18.3^{b}				
900 (482)	11.05	15.9^{b}	12.8	18.1^{b}				
950 (510)	10.8	15.6^{b}	12.6	17.80				
1000 (538)	10.55	15.0%	12.4	17.4^{b}				
1050 (566)	10.3	12.4^{b}	12.2	15.8^{b}				
1100 (593)	9.75	9.75	11.7	12.4^{b}		• • •		
1150 (621)	7.7	7.7	9.8	9.8				
1200 (649)	6.05	6.05	7.4	7.4				

TABLE 2—ASME Code Case 1423-2* design stress, ksi^a.

^a To convert ksi to MPa, multiply by 6.894757.

^b Due to the relatively low yield strength of these materials, these higher stress values were established at temperatures where the short time tensile properties govern to permit the use of these alloys where slightly greater deformation is acceptable. The stress values in this range exceed $62\frac{1}{2}$ percent, but do not exceed 90 percent of the yield strength at temperature. Use of these stresses may result in dimensional changes due to permanent strain.

* Approved by ASME Council March 9, 1972.

in Section I, Section III, and Section VIII of the ASME Boiler and Pressure Vessel Code [3].

With the adoption of Case 1423-1, the way was cleared for inclusion of Grades 304N and 316N in appropriate ASTM specifications, and such action was begun in Committees A-1 and A-10. (Both Committees were combined in a new A-1 Committee in June 1972). The chemical composition and mechanical properties adopted in Code Case 1423-1 and later amended to Case 1423-2 are given in Table 3. The Grades 304N and 316N being considered for inclusion in ASTM specifications are to have the composition and mechanical properties shown in Case 1423-2.

In summary, there is agreement that nitrogen is a strengthening agent within the elastic range, but there is an area of disagreement as regards whether or not nitrogen is beneficial in the plastic or creep range. Also, TABLE 3—Inquiry form of ASME Code Case 1423-2.

INTERPRETATIONS OF ASME BOILER AND PRESSURE VESSEL CODE

Approved by Council March 9, 1972

Case 1423-2 (Special Ruling) Wrought Type 304 & 316 with Nitrogen Added. Sections I, III, VIII, Divisions 1 and 2

Inquiry: Is it permissible in welded construction to the requirements of Section I, III, and VIII to use the austenitic grades of 304 and 316, with controlled additions

1. The chemical requirements shall be as follows:

of nitrogen, as wrought pipes and tube and as other wrought product forms conforming to the requirements of the applicable SA material specification listed in Tables PG-23.1 of Section I, I-1.2 of Section III, UHA-23 of Section VIII Division 1, and AHA-1 of Section VIII Division 2, except that the material shall conform to the following chemical and tensile requirements?

	Grade 304N	Grade 316N
Carbon, % max	0.08	0.08
Manganese, % max	2.00	2.00
Phosphorus, % max	0.03	0.03
Sulphur, % max	0.03	0.03
Silicon, % max	0.75	0.75
Nickel, %	8.0 to 11.0	11.0 to 14.0
Chromium, %	18.0 to 20.0	16.0 to 18.0
Molybdenum, %		2.0 to 3.0
Nitrogen, %	0.10 to 0.16	0.10 to 0.16
2 The room temperature minimum		

2. The room temperature minimum mechanical properties shall be as follows:

	Grade 304N	Grade 316N
Ultimate tensile strength, psi 0.2% offset yield strength, psi % Elongation2 in. (51 mm) gage length	80 000 (552 MPa)	80 000 (552 MPa)
0.2% offset yield strength, psi	35 000 (241 MPa)	35 000 (241 MPa)
% Elongation-2 in. (51 mm) gage length		
longitudinal	30	30
transverse	25	25
% reduction of area		
longitudinal	50	50
transverse	45	45

there is a minor difference of opinion regarding the nitrogen range; the range favored for elevated temperature use of 304N and 316N is 0.10 to 0.16, as adopted in the Case. Three public utilities have used 304N for high temperature steam piping, and use of the nitrogen grades is expected to grow.

References

[1] Freeman, J. W., and White, J. E., Transactions, American Society of Mechanical Engineers, Vol. 85, Series A, No. 2, April 1963, pp. 119–146.

- [2] Goodell, P. D., Cullen, T. M., and Freeman, J. W., "The Influence of Nitrogen and Certain Other Elements on the Creep Rupture Properties of Wholly Austenitic Type 304 Steel", presented at the Metals Engineering Conference, The American Society of Mechanical Engineers, Houston, Tex., April 1967.
- [3] "Boiler and Pressure Vessel Code, Section I, 1971, Table PG-23.1, pp. 146-153, Section III, ASME Boiler and Pressure Vessel Code Case 1423-2, Case Interpretations, 1972, pp. 603-605, and Section VIII, 1971, Table UHA-23, pp. 158-168", American Society of Mechanical Engineers.

Creep and Creep-Rupture Properties of Types 304N and 316N Stainless Steels

REFERENCE: Domis, W. F., "Creep and Creep-Rupture Properties of Types 304N and 316N Stainless Steels," Elevated Temperature Properties as Influenced by Nitrogen Additions to Types 304 and 316 Austenitic Stainless Steels, ASTM STP 522, American Society for Testing and Materials, 1973, pp. 86-99.

ABSTRACT: If Types 304 and 316 steels with improved elevated temperature strength existed, equipment manufacturers could design to higher stresses, thereby reducing the weight and consequently the cost of fabricated equipment. The addition of nitrogen appears to be one means of improving the strength of Types 304 and 316 steels with little additional cost. Although appreciable work has been done, most of it over a limited temperature range, the effect of nitrogen on the creep and creep-rupture properties of Types 304 and 316 steels within AISI composition to determine quantitatively the effect of nitrogen on the (a) creep and creep-rupture properties of Types 304 and 316 steels within AISI composition limits over a large temperature range and (b) toughness of Types 304 and 316 steels after long time exposures at elevated temperatures.

KEY WORDS: nitrogen, creep rate, creep rupture strength, creep strength, energy absorption, stainless steels, toughness

Austenitic chromium-nickel stainless steels are widely used for architectural, consumer, and industrial applications because of their excellent toughness, formability, and weldability. In addition, because of their excellent corrosion resistance, they are used for handling highly corrosive materials or for resisting severe oxidation. For elevated temperature applications such as tubing and piping in the chemical, oil, and electrical power industries, AISI 304 (18 percent chromium, 9 percent nickel) and 316 (17 percent chromium, 12 percent nickel, 2.5 percent molybdenum)

¹ Research engineer, United States Steel Corp., Applied Research Laboratory, Monroeville, Pa. 15146.

² The italic numbers in brackets refer to the list of references appended to this paper.

stainless steels are frequently used at temperatures up to 1500 F because of their relatively high creep and creep-rupture strengths.

Accordingly, for many elevated temperature applications, it would be desirable to increase the elevated temperature strengths of Types 304 and 316 steels. If these steels with improved elevated temperature strength were available, equipment manufacturers could design to higher stresses, thereby reduce the weight and, consequently, the cost of fabricated equipment.

The addition of nitrogen appears to be one means of improving the strength of Types 304 and 316 steels with little additional cost $[1-6]^2$ Recently, for applications at temperatures up to 1000 F, the American Society of Mechanical Engineers (ASME) Boiler and Pressure Vessel Code Committee assigned design stresses to high nitrogen (0.10 to 0.16 percent) Types 304 and 316 steels [7] that are higher than those assigned to normal nitrogen (0.04 percent) Types 304 and 316 steels. Although appreciable work has been done, the effect of nitrogen on the creep and creep-rupture properties of Types 304 and 316 steels has not been quantitatively established. Some investigators have worked with steels of high carbon content $\lceil 5 \rceil$ and others with steels having boron additions $\lceil 6 \rceil$. Also, most of the investigations have been conducted over a limited temperature range. Therefore, it was decided to determine quantitatively the effect of nitrogen on the creep and creep-rupture properties of Types 304 and 316 steels within AISI composition limits over a large temperature range. Because nitrogen may have an adverse effect on toughness, the effect of nitrogen on the toughness of Types 304 and 316 steels after long time exposures at elevated temperatures was determined. The results of these studies are presented herein.

Materials and Experimental Work

For this study, Type 304N steel, in the form of an 8-in. square bloom, was obtained from a commercial heat intended for a regular production order. The bloom was laboratory hot-rolled at 2200 F to a $4\frac{1}{2}$ -in. square bar and then forged at 2150 F to a 3-in. square bar. Specimens for all

Steel	С	Mn	Р	s	Si	Ni	\mathbf{Cr}	Mo	Ν
A B C D E F	$\begin{array}{c} 0.04 \\ 0.063 \\ 0.065 \\ 0.050 \\ 0.058 \\ 0.059 \end{array}$	$1.45 \\ 1.54 \\ 1.55 \\ 1.47 \\ 1.51 \\ $	$\begin{array}{c} 0.025 \\ 0.016 \\ 0.017 \\ 0.016 \\ 0.014 \\ 0.017 \end{array}$	$\begin{array}{c} 0.017 \\ 0.014 \\ 0.016 \\ 0.014 \\ 0.014 \\ 0.014 \\ 0.014 \end{array}$	$\begin{array}{c} 0.62 \\ 0.52 \\ 0.55 \\ 0.49 \\ 0.53 \\ 0.44 \end{array}$	9.4 10.1 9.8 11.5 11.9 11.8	18.5 19.2 18.8 17.0 16.6 17.3	${\rm ND}^{\alpha} \\ {\rm ND} \\ {\rm 2.32} \\ {\rm 2.38} \\ {\rm 2.52} \\$	$\begin{array}{c} 0.14 \\ 0.065 \\ 0.18 \\ 0.065 \\ 0.13 \\ 0.20 \end{array}$

TABLE 1—Chemical compositions of steels investigated, percent.

^a ND = not determined.

subsequent tests were machined from this 3-in. square bar. The chemical composition of this steel, designated as Steel A, is shown in Table 1.

Because of the limited amount of material available from the aforementioned commercial heat and because of the desire to establish the effect of nitrogen, two 100-lb laboratory heats of Type 304 stainless steel were melted with normal and high nitrogen contents, respectively, and were cast into 4 by 4 by 18-in. ingots. One heat was melted under argon to a nitrogen content of 0.065 percent, and the second heat was melted in air to a nitrogen content of 0.18 percent (Type 304N). The chemical compositions of these two steels, designated as Steels B and C, are also shown in Table 1.

In addition, three 300-lb induction furnace heats of Type 316 stainless steel with controlled variations in nitrogen content were melted, and each heat was cast into molds to obtain three 4-in. square by 18-in. long 100-lb ingots. One heat was melted under argon to a nitrogen content typical of Type 316 steel (0.06 percent), and was designated as Steel D. Two other heats were melted in air to nitrogen contents of 0.13 and 0.20 percent, and were designated as Steels E and F, respectively. The chemical compositions of the three steels are also shown in Table 1.

The ingots of the laboratory produced steels, Steels B, C, D, E, and F were machined, heated to 2150 F, and forged without reheating to 1.25-in. square bars. These bars and the material from the aforementioned commercial heat, Steel A, were annealed at 1975 F for 1 h and water quenched. Subsequently, standard 0.250-in. diameter creep and creep-rupture specimens and Charpy V-notch impact specimens were machined from these annealed bars.

Creep and creep-rupture tests were conducted on all six steels at 1200 and 1350 F. In addition, creep and creep-rupture tests were conducted on Steels A, D, E, and F at 1050 and 1500 F. Charpy V-notch impact specimens machined from bars of Steels C and F, that were exposed at 750, 900, 1050, and 1200 F for times extending to 10 000 hours, were tested at room temperature.

Results and Discussion

Steel A, commercially produced Type 304N steel

The results of creep and creep-rupture tests on Steel A at 1050, 1200, 1350, and 1500 F are summarized in Table 2 with the actual test results shown in Table 3. For comparison, the average creep and creep-rupture strengths for commercially produced Type 304 steel are also included in Table 2. As shown in this table, Steel A exhibited appreciably higher creep-rupture strength than Type 304 steel. For example, the stresses for rupture in 10 000 and 100 000 h at 1200 F for Steel A were 18.0 and 12.5 ksi, respectively, compared with 13.8 and 9.8 ksi for Type 304 steel. At 1500 F,

Test	Stress, ksi, fo	or Rupture in	Stress, ksi, for a Minimum-Creep Rate of		
Temperature, · deg F	10,000 h	100 000 h	0.0001 %/h	0.00001 %/h	
1050	34.0	27.0	31.0	25.0	
1200	18.0	12.5	15.0	10.2	
1350	8.8	6.0	7.5	5.5	
1500	4.8	3.3	4.0	a	
Averag	e Creep and Cree	ep-Rupture Proper	ties for Type 304 S	lteel ^b	
1050	28.0	20.0	20.5	14.0	
1200	13.8	9.8	10.8	7.2	
1350	6.7	4.7	5.7	3.6	
1500	3.25	2.3	2.95	1.8	

 TABLE 2—Creep and creep-rupture properties of Steel A (commercially produced Type 304N steel).

^a Insufficient data.

^b Data obtained from ASTM DS5S2, Supplement to DS5, Feb. 1969.

NOTE—Above values were obtained from interpolation and extrapolation of data on logarithmic plots.

these stresses were 4.8 and 3.3 ksi for Steel A, and 3.25 and 2.3 ksi for Type 304 steel. Overall, the stresses for rupture in 10 000 and 100 000 h at the four temperatures investigated were 21 to 47 percent higher for Steel A than the average stresses for Type 304 steel.

The creep data for Steel A are also summarized in Table 2. The results show that the creep strength of Steel A was appreciably higher than that of Type 304 steel. For instance, the stresses for minimum-creep rates of 0.0001 and 0.00001 percent per hour at 1200 F for Steel A were 15.0 and 10.2 ksi, respectively, compared with 10.8 and 7.2 ksi for Type 304 steel. At 1500 F, the stress for a minimum-creep rate of 0.0001 percent was 4.0 ksi for Steel A, and 2.95 ksi for Type 304 steel. Overall, the stresses for minimum-creep rates of 0.0001 and 0.00001 percent per h at the four temperatures investigated were about 31 to 78 percent higher for Steel A than the average stresses for Type 304 steel.

In Table 3, note that all the Steel A specimens tested at 1050 and 1200 F fractured at the gage marks, which may be indicative of notch weakening. Notched-bar tests would have to be conducted to confirm that the steel is notch sensitive. The minimum elongation (specimen fractured at the gage marks) and reduction of area values exhibited by Type 304N steel were 8 and 16 percent, respectively.

Steels B and C, laboratory produced Type 304 and 304N steels

The results of creep and creep-rupture tests on Steels B and C at 1200 and 1350 F are summarized in Table 4, and the actual test results are

Test Temperature, deg F	Stress, ksi	Time to Rupture, h	Elongation in 1 in., %	Reduction of Area, %	Minimum- Creep Rate, %/h
1050	50.0	208.5	18^a	38.5	0.0129
	45.0	579.8	15ª	34.7	0.00437
	40.0	1303.9	10ª	33.4	0.00156
	37.5	2733.3	9ª	28.8	0.00049
	35.0	7692.2	8ª	19.1	0.00022
	32.5	3000.05	\mathbf{ND}^d	ND^d	0.000135
1200	35.0	99.0	13ª	28.8	ND^d
	30.0	443.1	13ª	26.0	0.0069
	27.5	902.5	10^a	20.0	0.0035
	25.0	1411.9	9a	18.4	0.00152
	22.5	3197.5	10 ^a	16.0	0.00113
	20.0	5197.0	10^a	21.1	0.00069
	17.0	9000.0°	\mathbf{ND}	\mathbf{ND}	0.000216
1350	20.0	88.6	40	30.8	0.120
	17.5	217.7	35	32.1	0.0424
	15.0	675.9	39	43.4	0.0144
	12.5	1910.9	47	44.6	0.00562
	10.0	4720.9	34	34.4	0.00081
	8.0	8000.0°	ND	\mathbf{ND}	0.000146
1500	12.5	34.5	52	48.0	0.469
	10.0	128.7	50	46.2	0.111
	8.0	542.2	38	38.4	0.00713
	6.0	2443.6	30	34.8	0.00206

 TABLE 3—Creep and creep-rupture properties of Steel A (commercially produced Type 304N steel).

^a Fractured at gage marks.

^b Test discontinued prior to failure.

^c Elapsed time, test still in progress.

 d ND = not determinable.

TABLE 4—Creep and creep-rupture properties of Steels B and C.

Test	Stress, ksi, for Rupture in		Stress, ksi, for a Minimum-Creep Rate o	
deg F	1000 h	10 000 h	0.001%/h	0.0001%/h
	Steel B—Lab	oratory Produced 1	ype 304 Steel.	
1200	18.5	13.5	12.0	8.7
1350	10.5	6.4	5.8	4.2
	Steel C-Labo	ratory Produced Ty	ype 304N Steel.	
1200	23.0	17.5	17.5	12.0
1350	13.0	7.9	9.0	6.3

Note—Above values were obtained from interpolation and extrapolation of data on logarithmic plots.

Test Temperature, deg F	Stress, ksi	Time to Rupture, h	Elongation in 1 in., %	Reduction of Area, %	Minimum- Creep Rate, %/h
AL	Steel B-L	aboratory Prod	luced Type 30.	4 Steel.	
1200	30.0 25.0 20.0 17.0	$\begin{array}{c} 22.9 \\ 122.2 \\ 605.2 \\ 1861.1 \end{array}$	38 41 43 47	43.3 41.2 40.6 42.7	$1.13 \\ 0.192 \\ 0.0353 \\ 0.0126$
1350	20.0 15.0 12.5 10.0 8.0 Steel C—La	10.2 75.5 253.7 1391.9 3878.6 boratory Produ	55 53 57 57 47 uced Type 304.	51.0 52.8 50.6 48.9 48.9 N Steel.	$\begin{array}{c} 2.63 \\ 0.405 \\ 0.136 \\ 0.0227 \\ 0.00664 \end{array}$
1200	$\begin{array}{c} 35.0\\ 32.5\\ 30.0\\ 27.5\\ 25.0\\ 22.5\\ 20.0\\ 17.0\\ 15.0 \end{array}$	$\begin{array}{c} 11.1\\ 60.3\\ 118.8\\ 237.1\\ 511.3\\ 1357.8\\ 2808.8\\ 14435.8\\ 11000.0^{b} \end{array}$	6^{a} 10 6^{a} 7^{a} 10 11 13 21 ND ^c	30.8 22.6 21.4 17.6 15.8 12.9 24.1 ND ^c	$\begin{array}{c} ND^c\\ 0.0848\\ 0.0384\\ 0.0193\\ 0.0095\\ 0.0040\\ 0.00208\\ 0.000225\\ 0.00035\end{array}$
1350	20.0 17.5 15.0 12.5 10.0 8.0	$\begin{array}{r} 47.7\\178.8\\412.2\\1147.0\\3864.0\\9539.1\end{array}$	17 26 31 32 31 27	$20.0 \\ 24.3 \\ 31.1 \\ 26.7 \\ 29.4 \\ 21.8$	0.172 0.0682 0.0272 0.0105 0.00236 0.00045

TABLE 5—Creep and creep-rupture properties of Steels B and C.

^a Fractured at gage marks.

^b Test discontinued prior to failure.

 $^{\circ}$ ND = not determinable.

shown in Table 5. As mentioned previously, the chemical compositions of these two steels are similar with the exception of nitrogen. As disclosed in Table 4, Steel C (0.18 percent nitrogen) exhibited appreciably higher creep-rupture strength than Steel B (0.065 percent nitrogen). For example, the stresses for rupture in 1000 and 10 000 h at 1200 F for Steel C were 23.0 and 17.5 ksi, respectively, as compared with 18.5 and 13.5 ksi for Steel B. Overall, the stresses for rupture in 1000 and 10 000 h at the two temperatures investigated were 23 to 29 percent higher for Steel C than the corresponding stresses for Steel B.

As also shown in Table 4, similar to the stress-rupture time results, the creep strength of Steel C was appreciably higher than that of Steel B. For instance, the stresses for minimum-creep rates of 0.001 and 0.0001 percent

Exposure	Energy Absorbed at Room Temperature, ft·lb, After Exposure for Indicated Times			
deg F	1000 h	5000 h	10 000 h	
750	240	240	240	
900	240	240	240	
1050	192	135	112	
1200	106	63	55	

 TABLE 6—Charpy V-notch energy absorption of Steel C (laboratory produced Type 304N steel) after elevated temperature exposure.

Note-Charpy V-notch energy absorption of annealed Type 304N steel is 240 ft·lb.

per hour at 1200 F for Steel C were 17.5 and 12.0 ksi, respectively, as compared with 12.0 and 8.7 ksi for Steel B. Overall, the stresses for minimum-creep rates of 0.001 and 0.0001 percent per hour at the two temperatures investigated were 37 to 55 percent higher for Steel C than for Steel B.

At the two temperatures investigated, Steel C exhibited markedly lower elongation and reduction of area values than Steel B (see Table 5). For a rupture time of 1000 h at 1200 F, for example, Steel C would be expected to exhibit elongation and reduction of area values of approximately 11 and 16 percent, respectively, whereas Steel B would exhibit values of about 44 and 42 percent. For a rupture time of 1000 h at 1350 F, Steel C would be expected to exhibit elongation and reduction of area values of approximately 32 and 27 percent, respectively, whereas Steel B would exhibit values of about 56 and 48 percent. The minimum elongation and reduction of area values exhibited by Steel C in the tests at 1200 F were 10.0 and 12.9 percent, respectively. However, similar to some specimens of the commercially produced Steel A, several of the specimens tested at 1200 F fractured at the gage marks, which again may be indicative of notch weakening.

The results of room temperature Charpy V-notch impact tests on Steel C after exposure at 750, 900, 1050, and 1200 F for times up to 10 000 h are given in Table 6. As shown, the Charpy V-notch energy absorption of Steel C in the annealed condition was 240 ft·lb. Exposures extending to 10 000 h at 750 and 900 F had no effect on the energy absorption characteristics. After an exposure of 10 000 h at 1050 F, the Charpy V-notch energy absorption of Steel C decreased to 112 ft·lb and after 10 000 h at 1200 F to 55 ft·lb. Charpy V-notch energy absorptions of 95 ft·lb have been measured for Type 304 stainless steel after exposure for 10 000 h at 1200 F [8]. Although it appears that the addition of nitrogen decreased the energy absorption of Type 304 steel after exposure at elevated temperatures, the energy absorption of Type 304N steel was still very high.

In summation, nitrogen appears to increase markedly the creep and creep-rupture strengths of Type 304 steel within the temperature range investigated. However, as would be expected, this increase in strength is accompanied by a decrease in ductility (as measured by elongation and reduction of area) and a slight drop in room temperature energy absorption after elevated temperature exposure. Additional testing would be required to determine whether this decrease in ductility is detrimental to the life of an elevated temperature component.

Steels D, E, and F, laboratory produced Types 316 and 316N steels

The results of creep and creep-rupture tests at 1050, 1200, 1350, and 1500 F on Steels D, E, and F are summarized in Table 7 with the actual test data shown in Tables 8 through 10. Table 7 shows that nitrogen increased the strength of Type 316 steel at the temperatures investigated, with the exception that at 1500 F, the 0.13 percent nitrogen of Steel E did not result in increased strength. At 1200 F, for example, the stresses to produce rupture in 10 000 h for Steels D, E, and F were 15.0, 19.0, and 21.5 ksi, respectively. At 1050 F, the stresses to produce rupture in 10 000 h for Steels D, E, and F were 15.0, 19.0, and 21.5 ksi, respectively. At 1050 F, the stresses to produce rupture in 10 000 h for Steels D, E, and F were 15.0, 19.0, and 21.5 ksi.

Test	Stress, ksi, f	or Rupture in	Stress, ksi, for a Minimum-Creep Rate		
deg F	1000 h	10 000 h	0.001%/h	0.0001%/h	
	Steel 1	D (0.065 percent n	itrogen)		
1050	39.0	31.5	35.0	27.5	
1200	23.2	15.0	15.0	10.9	
1350	12.0	8.2	6.7	4.5	
1500	6.5	4.2	4 . 2	2.8	
	Steel 1	E (0.13 percent nit	rogen)		
1050	41.5	33.0	37.0	29.0	
1200	26.0	19.0	17.0	12.5	
1350	14.0	9.3	8.7	6.1	
1500	6.5	4.0	4.0	2.6	
	Steel	F (0.20 percent ni	trogen)		
1050	42.5	34.5	41.0	33.0	
1200	30.2	21.5	26.0	20.0	
1350	17.0	10.5	11.5	7.5	
1500	8.5	5.2	5.2	3.4	

TABLE 7—Results of creep and creep-rupture tests on Steels D, E, and F (Type 316 steels).

NOTE—Above values were obtained from interpolation and extrapolation of data on logarithmic plots.

Stress, ksi	Time to Rupture, h	Elongation in 1 in., %	Reduction of Area, %	Minimum- Creep Rate, %/h
	Steel D (0.0	065 percent nitro	gen)	
50.0	74.3	37	28.7	0.039
45.0	209.5	19	24.4	0.0193
40.0	749.9	15	20.8	0.00362
35.0	4651.1	17	23.2	0.00102
32.5	7131.4	ND^b	ND^{b}	0.00031
30.0	5000.0^{a}	\mathbf{ND}	\mathbf{ND}	0.000242
	Steel E (0.	13 percent nitrog	gen)	
55.0	73.5	27	32.6	0.0758
50.0	137.6	21	24.9	0.0215
45.0	357.5	15	23.3	0.00916
40.0	1171.3	13	20.0	0.00246
35.0	2838.6	13	19.3	0.000937
	Steel F (0.	20 percent nitrog	gen)	
55.0	161.2	18	27.6	0.0090
50.0	272.5	13	17.1	ND^b
45.0	506.0	12	25.3	0.0028
42.5	960.5	$\overline{12}$	19.5	0.00159
40.0	1237.0	10	13.9	0.00086
35.0	8543.0	5	8.1	0.000128
30.0	6200.0^{a}	ND	ND	0.000040

TABLE 8—Creep and creep-rupture properties of Steels D, E, and F at 1050 F (Type 316 steels).

^a Discontinued prior to failure.

^b ND = not determinable.

The creep strength of Type 316 steel also increased with increasing nitrogen content at 1050, 1200, 1350, and 1500 F (see Table 7), except that Steels D and E again exhibited similar creep strengths at 1500 F. At 1200 F, for instance, the stresses to produce a minimum-creep rate of 0.0001 percent per hour for Steels D, E, and F were 10.9, 12.5, and 20.0 ksi, respectively. At 1500 F, the stresses to produce a minimum-creep rate of 0.0001 percent per hour for Steels D, E, and F were 2.8, 2.6, and 3.4 ksi. However, longer time tests would be desirable to verify that the 0.13 percent nitrogen of Steel E did not result in increased strength at 1500 F.

At 1050 F (Table 8) increasing nitrogen content appeared to decrease the ductility somewhat (as measured by elongation and reduction of area values) of Type 316 steel. For example, the estimated elongations at rupture for a rupture time of 1000 h were 15.0, 13.0, and 11.5 percent for Steels D, E, and F, respectively. In longer time tests, 0.20 percent nitrogen further decreased the elevated temperature ductility of Type 316 steel at 1050 F, with Steel F exhibiting an elongation of five percent in the 8543.0-h test. This may be too low to be acceptable in the design of most components for applications at this temperature. Tests on Steel E for longer times are in progress to determine the effect of 0.13 percent nitrogen on its ductility.

At the higher temperatures of 1200, 1350, and 1500 F, however, the trend of elongation and reduction of area values with respect to rupture time varied for Steels D, E, and F. At 1200 F, as shown in Table 9, the elongation at rupture of Steel D generally decreased with increasing rupture time, whereas the elongation at rupture of Steel F generally increased with increasing rupture time. In the 14.3-h test at 1200 F, Steel D exhibited a 48 percent elongation at rupture, but this decreased to 24 percent in the 5963.2-h test. However, in the 6.7-h test, Steel F exhibited 16 percent elongation at rupture, and this increased to 52 percent in the 6316.2-h test. At 1350 F (Table 10) similar trends in the elongation at rupture of the three steels are again evident. The elongation of Steel D remained essentially constant with time, and the elongations of Steels E and F generally

Stress, ksi	Time to Rupture, h	Elongation in 1 in., %	Reduction of Area, %	Minimum- Creep Rate, %/h
	Steel D (0.	065 percent nitro	ogen)	
35.0	14.3	48	56.6	ND ⁶
30.0	113.8	54	57.1	0.164
25 0	548 1	66	62 0	0 0363
22.5	1251.3	50	51.0	0.0147
20.0	2180.1	39	41.9	0.00806
17.0	5963.2	24	28.1	0.00213
	Steel E (0.	13 percent nitro	gen)	
40.0	20.3	39	41.3	ND
35.0	88.5	39	37.1	0.175
30.0	261.2	23	30.1	0.4077
25.0	1556.3	44	48.4	0.0109
20.0	3392.5	65	59.3	0.00428
17.0	8000.0^{a}	ND^b	ND^{b}	0.00128
	Steel F (0.	20 percent nitro	gen)	
45.0	6.7	16	27.3	ND
40.0	136.3	13	16.8	0.030
35.0	285.7	12	21.3	0.0105
30.0	1067.5	25	36.6	0.00356
27.5	2344.8	42	42.7	0.00253
25.0	3648.3	26	33.6	0.00082
22.5	6316.2	52	54.0	0.00029

TABLE 9—Creep and creep-rupture properties of Steels D, E, and F at 1200 F (Type 316 steels).

^a Discontinued prior to failure.

 b ND = not determinable.

 Stress, ksi	Time to Rupture, h	Elongation in 1 in., %	Reduction of Area, %	Minimum- Creep Rate, %/h
	Steel D (0.)	065 percent nitro	gen)	
20.0	35.8	59	63.0	0.671
17.5	106.2	82	69.2	0.129
15.0	265.6	77	76.8	0.0848
12.5	847.1	75	71.3	0.0345
10.0	3208.9	78	67.0	0.00986
8.0	8000.0^{a}	ND^b	ND^{b}	0.00238
7.0	2000.0^{a}	ND	ND	0.00089
	Steel E (0.	13 percent nitrog	gen)	
25.0	27.4	52	50.6	0.698
20.0	116.9	72	62.7	0.241
15.0	585.6	74	60.2	0.0367
12.5	1787.7	103	65.7	0.0146
10.0	5969.4	64	56.3	0.00352
	Steel F (0.	20 percent nitrog	ien)	
30.0	36.7	22	25.9	0.278
25.0	163.2	35	37.7	0.072
22.5	290.0	52	48.4	0.0398
20.0	587.3	61	56.8	0.0209
17.5	927.0	65	66.6	0.00962
15.0	1654.2	86	70.3	0.00515
12.5	2940.1	55	51.0	0.00196
8.0	8710.0ª	ND	ND	0.000136

TABLE 10—Creep and creep-rupture properties of Steels D, E, and F at 1350 F (Type 316 steels).

^a Discontinued prior to failure.

^b ND = not determinable.

increased with time. However, at 1500 F the elongations of all three steels decreased with increasing rupture time (Table 11). In general, the addition of nitrogen did not appear to lower the long time elongation and reduction of area values below acceptable levels at temperatures in the 1200 to 1500 F range.

The results of room temperature Charpy V-notch impact tests conducted on Steel F after exposure at 750, 900, 1050, and 1200 F for times up to 10 000 h are given in Table 12. The Charpy V-notch energy absorption of Steel F in the annealed condition was 240 ft·lb. Exposures extending to 10 000 h at 750 and 900 F had no effect on the energy absorption characteristics. However, after exposure for 10 000 h at 1050 F, the energy absorption decreased to 46 ft·lb, and after exposure for 10 000 h at 1200 F, to 10 ft·lb. Charpy V-notch energy absorptions of 70 ft·lb have been measured for Type 316 stainless steel that had been exposed for 10 000 h

Stress, ksi	Time to Rupture, h	Elongation in 1 in., %	Reduction of Area, %	Minimum- Creep Rate, %/h
	Steel D (0.	065 percent nitro	ogen)	
12.5	28 4	78	70.6	0.833
10.0	98.8	88	65.2	0.275
8.0	392.4	55	56.4	0.0570
6.0	1179.5	50	41.9	0.0165
5.0	4293.6	23	29.5	0.00144
4.0	2000.0^{a}	ND^{b}	ND^b	0.00075
	Steel E (0.	13 percent nitrog	gen)	
12.5	45.7	80	60.1	ND^{b}
10.0	149.2	62	53.7	0.18
8.0	424.0	54	48.7	0.0563
6.0	1461.3	44	41.5	0.0134
5.0	3551.1	45	34.7	0.00503
	Steel F (0.	20 percent nitrog	gen)	
17.5	35.8	65	60.6	0.518
15.0	80.8	62	59.6	0.212
12.5	165.5	77	65.4	0.0825
10.0	477.9	ND	54.7	0.0185
6.0	4864.5	38	42.5	0.00365
5.0	2300.0^a	\mathbf{ND}	\mathbf{ND}	0.00045

TABLE 11-Creep and	creep-rupture	properties	of	Steels	D,	Ε,	and	F	at	1500	F
	(Type	316 steels).									

^a Discontinued prior to failure. ^b ND = not determinable.

TABLE 12—Charpy	V-notch energy	absorption	of	Steel	F	after	elevated
	temperature	exposure.					

Exposure	Energy Abso After F	orbed at Room Temp Exposure for Indicate	erature, ft•lb, d Times
deg F	1000 h	5000 h	10 000 h
750	240	240	240
900	240	240	240
1050	166	70	46
1200	44	20	10

Note-Charpy V-notch energy absorption of Steel F in the annealed condition was 240 ft·lb.

at 1200 F [8]. It is believed that the deterioration in energy absorption with exposure at elevated temperatures would also occur in the lower nitrogen (0.13 percent) steel, although probably at a slower rate. The energy absorption of Steel E after 10 000 h exposure at 1200 F would presumably be between 10 and 70 ft·lb. However, further testing would be required to verify this.

As was true for the Type 304 steel, the addition of nitrogen to the Type 316 steel increased the creep and creep-rupture strengths within the temperature range investigated with the exception of the 0.13 percent nitrogen addition at 1500 F. These additions of nitrogen, in turn, decreased the long time creep-rupture ductility at 1050 F. Nevertheless, nitrogen did not appear to affect the long time ductility at the other three temperatures investigated, namely, 1200, 1350, and 1500 F. The 0.20 percent nitrogen addition did, however, sharply decrease energy absorption of Type 316 steel after elevated temperature exposure.

Summary

The results of the current investigation on the effect of nitrogen on the creep and creep-rupture properties of Types 304 and 316 steels may be summarized as follows:

1. Type 304N steel exhibited appreciably higher creep and creep-rupture strengths at 1050, 1200, 1350, and 1500 F than did Type 304 steel.

2. Type 304N steels exhibited lower elongation and reduction of area values than Type 304 steel and a high incidence of fracture at the gage marks of the specimens, which may be indicative of notch weakening.

3. The Charpy V-notch energy absorption of Type 304N steel after 10 000 h exposure at 1050 and 1200 F was moderately lower than that of Type 304 steel.

4. The creep and creep-rupture strengths of Type 316 steel at 1050, 1200, 1350, and 1500 F generally increased with increasing nitrogen contents from 0.06 up to 0.20 percent.

5. The addition of nitrogen to Type 316 steel did not adversely affect the long time creep-rupture elongation and reduction of area values at 1200, 1350, and 1500 F, but did decrease the creep-rupture ductility after long time exposure at 1050 F.

6. The addition of 0.20 percent nitrogen to Type 316 steel resulted in a sharp decrease in Charpy V-notch energy absorption after 10 000 h exposure at 1050 and 1200 F.

References

[1] Gerlach, H. and Kautz, H. R., Stahl und Eisen, Vol. 88, No. 25, 1968.

- [2] Goodell, P. D., Cullen, T. M., and Freeman, J. W., Transactions, American Society of Mechanical Engineers, Sept. 1967, pp. 517-524.
- [3] Smith, G. V., Garofalo, F., and Zwell, L., "Influence of Carbon and Nitrogen on Creep and Creep-Rupture Properties and Structure of 18Cr-8Ni and 18Cr-8Ni-Mo Steels,"

presented at the American Institute of Mining, Metallurgical, and Petroleum Engineers Symposium, The Relation of Structure to High Temperature Properties, Chicago, 4 Nov. 1957.

- [4] Brady, R. R., Transactions, American Society for Testing and Materials, Vol. 59, 1959, pp. 774-785.
- [5] Kawabe, Y., Nakagawa, R., and Mukoyama, T., Transactions, Iron and Steel Institute of Japan, Vol. 8, No. 6, 1968.
- [6] Mercier, A., Leveque, R., and Remy, G., Revue De Metallurgie, Vol. 64, No. 12, 1967.
- [7] "Types 304N and 316N Modified Stainless Steel in Pipe Product Form," Interpretations of ASME Boiler and Pressure Vessel Code, Case 1423, American Society of Mechanical Engineers, Approved by Council, 30 May 1969.
- [8] Mechanical and Physical Properties of Austenitic Chromium-Nickel Stainless Steels at Ambient Temperatures, Section I, Bulletin A, The International Nickel Company, 1963.
- [9] An evaluation of the Yield, Tensile, Creep, and Rupture Strengths of Wrought 304, 316, 321, 347, Stainless Steels at Elevated Temperatures, ASTM DS 5S2, Supplement to DS5, American Society for Testing and Materials, Feb. 1969.

Service Experience with H Grades of Austenitic Steel

REFERENCE: Schnabel, G. J., "Service Experience with H Grades of Austenitic Steel," Elevated Temperature Properties as Influenced by Nitrogen Additions to Types 304 and 316 Austenitic Stainless Steels, ASTM STP 522, American Society for Testing and Materials, 1973, pp. 100–104.

ABSTRACT: Premature creep-rupture swelling of AISI 321 austenitic stainless steel superheater tubes resulted in failures in the middle 1950s and promoted an extensive investigation. These failures in large utility boilers operating at pressures above 1500 psi and temperatures above 1000 F seriously challenged the economics and reliability of high pressure and high temperature. Failures were initially attributed to an extremely fine grain size occasioned by a low temperature solution heat treatment. Reheat treatment at higher temperatures to produce a coarser grain structure reduced the incidence of failure.

Subsequent results of a research project provided evidence that the final solution heat treatment and not the grain size was the most significant factor. Control of the carbon content was also found to be an important factor. A new grade called the "H" grade was introduced which incorporated specific heat treatments and control of carbon. This was applied to the 300 series steel in all product forms. Service experience has been successful with the H grades.

Recent investigations of Type 304 austenitic stainless steel indicate that a controlled nitrogen addition significantly enhanced the rupture life of this material, thereby increasing reliability. While the quest for higher temperature steam fossil units has been blunted with the advent of nuclear power, the fact still exists that higher pressures and temperatures result in significant increases in efficiency, and eventually will be pursued.

KEY WORDS: nitrogen, creep rupture, failures, austenitic stainless steel, solution heat treatment, grain size, high pressure, high temperature

Failures in service of AISI 321 austenitic stainless steel superheater tubing occurred in the middle 1950s. Premature creep-rupture swelling was found as the cause. The tubing material exhibited extremely fine grain

¹ Assistant to chief mechanical engineer, Public Service Electric and Gas Co., Newark, N.J. 07101.

www.astm.org

structure. The "H" grades were introduced to establish a more positive control of the short time creep-rupture properties and to assure continuation of long term service life. Experience with the H grades has been successful with service time up to three times the service life of the pre H grades which exhibited difficulty.

History

AISI 321 austenitic stainless steel was used in quantity immediately after World War II when the utility industry placed orders for new electric generating facilities to keep pace with increased power demands and to replace units which had been used past their normal retirement period during the war. Many of the new units were designed for 1500 psi (10.3 MPa) and 1000 to 1050 F (538 to 566 C) operating conditions. These were built to the 1946 or 1949 American Society of Mechanical Engineers (ASME) Boiler and Pressure Vessel Code. At that time the code allowable stresses were considerably lower than those published in the 1952 through 1962 codes.

The increased allowable stresses between 1949 and 1952 were primarily due to changes in criteria. These enabled a greater production with less use of materials, particularly the scarce ones which were needed by the military in the Korean conflict. No specific changes were made in the material specifications. The utilities, like many other industries who use pressure vessels, look to the ASME Boiler and Pressure Vessel Code to provide minimum standards for safety and reliability of operation.

The ASME Code is recognized by most of the state regulation boards as a basic minimum safety code, and as such, they have adopted this code with little, if any, modification as their legal codes. Because the power industry needs huge outlays of money and equipment to meet the demands of electricity in the United States, it has developed into a noncompetitive industry with a captive clientele. The government has several agencies such as the Public Utilities Commission and the Federal Power Commission to assure that all such utilities are operated in a safe, reliable, and efficient manner. Therefore, the earning power and costs are carefully monitored to assure that no undue irregularities exist. Consequently, the utilities have diligently searched for new and better ways of producing power more economically while maintaining the safety and reliability expected of them in serving the public. Increasing pressure, temperature, and size have enabled the utilities to produce electricity more economically. Therefore, when the allowable stresses were increased between 1949 and 1952, greater savings were available for the same types of unit produced or installed previously, and newer, larger, and higher temperature units were made more attractive economically. As a comparison, the same size Type 321 tube used for a 1750 psi (12.1 MPa), 1050 F (566 C) unit could now be used for a 2400 psi (16.5 MPa), 1100 F (593 C) unit.

Problems

The initial failures of superheater tubes with thinner walls and higher operating stresses occurred after approximately three years of service at elevated temperatures and was most serious in units operating at 1050 or 1100 F (566 or 593 C) terminal steam temperatures. It was found that two tubes welded together would show excessive swelling in the tube which had extremely small grain size while the tube on the other side of the weld, which exhibited a grain size of seven or coarser, did not indicate any excessive swelling. Investigations showed that the tubes had received a 1750 to 1900 F (954 to 1038 C) solution heat treatment, but it was difficult to establish the exact temperature level. Laboratory heat treatment of the fine grain tubes showed a response of grain coarsening when reheated to more than 2000 F (1093 C). Subsequently, it was decided to reheat all of the tubing which did not show an excess of two percent creep, and to replace the tubes that did with a tube material which was heat treated to produce a grain size of seven or coarser.

The degree of creep damage which occurred before reheat treatment was determined by a physical measurement of the outside diameter of the tube. After heat treatment a new primary creep growth occurred again. Therefore, some of the early reports of continued creep after heat treatment were probably because of measurements taken too soon after the reheat treatment.

Investigation

A research project sponsored by the ASTM-ASME Joint Committee on the Effect of Temperature on the Properties of Metals under the Steam Power Panel was conducted by Dr. Freeman of the University of Michigan [1].² The research showed that heat treatment and not grain size was the most significant factor. It was also found that a carbon level of 0.04 percent minimum was needed to assure higher creep rupture strength at least for the short time portion of the rupture life. It should be noted that all of the problems and corrections were related to Type 321 materials, primarily in superheater tube product forms. The institution of the H grades, however, has been applied to most of the 300 series austenitic steels and to all product forms.

Results

A recent survey of the service history of H grades of materials installed in superheaters and reheaters of high pressure, high temperature boilers indicates that there have been recurrences or continuations of problems because of swelling and rupture failures. All of the cases reported were in materials which were reheat treated after some time in service or had been

² The italic numbers in brackets refer to the list of references appended to this paper.
purchased originally as regular grades and subsequently reheat treated to conform to the H grade requirements before service. In many instances the mode of failure was somewhat clouded by the influence of lug weld attachments. Wall thinning, because of wastage and local overheating due to firing conditions or circulation distribution, has also been a factor. In addition, several failures have been attributed to improper welding or weld contours.

Replacement of superheaters has been dictated, in many instances, by excessive wall thinning caused by wastage rather than by excessive creep. Twenty-one utilities, with nearly 100 boilers built between 1950 and 1963, and with original materials of either pre H, modified H, or H grades of Type 321 material, have all had some difficulty with the pre H grades. Some of the modified H grades were entirely successful while others had difficulty ranging from slight to serious. Many of the more serious situations were those that had been modified after a significant period of service. It appears that the longer the prior service, the more difficulty after reheat treatment.

At least seven of the utilities have elected to replace superheaters with Type 347H. Two utilities have also used 316H or have replaced with 316H. Most of the 347H replacements were also associated with excessive tube wastage of the superheaters. All of the units which were put into service with H grades of materials or were replaced with H grade material had been satisfactory, insofar as creep rupture is concerned. However, since the service life to date of the H grades is approximately half of the pre H grade, an absolute statement of satisfactory service cannot be made. It is significant that the H grade now have up to three times the service life without any indications of similar problems of premature creep failures as the pre H grades had when they first exhibited difficulty.

Additional Developments

Recent investigations [2] have indicated that the rupture life of Type 304 austenitic stainless materials is enhanced by controlled additions of nitrogen. Many of the more recent heats of Type 304 have exhibited higher short time rupture strengths when compared with those produced more than a decade ago. The newer heats were significantly higher in nitrogen content than the older ones. It was also noted that the manganese content was considerably higher in the newer heats. It was difficult to get sufficient data on many older heats because nitrogen had not been an element which had been generally reported. In both cases however, the heats were within the limits of the ASTM specification for manganese, which was reported consistently. Nitrogen has never been a controlled element in the specifications, but it seems reasonable to expect that some elemental difference between the modified H grades and the actual H grades, other than heat treatment and a minimum carbon content, could account for the wide difference in service experience. It has been recently stated that with the advent of nuclear power and the decline in the use of high temperature installations of steam fossil units, the requirements for the H or nitrogen-bearing grades of materials will be of less concern, and, therefore, a continued effort on the H grade type of material is questionable. It seems reasonable, however, that for the same reasons that the industry has gone to the high temperature, high pressure units in the fossil industry, it will likewise, in the near future, be reaching for the high temperature, high pressure category in the nuclear field as well. The present fossil units will be required to operate for many more years, and when replacements are necessary for one reason or another, they should be made with the best material we know how to produce to maintain reliability.

References

- [1] Freeman, J. W., Journal of Engineering for Power, Vol. 85, Series A, No. 2, April 1963, p. 119.
- [2] "Improved Properties of Type 304 Austenitic Steels," EEI Publication No. 64-67, Edison Electric Institute, Dec. 1964.

Effect of Elevated Temperatures on the Properties of Nitrogen-Bearing Type 216 Steel

REFERENCE: Chivinsky, J. A., "Effect of Elevated Temperatures on the Properties of Nitrogen-Bearing Type 216 Steel," Elevated Temperature Properties as Influenced by Nitrogen Additions to Types 304 and 316 Austenitic Stainless Steels, ASTM STP 522, American Society for Testing and Materials, 1973, pp. 105–115.

ABSTRACT: Preliminary data indicate that Type 216, a high-nitrogen and molybdenum-bearing austenitic stainless steel offers improved strength over conventional Type 316 to temperatures approaching 1350 F (732 C). However, impact and intergranular corrosion tests conducted on the material suggest that Type 216 may be restricted to service temperatures near 1200 F (649 C). The superior strength of Type 216, coupled with its excellent corrosion properties, makes it attractive for application in the process industries (chemical, petrochemical, pulp, paper, etc.) as well as marine and space markets.

KEY WORDS: nitrogen, austenitic stainless steels, mechanical properties, elevated temperatures

The purpose of this presentation is to demonstrate with preliminary data the potential advantages and limitations of a high-nitrogen and molybdenum-bearing austenitic stainless steel. This material, T-216, has superior high temperature strengths (namely, tensile, stress to rupture, and creep) compared to Type 316. Type 216 like Type 316 does show reduced impact strengths and is susceptible to intergranular cracking in acidified copper sulfate after it has been exposed to elevated temperatures for long periods of time. However, the data suggest that Type 216 is suitable for applications in the chemical and petrochemical fields.

Type 216 is a chromium-manganese-nickel-molybdenum-nitrogen grade.

¹Senior metallurgist, Allegheny Ludlum Steel Corp., Div. of Allegheny Ludlum Industries, Inc., Leechburg, Pa. 15656, formerly, senior research metallurgist at the Research Center, Brackenridge, Pa. 15014, when this work was performed.

			TAE	3LE 1—Compo.	sitions (weigh	t percent).			
Material	C	Mn	Р	ß	Si	Cr	Ni	Mo	N
$Type\ 216:$ Heat A Heat A Heat B Heat C Heat D Type $316:$ Heat E Heat F	0.08 max 0.071 0.067 0.045 0.045 0.026 0.026 0.059 0.059	7.50/9.00 8.25 8.08 8.40 7.50 7.50 1.68 1.40	0.045 max 0.020 0.015 0.017 0.018 0.018 0.018 0.021 0.021	0.030 max 0.020 0.023 0.007 0.009 0.009 0.015 0.015 0.018	0.50 max 0.50 0.69 0.69 0.19 0.23 1.00 max 0.55	$\begin{array}{c} 17.50/22.00\\ 19.33\\ 19.57\\ 19.57\\ 19.70\\ 19.70\\ 16.00/18.00\\ 17.32\\ 17.45\end{array}$	5.00/7.00 6.12 5.67 7.06 6.55 6.55 6.55 10.00/14.00 13.72 13.72	2.00/3.00 2.52 2.13 2.13 2.83 2.52 2.60/3.00 2.47 2.45	0.25/0.50 0.32 0.34 0.35 0.35 0.34

TABLE 2-Room temperature mechanical properties.

Product	Type	0.2% Yield Strength, ksi ^b	Ultimate Tensile Strength, ksi ^b	% Elonga- I tion, 2 in. or 50.8 mm	seduction of Area, %	Hardness, R _B ^c	Bend
Annealed. ⁴ Sheet (0.040 in. or 1.02 mm) Sheet (0.060 in. or 1.52 mm) Plate (0.250 in. or 6.35 mm) Plate (0.830 in. or 21.1 mm) Bar (1 in. by 3 in. or 25.4 mm by 76.2 mm) Sheet and bar	216 216L 216L 216L 216L 316 316	$\begin{array}{c} 66.8(47.0)\\ 68.2(47.8)\\ 66.7(46.8)\\ 62.1(43.7)\\ 62.4(43.8)\\ 38(27)\\ \end{array}$	$\begin{array}{c} 115.6(81.3)\\ 115.2(81.0)\\ 115.2(81.0)\\ 115.0(80.8)\\ 112.6(79.3)\\ 108.8(76.5)\\ 85(60)\end{array}$	46.5 45.0 46.3 51.0 51.5 60	64.7 64.7 68.8 68.8 74.1	92 (192) 92 (192) 97 (223) 93 (197) 89 (179)	180 deg; 2T not tested not tested 180 deg; 2T not tested
^a Annealed at 1950 F (1066 C). ^b Values listed in parentheses are metric equiv ^c Values listed in parentheses are Vickers hard	alents (kg Iness equiv	/mm²). ⁄alents.					

107

Material	10% Ferric Chloride Room Temperature 300 h Exposure, pits per in. ² (pits per cm ²)	90% Acetic Acid plus Sodium Chloride ^a (boiling) 300 h Exposure, pits per in. ² (pits per cm ²)
Type 216	Nil (Nil)	81 (12.6)
Type 316	5 (0.78)	523 (81.3)

TABLE 3—Resistance to pitting corrosion.

^a Chloride ion added to a concentration of 200 ppm.

The specific heat analyses, shown in Table 1, are those of the material used in developing these data. It possesses corrosion resistance equivalent to Type 316, and in some instances, particularly pitting media, Type 216 is superior. In addition, the alloy offers high strength in various product forms. Because of this desired combination of properties, it would be appropriate to describe briefly the corrosion and room temperature strengths before discussing high temperature data.

Room Temperature Properties

Strength in Various Product Forms

One can consider Type 216 as a 200 Series counterpart to Type 316 having significantly greater mechanical strength than the chromium-nickel type, due to the nitrogen addition. Type 216 is nonhardenable by heat treatment; annealed products, as shown in Table 2, have yield strengths of approximately 60 000 psi (42.2 kg/mm^2), tensile strengths in excess of 100 000 psi (70.3 kg/mm^2), and elongations in 2 in. (50.8 mm) of 45 percent. Such material can be cold worked to tensile strengths well over 200 000 psi (140.6 kg/mm^2) and remain nonmagnetic. It can be produced as plate, sheet, strip, bars, tubes, and wire using electric furnace (air) melting practices.

Corrosion Behavior

Type 216 possesses corrosion resistance equal to, and in some instances superior to, that of Type 316. For instance:

1. In solutions of 10 percent ferric chloride or chloride containing solu-

Material	Average, Five 48-h Periods in./month (mm/month)
Type 216	0.0010 (0.0254)
Type 304	$0.0018 (0.0408) \\ 0.0008 (0.0203)$

TABLE 4-65 percent boiling nitric acid.

tions of boiling acetic acid, Type 216 displays superior resistance to localized chemical attack compared to Type 316 (see Table 3).

2. In severe oxidizing conditions typified by 65 percent boiling nitric acid, corrosion rates of Type 216 are comparable to the chromium-nickel Types 304 and 316 (see Table 4).

3. Type 216 is slightly less resistant to attack by dilute sulfuric acid when compared to Type 316.

4. Stress corrosion tests in boiling magnesium chloride show Type 216 to be comparable with Type 316; both grades cracked within 4 h of exposure.

Effect of Elevated Temperatures on Properties

More recently, considerable attention has been given by industry to the effects of nitrogen in strengthening conventional grades of stainless steel, Types 304 and 316, at high temperatures. European and American interests are considering up to 0.20 percent nitrogen in conventional chromiumnickel grades for increased creep strengths and tensile strengths. Concern could be expressed over possible embrittling effects of nitrogen when such material has been exposed to elevated temperatures for long periods of time, particularly for materials with levels of nitrogen above 0.20 percent. Therefore, it is only proper that Types 216 and 316 be discussed comparatively with these aspects in mind.

Elevated Temperature Tensile Strengths

Table 5 illustrates the markedly superior short time strengths of Type 216 over that of Type 316 at temperatures up to 1400 F (760 C). The properties of Type 216 were developed on annealed bar material. Note that the elongation values of the high nitrogen material were above 30 percent in every instance.

Welded sheet specimens of Type 216 displayed as-weld strengths and elongations comparable to annealed material. All of these specimens were gas-tungsten-arc (GTAW) welded and failed outside of the weld area. These data are shown in Table 6.

Stress Rupture Strengths

The higher strengths of Type 216 were further substantiated by stress rupture tests as shown in Table 7. Note that the high nitrogen material has markedly superior stress rupture strength at 1200 F (649 C) when compared to Type 316. For longer times at higher temperatures, the rupture strengths of both grades become comparable. Table 7A lists the specific stress rupture data for the specimens tested.

Creep Strengths

Limited testing to date indicates that Type 216 has notably superior creep strength over Type 316 (Table 8). Additional testing at higher temperatures is underway.

Test Tem- perature,	0.2% Yield	Strength, ksi	Ultimate Tensil	e Strength, ksi	% Elongati 50.8	ion, 2 in. or mm	Reduction e	of Area, %
deg C) (deg C)	T-216ª	$T-316^{b}$	$T-216^a$	$T-316^{b}$	$T-216^a$	$T-316^{b}$	$T-216^{a}$	$T-316^{b}$
68(20)	62.4(43.9)	42.4(29.9)	108.8(76.5)	82.4(58.0)	51.3	67.5	74.1	80.5
800(427)	40.1(27.2)	26.5(18.6)	82.8(58.3)	71.4(50.3)	45.0	47.0	70.1	71.0
1000(538)	36.3(25.5)	23.4(16.4)	75.4(53.0)	68.4(48.2)	43.3	55.0	68.7	70.0
1200(649)	35.7(25.1)	22.6(15.8)	66.7(46.8)	50.6(35.6)	42.3	24.0	69.69	32.0
1400(760)	33.0(23.2)	•	51.4(36.1)	30.7(21.6)	36.8	25.5	39.3	35.0
a Specimen mm ²).	s removed from	T-216L (annealed	1 in. by 3 in. bar o	r 25.4 mm by 76.2 1	nm). Values in	parentheses an	e metric equiv	alents. (kg/
^b Allegheny	/ Ludlum Steel	Corporation, Blue	Sheet. Values in pa	rentheses are metri	c equivalents (1	kg/mm²).		

TABLE 5—Short time elevated tensile properties of annealed materials.

6.
31
g
de
vel
ર્સ
nd
a
16
62
be
Γy
~
lee
rea
uu
8
of0
es
rti
be
10
d
ile
ns
te
uo
es
ur
at
er.
ťu
t_{e_i}
p_{i}
at
lev
со L
6
ect
Ð
Ξ
.L
ы
EI
Э
ΓA
- N - 1

Test	0.2% Yield	Strength, ksi ^a	Ultimate Tensi	le ourengun, Ksl"	% Flongation	z in. or 50.5 mm
deg F (deg C)	Annealed	As-Welded ^b	Annealed	As-Welded ^b	Annealed	As-Welded ^b
68(20)	77.2(54.3)	79.7(55.9)	120.8(85.0)	125.9(88.5)	44.0	38.3
800(427)	41.5(29.1)	42.5(29.9)	91.5(64.3)	92.0(64.7)	43.3	36.0
1100(593)	37.3(26.2)	37.7(26.5)	79.7(55.3)	81.2(57.0)	32.8	31.8
1300(704)	33.9(23.4)	36.3(25.5)	62.6(44.0)	64.4(45.3)	37.8	40.0
1500(816)	26.9(18.9)	26.3(17.8)	40.9(27.7)	41.3(29.0)	61.0	48.3

^a values in pareneneses are metric equivalents (kg/mur/). ^b Annealed strips (0.060 in. or 1.52 mm), gas-tungsten-arc welded parallel to the rolling direction. All weld specimens broke in the base metal.

Test	10	0 h	100	00 h
deg F (deg C)	T-216 ^a ksi	T-316 ^b ksi	T-216ª ksi	T-316 ^b ksi
1200(649) 1250(722)	47.0(33.0)	32.0(20.4)	39.0(27.4)	25.0(15.5)
1500(816)	8.5(6.0)	9.0(6.3)	3.5(2.5)	5.5(4.2)

TABLE 7—Stress-rupture strengths of annealed materials.

 a Annealed strip (0.060 in. or 1.52 mm). Values in parentheses are metric equivalents (kg/mm²).

 b ASTM Special Publication No. DS5S2. Values in parentheses are metric equivalents (kg/mm²).

Room Temperature Impact Strengths

If nitrogen has a beneficial effect on improving high temperature strengths of Type 216, could long times at elevated temperatures produce detrimental effects on the notch toughness or intergranular corrosion resistance of Type 216? This could be of particular concern in chemical or petrochemical equipment that is shutdown for repairs.

Oversize Charpy blanks removed from annealed products of Types 216 and 316 were exposed to temperatures between 700 F (371 C) and 1500 F (816 C) for times up to 1000 h. After exposure, longitudinal V-notch Charpies (0.394 in. or 10.0 mm) were machined with the length of the notch going through the thickness of the material. The impact tests were conducted at room temperature, and the average data developed for the heats are shown in Table 9. Note that as the time and temperatures of exposure increase, the impact values of the alloys decrease, both grades being essen-

Test Temperature, deg F (deg C)	Stress, ksi (kg/mm²)	Time, h	Elongation, %
1200(649)	50.0(35.1)	143.5	35.0
1200 (649)	48.0(33.7)	166.3	21.1
1200(649)	46.0(32.3)	132.4	35.9
1200 (649)	40.0(28.1)	624.0	43.0
1200(649)	35.0(24.6)	841.2	11.2
1200(649)	30.0(21.1)	2691.4	33.0
1350(732)	28.0(19.7)	32.0	49.5
1350(732)	20.0(14.1)	128.6	42.0
1350(732)	10.0(7.0)	3385.5	22.3
1350(732)	12.0(8.4)	942.4	52.0
1500 (816)	22.0(15.5)	8.3	54.4
1500 (816)	12.0(8.4)	26.3	56.2
1500(816)	6.0(4.2)	698.9	64.6
1500 (816)	6.0(4.2)	698.9	64.6

TABLE 7A—Stress-rupture properties of annealed Type 216^a.

^a Annealed strip (0.060 in. or 1.52 mm).

Test	0.1% in	1000 h	0.01% i	n 1000 h
deg F (deg C)	Type 216, ^a ksi	Type 316, ^b ksi	Type 216, ^a ksi	Type 316, ^b ksi
$\frac{1100}{1200}(593)$ $1200(649)$	34.0(23.9) 17.5(12.3)	$22.5(15.8) \\ 14.2(10.0)$	$27.0(18.9) \\ 12.9(9.1)$	$12.4(8.7) \\ 7.9(5.6)$

TABLE 8—Creep strength of annealed materials.

 a Annealed plate (0.500 in. thick or 12.7 mm). Values in parentheses are metric equivalents (kg/mm²).

^b ASTM DS5S2. Values in parentheses are metric equivalents (kg/mm²).

tially unaffected at temperatures up to 1000 F (538 C) for 1000 h. However, at 1200 F (649 C) the alloys show significant decrease in impact strength. At 1350 F (732 C) and 1500 F (816 C), Type 216 displayed very low values (3 ft·lb or 4.1 joules). The embrittlement of Type 216 appears to be due to chi-phase, whereas sigma caused embrittlement in Type 316.

Effect of Elevated Temperatures on Intergranular Corrosion

Intergranular corrosion tests are in progress to determine the comparative resistances of Types 216 and 316 to stress cracking after exposure to acidified copper sulfate solutions. This work follows the test procedures employed by Dr. Carl Samans [1].² According to Dr. Samans, the copper sulfate test can be used as an acceptance test for determining a material's resistance to stress cracking in polythionic acid, because the copper sulfate is a more severe test. Here, annealed specimens (0.060 by $\frac{1}{4}$ by 3 in. long, or 15.2 by 6.35 by 76.2 mm) were heat treated at temperatures between 700 F (371 C) and 1500 F (816 C) for times up to 1000 h. The specimens were then exposed to boiling copper sulfate (ASTM, A-393) and bent 180 deg around a diameter of $\frac{1}{4}$ in. (6.35 mm). The specimens were examined at $\times 20$ for evidence of cracking. Initial data for 1000 h tests showed that the nitrogenbearing grade can be crack-free at 1000 F (538 C) or lower. As the temperature increased, failure occured. The chromium-nickel types revealed cracking at 1200 F (648 C) and above. Data developed by Dr. Samans show that Types 316 and 316L are sensitized to copper sulfate test. This agrees with our initial findings.

Polythionic acid can form in sulfur-bearing refining streams and, thus, cause stress cracking of material during shutdown of equipment which had been exposed to service temperatures between 900 F (482 C) and 1500 F (816 C). The stress cracking data developed to date are minimal, therefore, it would be premature to form any conclusions.

² The italic numbers in brackets refer to the list of references appended to this paper.

Impact Strength. ^a
Temperature
Room
ю
Temperatures
f Elevated
0
9-Effect
TABLE

	1	Ч	100	Ч (100	ч о
Exposure Temperature, deg F (deg C)	Type $216/216L$, ft·lb ^b	Type 316/316L, ft·lb ^b	Type $216/216L$, ft·lb ^b	Type $316/316L$, ft.lb ^b	$Type 216/216L, ft \cdot lb^{\delta}$	Type 316/316L, ft.lb ⁶
700(371)	216(293)	190(258)	217(294)	204(277)	217(294)	166(225)
850(454)	213(289)	196(266)	216(293)	209(284)	225(305)	176(239)
1000(538)	212(288)	202(274)	209(284)	202(274)	206(280)	188(255)
1200(649)	216(293)	190(258)	158(214)	160(217)	60(83)	98(133)
1350(732)	193(262)	202(274)	46(62)	97(132)	4(5.4)	58(79)
1500(816)	154(209)	161(219)	30(41)	91(124)	4(4.1)	37(50)
a Longitudinal CVN from	a annealed nlate an	d bar stock (0 500 i	n or 12.7 mm thick	and 0.625 in. or 15	9 mm dia notched	through thickness,

þ in annealed plate and bar stock (0.200 m. JOURGUALITIAL CV 11 HUMI ALIFICATED FLAUE AND UNA SUCCE tested in triplicate.
Values in parentheses are metric equivalents (joules).

Summary

It is evident from the above data that Type 216 offers improved strengths over Type 316 to temperatures approaching 1350 F (732 C). Tension testing shows that this nitrogen-bearing grade has greater strength than Type 316 for temperatures of 1400 F (760 C) and higher. When the time at temperature is extended, as in the case of stress rupture testing, the strengths of both grades are comparable at 1350 F (732 C), while Type 216 maintains superior strength at 1200 F (649 C).

However, impact testing of materials exposed to elevated temperatures suggest that Type 216 may be restricted to service temperatures near 1200 F (649 C). For instance, Type 216 like Type 316 showed notable drops in impact strength (room temperature) after exposure to 1200 F (649 C) for 1000 h. Very low impact values of 4 ft·lb (5.4 joules) were evident for Type 216 after exposure to 1350 F (732 C) for 1000 h compared to 58 ft·lb (79 joules) for Type 316. This embrittling behavior is primarily attributed to chi formation and sigma formation in Types 216 and 316, respectively. Limited intergranular corrosion testing in acidified copper sulfate solution suggests that Type 216 can resist stressing (bending radius = twice thickness) so as to be crack-free after exposure to temperatures of 1000 F (538 C) or lower. The chromium-nickel types revealed cracking at 1200 F (649 C) and above.

The superior strength of Type 216 coupled with its excellent corrosion properties make Type 216 attractive for applications in the process industries (namely, chemical petrochemical, pulp, and paper) as well as marine and space markets.

Reference

[1] Samans, C. H., Corrosion, Vol. 20, 1960, pp. 256t-262t.